Size-resolved Composition and Morphology of Particulate 1 Matter During the Southwest Monsoon in Metro Manila,

2

Philippines 3

- 4
- Melliza Templonuevo Cruz^{1,2}, Paola Angela Bañaga^{1,3}, Grace Betito³, Rachel A. Braun⁴, Connor 5
- Stahl⁴, Mojtaba Azadi Aghdam⁴, Maria Obiminda Cambaliza^{1,3}, Hossein Dadashazar⁴, Miguel 6
- Ricardo Hilario³, Genevieve Rose Lorenzo¹, Lin Ma⁴, Alexander B. MacDonald⁴, Preciosa 7
- Corazon Pabroa⁵, John Robin Yee⁵, James Bernard Simpas^{1,3}, Armin Sorooshian^{4,6} 8
- 9

- 11 ²Institute of Environmental Science and Meteorology, University of the Philippines, Diliman, Quezon City 1101,
- 12 Philippines
- 13 ³Department of Physics, School of Science and Engineering, Ateneo de Manila University, Quezon City 11011108, 14 Philippines
- 15 ⁴Department of Chemical and Environmental Engineering, University of Arizona, Tucson, AZ, USA
- 16 ⁵Philippine Nuclear Research Institute, Commonwealth Avenue, Diliman, Quezon City 1101, Philippines
- 17 ⁶Department of Hydrology and Atmospheric Sciences, University of Arizona, Tucson, AZ, USA
- 18
 - 19 Correspondence to: Melliza Templonuevo Cruz (liz@observatory.ph)
 - 20

¹⁰ ¹Manila Observatory, Quezon City 11011108, Philippines

21 Abstract

22 This paper presents novel results from size-resolved particulate matter (PM) mass, composition, 23 and morphology measurements conducted during the 2018 Southwest Monsoon (SWM) season in 24 Metro Manila, Philippines. Micro-Orifice Uniform Deposit Impactors (MOUDIs) were used to 25 collect PM sample sets composed of size-resolved measurements at the following aerodynamic 26 cutpoint diameters (D_p): 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 µm. that wereEach sample set was analyzed for mass, morphology, black carbon (BC), and composition of the water-27 28 soluble fraction. Analysis for mass were done on two sample sets while black carbon (BC) and 29 morphology analysis were done on a single sample set. The bulk of the PM mass was between 30 0.18–1.0 µm with a dominant mode between 0.32–0.56 µm. Similarly, most of the black carbon (BC) mass was found between 0.10–1.0 µm, peaking between 0.18–0.32 µm. These peaks are 31 located in the Greenfield Gap or the size range between 0.10–1.0 µm, where wet scavenging by 32 33 rain is relatively inefficient. Similarly, most of the black carbon (BC) mass was found between 0.10 1.0 µm (the so-called Greenfield gap), peaking between 0.18 0.32 µm, where wet 34 35 scavenging by rain is inefficient. In the range of $0.10 - 0.18 \mu m$, BC constituted 78.1% of the 36 measured mass. Comparable contributions of BC (26.9%) and the water-soluble fraction (31.3%) 37 to total PM were observed and most of the unresolved mass, which in total amounted to 41.8%, 38 was for diameters exceeding 0.32 µm. The water-soluble ions and elements exhibited an average combined concentration of 8.53 µg m⁻³, with SO₄²⁻, NH₄⁺, NO₃⁻, Na⁺, and Cl⁻ as the major 39 40 contributors. Positive Matrix Factorization (PMF) was applied to identify the possible aerosol 41 sources and estimate their contribution to the water-soluble fraction of collected PM. The factor 42 with the highest contribution was attributed to "Aged/Transported" aerosol (48.0%) while "Sea 43 Salt" (22.5%) and "Combustion" emissions (18.7%) had comparable contributions. 44 "Vehicular/Resuspended Dust" (5.6%) as well as "Waste Processing" emissions (5.1%) were also 45 identified. Microscopy analysis highlighted the ubiquity of non-spherical particles regardless of 46 size, which is significant when considering calculations of parameters such as single scattering 47 albedo, asymmetry parameter, and extinction efficiency.

48 Results of this work have implications for aerosol impacts on public health, visibility, and regional

49 climate as each of these depend on physicochemical properties of particles as a function of size.

- 50 The significant influence from Aged/Transported aerosol to Metro Manila during the SWM season
- 51 indicates that local sources in this megacity do not fully govern this coastal area's aerosol

52 properties and that PM in Southeast Asia can travel long distances regardless of the significant 53 precipitation and potential wet scavenging that could occur. That the majority of the regional 54 aerosol mass burden is accounted for by BC and other insoluble components has important downstream effects on the aerosol hygroscopic properties, which depend on composition. The 55 56 results are relevant for understanding the impacts of monsoonal features on size-resolved aerosol 57 properties, notably aqueous processing and wet scavenging. Finally, the results of this work 58 provide contextual data for future sampling campaigns in Southeast Asia such as the airborne component of the Cloud, Aerosol, and Monsoon Processes Philippines Experiment (CAMP²Ex) 59 60 planned for the SWM season in 2019. Aerosol characterization via remote sensing is notoriously difficult in Southeast Asia, which elevates the importance of datasets such as the one presented 61 62 here.

63

64 **1. Introduction**

65

Ambient atmospheric aerosol particles impact human health, visibility, climate, and the 66 67 hydrological cycle. Major factors governing these behaviors, such as deposition fraction in the 68 respiratory system and activation into cloud condensation nuclei (CCN), include size and chemical 69 composition. Therefore, size-resolved measurements of ambient aerosol particles can lend 70 additional insights to the behaviors and implications of particulate matter (PM) in the atmosphere. 71 One region of interest for characterization of aerosols is Southeast Asia due to increasing 72 urbanization and the exposure of the population to a variety of aerosol sources, both natural and 73 anthropogenic (Hopke et al., 2008). However, use of space-borne remote-sensing instrumentation 74 presents a challenge for characterization of aerosol in this region, due to issues such as varying 75 terrain and cloud cover (Reid et al., 2013).

76 The Philippines represents a country in Southeast Asia with a developing economy, rapid 77 urbanization, old vehicular technology, and less stringent air quality regulations (e.g., Alas et al., 78 2017). It is also highly sensitive to the effects of climate change including prolonged dry periods 79 and reductions in southwest monsoon (SWM) rainfall in recent decades (e.g., Cruz et al., 2013). 80 Metro Manila is the country's capital and center of political and economic activities. Also referred 81 to as the National Capital Region, Metro Manila is composed of 16 cities and a municipality that collectively occupy a land area of ~619 km². As of 2015, Metro Manila had a population of 82 83 approximately 12.88 million (Philippine Statistics Authority, 2015). Of the cities comprising the 84 Metro Manila area, the one that is the focus of this study, Ouezon City, is the most populated (2.94) million people) with a population density of ~17,000 km⁻² as of 2015 (Philippine Statistics 85 86 Authority, 2015).

87 The rainfall pattern in Southeast Asia is governed by topographic effects and the prevailing 88 surface winds brought by the monsoons. Mountain ranges in the Philippines are generally oriented 89 north to south in the eastern and western coasts. As such, northeasterly winds during the East Asian 90 winter monsoon that starts in November brings wetness (dryness) on the eastern (western) coasts 91 of the country. In contrast, the rainy season starts in May when the Western North Pacific 92 subtropical high moves northeast and the Asian summer monsoon enables the propagation of 93 southwesterly wind through the Philippines (Villafuerte et al., 2014). Metro Manila, located on 94 the western side of the Philippines, therefore experiences wet (May-October) and dry (November-95 April) seasons. The large seasonal shift in prevailing wind directions can cause changes in the

96 source locations of aerosol transported to the Philippines and the subsequent direction in which 97 emissions from the Philippines are transported, such as to the northwest (e.g., Chuang et al., 2013) 98 or southwest (e.g., Farren et al., 2019). However, one interesting feature of Metro Manila is the consistency of $PM_{2.5}/PM_{10}$ mass concentrations during both the dry (44/54 µg m⁻³) and wet seasons 99 (43/55 µg m⁻³) (Kim Oanh et al., 2006), which stands in contrast to typical assumptions that 100 101 increased wet scavenging during rainy seasons would lead to decreases in measured PM (e.g., Liao 102 et al., 2006). While similar results are observed in Chennai, India, this behavior is different than 103 other cities in Asia, including Bandung City (Indonesia), Bangkok (Thailand), Beijing (China), 104 and Hanoi City (Vietnam), that-which exhibit reduced $PM_{2.5}$ levels during the wet season as 105 compared to the dry season (Kim Oanh et al., 2006). While the total PM levels may stay constant 106 across the wet and dry seasons, seasonally-resolved analyses will provide additional insights into 107 how the composition, morphology, and sources (transported vs. local emissions) change on a 108 seasonal basis.

109 Metro Manila has been drawing growing interest for PM research owing to the significant 110 levels of black carbon (BC). A large fraction of PM in Metro Manila can be attributed to BC (e.g., 111 ~50% of PM_{2.5}; Kim Oanh et al., 2006), with previously measured average values of BC at the Manila Observatory (MO) reaching ~10 μ g m⁻³ for PM_{2.5} (Simpas et al., 2014). The impacts of the 112 113 high levels of BC present on human health have also received attention (Kecorius et al., 2019). 114 Identified major sources of BC include vehicular, industrial, and cooking emissions (Bautista et 115 al., 2014; Kecorius et al., 2017). Vehicular emissions, especially along roadways where personal 116 cars and motorcycles, commercial trucks, and motorized public transportation, including powered 117 tricycles and *jeepneys*, are plentiful. For instance, measurements of PM_{2.5} at the National Printing 118 Office (NPO) located alongside the major thoroughfare Epifanio de los Santos Avenue (EDSA) were on average 72 μ g m⁻³; this value is twice the average concentration at the Manila Observatory 119 120 (MO), an urban mixed site located approximately 5 km from NPO (Simpas et al., 2014). In 121 addition to local emissions, long-range transport of pollution, such as biomass burning, can also 122 impact the study region (e.g., Xian et al., 2013; Reid et al., 2016a/b). However, most past work 123 referenced above has focused on either total PM_{2.5} or PM₁₀ composition, and therefore, detailed 124 size-resolved composition information has been lacking in this region. Like other monsoonal 125 regions (Crosbie et al., 2015; Qu et al., 2015), it is of interest for instance to know if products of 126 aqueous processing (e.g., sulfate, organic acids) during the monsoonal period, promoted by the

high humidity, become more prominent in certain size ranges to ultimately enhancehygroscopicity, which is otherwise suppressed with higher BC influence.

129 A year-long sampling campaign (Cloud, Aerosol, and Monsoon Processes Philippines 130 Experiment (CAMP²Ex) weatHEr and CompoSition Monitoring (CHECSM) study) was 131 established in July 2018 to collect size-resolved aerosol measurements in Metro Manila. The aim 132 of this study is to report size-resolved PM measurements taken over the course of the SWM (July-133 October) of 2018 in Quezon City, Metro Manila, Philippines as part of CHECSM. The results of 134 this study are important for the following reasons: (i) they provide size-resolved analysis of BC in 135 an area previously characterized as having one of the highest BC mass percentages in the whole 136 world; (ii) they provide a basis for better understanding the unusual phenomenon of having similar 137 PM levels during a wet and dry season; (iii) they provide contextual data for contrasting with both 138 other coastal megacities and also other monsoonal regions; and (iv) they can lend insights into the 139 characteristics of aerosol transported both into and out of Metro Manila and how important local 140 sources are in Metro Manila relative to transported pollution.

141 Outcomes of this study include (i) the first size-resolved characterization of both aerosol 142 composition and morphology in Metro Manila for the SWM, with implications in terms of PM 143 effects on climate, visibility, the hydrological cycle, and public health owing to the dependence of 144 these impacts on particle size; (ii) archival data that contributes to the timeline of aerosol research 145 in Metro Manila, and more broadly Southeast Asia, where there is considerable concern over air 146 pollution; and (iii) baseline data for aerosol composition to be used to inform and assist research 147 to be conducted during future field campaigns in Southeast Asia including the same seasonal period (i.e., SWM) in 2019 as part of CAMP²Ex, which will involve both surface and airborne 148 149 measurements.

150 2. Experimental Methods

151 **2.1 Sample Site**

Sampling was performed at MO in Quezon City, Philippines (14.64° N, 121.08° E). The sampling instrumentation was located Two MOUDIs were placed -inside an unoccupied room on the 3rd floor of the MO office administration building (~85–87 m above sea level). The inlet, located just outside the window, consists of a 2 m long stainless steel tube and a reducer and a reducer and a reducer that is connected directly to the MOUDI inlet. Figure 1 visually shows the sampling

157 location and potential surrounding aerosol sources. Past work focused on PM_{2.5} suggested that the 158 study location is impacted locally mostly by traffic, various forms of industrial activity, meat 159 cooking from local eateries, and, based on the season, biomass burning (Cohen et al., 2009). This is consistent with another source apportionment study which reported that potential sources in six 160 161 sites across Metro Manila include traffic, secondary particles, and biomass burning (Kim Oanh et al., 2013). Fourteen sample sets were collected during the SWM season (July October 2018), with 162 163 details about the operational and meteorological conditions during each sample set shown in Table 164 1.

Meteorological data were collected using a Davis Vantage Pro 2 Plus weather station colocated with the aerosol measurements at located on the roof (~90 m above sea level, ~15 m above ground level) above where the MOUDIs were located-MO. Except for precipitation, which is reported here as accumulated rainfall, reported values for each meteorological parameter represent averages for the sampling duration of each aerosol measurement.

170 -The mean temperature during the periods of MOUDI sample collection ranged from 24.9 to 28.1° C, with accumulated rainfall ranging widely from no rain to up to 78.4 mm. To 171 172 identify sources impacting PM via long-range transport to the Metro Manila region, Figure 173 1a summarizes the five-day back-trajectories for air masses arriving at MO on the days when 174 samples were being collected, calculated using the NOAA Hybrid Single-Particle Lagrangian 175 Integrated Trajectory (HYSPLIT) model (Stein et al., 2015; Rolph, 2016). Trajectory calculations 176 were started at 00, 06, 12, and 18 hours in MO at the height of the MOUDI inlet (~ 12 m above 177 ground level) using meteorological files from the NCEP/NCAR Reanalysis dataset. Trajectory 178 cluster analysis was conducted using TrajStat (Wang et al., 2009). The back-trajectories in Figure 179 1a show that indeed 66% of the wind came from the southwest during the sampling periods.

180 2.2 MOUDI Sample Sets

Particulate matter<u>PM</u> was collected on Teflon substrates (PTFE membrane, 2 μm pore,
 46.2 mm, Whatman) in Micro-Orifice Uniform Deposit Impactors (MOUDI, MSP Corporation,
 Marple et al., 2014). Size-resolved measurements were taken at the following aerodynamic
 cutpoint diameters (D_p): 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 μm. Fourteen
 sample sets were collected during the SWM season (July-October 2018), with details about the
 operational and meteorological conditions during each sample set shown in Table 1. To determine

187 the optimum sampling time that will collect enough sample for subsequent analyses, collection 188 time for the first four samples ranged from 24 to 119 hours. Subsequent sampling were then fixed 189 to 48 hours with one sample set collected every week. The sampling collection was designed to 190 include samples from each day of the week so the collection cycled between Monday -191 Wednesday, Tuesday – Thursday, Wednesday – Friday, and Saturday – Monday, starting at 1400 192 (local time) for the weekday samples and 0500 for the weekend samples. The Teflon substrates 193 were pretreated by washing with deionized water and air drying in a covered box. Substrates were 194 placed and retrieved from the cascade impactor inside the laboratory in an adjacent building and 195 transported to and from the sampling site using an impactor holder (Csavina et al., 2011). Samples 196 are immediately placed in the freezer upon retrieval. 197 For a subset of the sampling periods On two occasions, two pairs of MOUDI sets (Sets

198 MO3/MO4 and MO13/MO14) were collected simultaneously such that both sets one set in each 199 pair could undergo different types of analyses. Sets 3 and 13 One set in each pair underwent 200 gravimetric analysis using a Sartorius ME5-F microbalance. Substrates were conditioned for at 201 least 24 h at a mean temperature of 20-23 °C and a mean relative humidity of 30-40% before preand post-weighing (U.S. Environmental Protection Agency, 2016). MOUDI set 13 was 202 203 additionally examined with a Multi-wavelength Absorption Black Carbon Instrument (MABI; 204 Australian Nuclear Science and Technology Organisation). This optically-based instrument 205 quantifies absorption and mass concentrations at seven wavelengths between 405 and 1050 nm; 206 however, results are reported only for 870 nm to be consistent with other studies, as BC is the 207 predominant absorber at that wavelength (e.g., Ramachandran and Rajesh, 2007; Ran et al., 2016). 208 One additional sample set for microscopy analysis was collected for one hour on August 1 using 209 aluminum substrates.

210 **2.3 Chemical Composition Analysis**

211 <u>Twelve sample sets, composed of 11 samples each, were analyzed for water-soluble ions</u> 212 <u>and elements (Table 2).</u> In order to preserve samples for additional analysis, each Teflon substrate 213 was cut in half. A half of each substrate was extracted in 8 mL of Milli-Q water (18.2 M Ω -cm) 214 through sonication for 30 min in a sealed polypropylene vial. A blank substrate was processed in 215 the same method to serve as a background control sample. Subsequent chemical analysis of the 216 water-soluble components in the aqueous extracts were performed using ion chromatography (IC; 217 Thermo Scientific Dionex ICS - 2100 system) for the following species: cations = Na^+ , NH_4^+ , Mg^{2+} , Ca^{2+} , dimethylamine (DMA), trimethylamine (TMA), diethylamine (DEA); anions =, 218 methanesulfonate (MSA), pyruvate, adipate, succinate, maleate, oxalate, phthalate, Cl⁻, NO₃⁻, 219 220 SO₄²⁻. Owing to co-elution of TMA and DEA in the IC system, a cumulative sum of the two is 221 reported here, which represents an underestimate of their total mass concentration owing to overlap 222 in parts of their peaks. Limits of detection (LOD) were calculated for each species based on their 223 respective calibration curve (Table S1), with LOD being three times the standard deviation of the 224 residuals (predicted signal minus measured signal) divided by the slope of the calibration curve 225 (Miller and Miller, 2018).

The aqueous extracts were simultaneously characterized for elemental composition using triple quadrupole inductively coupled plasma mass spectrometry (ICP-QQQ; Agilent 8800 Series) for the following species: K, Al, Fe, Mn, Ti, Ba, Zn, Cu, V, Ni, P, Cr, Co, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Ag, Cd, Sn, Cs, Hf, Tl, Pb. Limits of detection of the examined elements were calculated automatically by the ICP-QQQ instrument and were in the ppt range (Table S1). The sample concentrations represent an average of three separate measurements with a standard deviation of 3% or less.

Note that some species were detected by both IC and ICP-QQQ (i.e., Na⁺, K⁺, Mg²⁺, Ca²⁺), and that the IC concentrations are used here for all repeated species with the exception of K⁺ owing to better data quality from ICP-QQQ. All IC and ICP-QQQ species concentrations for samples have been corrected by subtracting concentrations from background control samples. For more examples of the application of these methods used for substrate collection and IC/ICP analysis, the reader is referred to other recent work (Braun et al., 2017; Ma et al., 2019; Schlosser et al., 2017).

240 2.4 Microscopy Analysis

As already noted, one MOUDI set on August 1 was devoted to microscopy analysis. Morphology and additional elemental composition analysis was carried out on this set of aluminum substrates using scanning electron microscopy equipped with energy dispersive X-ray spectroscopy (SEM-EDX) in the Kuiper Imaging cores at the University of Arizona. Secondary electron (SE) imaging and EDX elemental analysis were performed using a Hitachi S-4800 high resolution SEM coupled to a Noran system Six X-ray Microanalysis System by Thermo Fisher

9

Scientific. EDX analysis on individual particles was performed with 30 kV accelerating voltage to obtain weight percentages of individual elements. SEM-EDX results showed that the background control aluminum substrate was dominated by Al (88.27%), with minor contributions from Ag (5.34%), C (4.87%), O (0.79%), Fe (0.67%), and Co (0.05%). Such contributions were manually subtracted from spectra of individual particles on sample substrates, with the remaining elements scaled up to hundred percent. Image processing was conducted with Image J software to measure particle dimensions and adjust the contrast and brightness of images to provide better visualization.

254 2.5 Computational Analysis

 $\sigma_{ij} = 0.05 \cdot X_{ij} + LOD_{ij},$

255 This study reports basic descriptive statistics for chemical concentrations and correlations 256 between different variables. Statistical significance hereafter corresponds to 95% significance 257 based on a two-tailed Student's t-test. To complement correlative analysis for identifying sources 258 of species, positive matrix factorization (PMF) modeling was carried out using the United States 259 Environmental Protection Agency's (US EPA) PMF version 5. A total of 132 samples from the 12 260 sets analyzed for water-soluble ions and elements were used in the PMF analysis. Species 261 concentrations were examined before being using as inputted to PMF. -Species considered as 262 "strong" based on high signal-to-noise ratios (S/N > 1) and those with at least 50% of the 263 concentrations above the detection limitLOD were used in the PMF modeling (Norris et al., 2014). 264 This resulted in a 132 (samples) ****** 30 (species) data matrix that was used as-inputted to PMF. 265 Data points with concentrations exceeding the LOD had uncertainty quantified as-follows: 266

(Equation 1)

where σ_{ij} , X_{ij} , and LOD_{ij} are the uncertainty, concentration, and LOD, respectively, of the j^{th} 269 species in the i^{th} sample (Reff et al., 2007). When concentration data were not available for a 270 271 particular stage of a MOUDI set for a species, the geometric mean of the concentrations for that 272 MOUDI stage and species was applied with uncertainty counted as four times the geometric mean 273 value (Polissar et al., 1998; Huang et al., 1999). A 25% extra modeling uncertainty was applied 274 to account for other sources of errors, such as changes in the source profiles and chemical 275 transformations (Dumanoglu et al., 2014; Norris et al., 2014). The model was run 20 times with a 276 randomly chosen starting point for each run.

277 **3. Results**

278 **3.1 Total Mass Concentrations and Charge Balance**

279 The average total mass concentration (\pm standard deviation) of water-soluble species across all MOUDI stages (Table 1) during the study period was $8.53 \pm 4.48 \ \mu g \ m^{-3}$ (range = 2.7–16.6 μg 280 m⁻³). The species contributing the most to the total water-soluble mass concentration during the 281 282 SWM included SO₄²⁻ (44% ± 6%), NH₄⁺ (18% ± 5%), NO₃⁻ (10 ± 3%), Na⁺ (8 ± 3%), and Cl⁻ (6% 283 \pm 3%). The meteorological parameters from Table 1 best correlated to total water-soluble mass 284 concentrations were temperature (r = 0.64) and rainfall (r = -0.49). The highest total mass 285 concentration (set MO13/14 = 16.6 μ g m⁻³) occurred during the period with one of the highest 286 average temperatures (27.8 °C) and second least total rainfall (0.8 mm). Other sampling periods 287 with high mass concentrations (sets MO7, MO8, and MO12) coincided with the highest 288 temperature and lowest rainfall observations. High temperatures, and thus more incident solar 289 radiation, presumably enhanced production of secondary aerosol species via photochemical 290 reactions as has also been observed in other regions for their respective monsoon season (Youn et 291 al., 2013).

292 Low rainfall is thought to have been coincident with reduced wet scavenging of aerosol at 293 the study site as has been demonstrated for other regions such as North America (Tai et al., 2010) 294 and megacities such as Tehran (Crosbie et al., 2014). However, set MO11 exhibited a very low 295 concentration even with high temperature and lack of rainfall, which may be due to changes in the 296 source and transport of aerosol since this sample set coincided with a significant change in average 297 wind direction (290.2° for MO11 vs. $90.1^{\circ} - 127.5^{\circ}$ for all other MOUDI sets). While the reported 298 rainfall measurements were taken at MO, inhomogeneous rainfall patterns in the regions 299 surrounding the Philippines could also contribute to the wet scavenging of PM, thereby lowering 300 the quantity of transported particles reaching the sample site. Future work will address the 301 influence of spatiotemporal patterns of precipitation on PM loadings in the Philippines as a point 302 measurement at an aerosol observing site may be misleading.

303 On two occasions, two simultaneous MOUDI sets (Sets MO3/MO4 and MO13/MO14) 304 were collected for the potential to compare different properties that require separate substrates. 305 The total mass concentrations based on gravimetric analysis of sets MO3 and MO13 were 18.6 μ g 306 m⁻³ and 53.0 μ g m⁻³, respectively (Figure 2). Both sets exhibited a dominant concentration mode 307 between 0.32–0.56 μ m and the MO3 set was different in that it exhibited bimodal behavior with a 308 second peak between 1.8–3.2 μ m. The sum of speciated water-soluble species accounted for only 309 27.8% and 31.3% of the total gravimetric mass of sets MO3 and MO13, respectively, indicative 310 of significant amounts of water-insoluble species undetected by IC and ICP-QQQ. When adding 311 the total mass of BC (14.3 μ g m⁻³) to the other resolved species from set MO13 (the one time BC 312 was measured), there was still 22.1 μ g m⁻³ of unresolved mass (41.8% of total PM). Most of the 313 unaccounted mass was for D_p > 0.32 μ m.

The observation of BC accounting for 26.9% of total PM (14.3 μ g m⁻³) is consistent with 314 315 past work highlighting the significant fraction of BC in the ambient aerosol of Manila (Kim Oanh 316 et al., 2006; Bautista et al., 2014; Simpas et al., 2014; Kecorius et al., 2017). However, this fraction 317 of BC is very high compared to measurements during the monsoon season in other parts of the 318 world. The mass fraction of BC in total suspended PM (TSPM) was 1.6%/2.2% for the monsoon 319 season in 2013/2014 in Kadapa in southern India, even though the TSPM measured was comparable to that in Manila (64.9 and 49.9 µg m⁻³, for 2013 and 2014 in Kadapa, respectively) 320 321 (Begam et al., 2017). Multiple studies during the monsoon season in a coastal region in southwest 322 India showed BC mass contributions of 1.9 - 5% (Aswini et al., 2019 and references therein). 323 Airborne measurements around North America and in Asian outflow revealed that BC accounted 324 for only ~1-2% of PM_{1.0} (Shingler et al., 2016) and ~5-15% of accumulation mode aerosol mass 325 (Clarke et al., 2004), respectively.

326 To investigate further about the missing species, a charge balance was carried out for all 327 MOUDI sets (Table 2) to compare the sum of charges for cations versus anions based on IC 328 analysis including K from ICP-QQQ analysis (species listed in Section 2.3). The slope of the 329 charge balances (cations on y-axis) for the cumulative dataset was 1.33 and ranged from 0.89 to 330 1.41 for the 12 individual MOUDI sets that had IC and ICP-QQQ analysis conducted on them. 331 Eleven of the 12 sets exhibited slopes above unity indicating that there was a deficit in the amount 332 of anions detected, which presumably included species such as carbonate and various organics. To 333 further determine if there were especially large anion or cation deficits in specific size ranges, 334 slopes are also reported for $0.056-1 \mu m$ and $> 1 \mu m$. There were no obvious differences other than 335 two MOUDI sets exhibited slopes below 1.0 for the smaller diameter range (0.056–1 μ m) while 336 all slopes exceeded unity for $> 1 \mu m$.

337 **3.2 Mass Size Distributions and Morphology**

338 3.2.1 Black Carbon

339 The size-resolved nature of BC has not been characterized in Manila and MOUDI set 340 MO13 offered a view into its mass size distribution (Figure 3a). There was a pronounced peak between 0.18–0.32 μ m (5.0 μ g m⁻³), which is evident visually in the substrate's color when 341 342 compared to all other stages of that MOUDI set (Figure 3b). This observed peak in the mass size 343 distribution of BC is similar to previous studies of the outflow of East Asian countries (Shiraiwa 344 et al., 2008), biomass burning and urban emissions in Texas (Schwarz et al., 2008), measurements 345 in the Finnish Arctic (Raatikainen et al., 2015), and airborne measurements over Europe 346 (Reddington et al., 2013). In contrast, measurements in Uji, Japan showed a bimodal size 347 distribution for the mass concentration of BC in the submicrometer range (Hitzenberger and 348 Tohno, 2001).

349 In the present study, there were significant amounts of BC extending to as low as the 0.056-0.1 µm MOUDI stage (0.28 µg m⁻³) and extending up in the supermicrometer range, with up to 350 351 0.25 µg m⁻³ measured between 1.8–3.2 µm. Remarkably, BC accounted for approximately 78.1% 352 (51.8%) by mass of the total PM in the range of $0.10 - 0.18 \mu m (0.18 - 0.32 \mu m)$. For comparison, 353 the mass percent contribution of BC measured in the megacity of Nanjing, China was 3.3% (1.6%) 354 at 0.12 (0.08) µm (Ma et al., 2017). Based on visual inspection of color on all MOUDI sets, MO13 355 appears to be representative of the other sets based on the relative intensity of the color black on 356 substrates with different cutpoint diameters (Figure 3b); the $0.18-0.32 \mu m$ substrate always was 357 the most black, with varying degrees of blackness extending consistently into the supermicrometer 358 stages.

359 Microscopy analysis revealed evidence of non-spherical particles in each MOUDI stage 360 below 1 µm (Figure 4), which is significant as the common assumption theoretically is that 361 submicrometer particles are typically spherical (e.g., Mielonen et al., 2011). Errors in this 362 assumption impact numerical modeling results and interpretation of remote sensing data for 363 aerosols (e.g., Kahnert et al., 2005), owing to incorrect calculations of parameters such as single 364 scattering albedo, asymmetry parameter, and extinction efficiency (e.g., Mishra et al., 2015). Some 365 studies have noted that submicrometer particles could be composed of an agglomeration of small 366 spherical particles originally formed through gas-to-particle conversion processes (Almeida et al., 367 2019), which could potentially explain the appearance for some of the observed particles in Figure 368 4. Since only single particles were examined that may not be fully representative of all particles

369 on a particular MOUDI substrate, it is noteworthy that all five particles shown between 0.056 - 1370 µm were irregularly shaped with signs of both multi-layering and constituents adhered to one 371 another.

372 The images show that a potentially important source of BC in the area could be soot 373 aggregates, which are formed by a vaporization-condensation process during combustion often 374 associated with vehicular exhaust (e.g., Chen et al., 2006; Chithra and Nagendra, 2013; Wu et al., 375 2017). Kecorius et al. (2017) projected that 94% of total roadside refractory PM with number 376 concentration modes at 20 and 80 nm in the same study region was linked to *jeepneys*, the most 377 popular and inexpensive mode of public transport in Metro Manila. with number concentration 378 modes at 20 and 80 nm. They associated the larger mode with soot agglomerates, which is 379 consistent with the smallest MOUDI size range examined here (0.056-0.1 µm; Figure 4b) 380 exhibiting signs of agglomeration.

381 The total BC mass concentration integrated across all stages of MOUDI set MO13 (14.3 µg m⁻³) was remarkably high in contrast to BC levels measured via either filters, aethalometers, or 382 383 single particle soot photometers in most other urban regions of the world (Metcalf et al., 2012 and references therein): Los Angeles Basin (airborne: 0.002–0.53 µg m⁻³), Atlanta, Georgia (ground: 384 0.5–3.0 µg m⁻³), Mexico City (airborne: 0.276–1.1 µg m⁻³), Sapporo, Japan (ground: 2.3–8.0 µg 385 m⁻³), Beijing, China (ground: 6.3–11.1 µg m⁻³), Bangalor, India (ground: 0.4–10.2 µg m⁻³), Paris, 386 387 France (ground: 7.9 µg m⁻³), Dushanbe, Russia (ground: 4–20 µg m⁻³), Po Valley, Italy (ground: $0.5-1.5 \ \mu g \ m^{-3}$), Thessaloniki, Greece (ground: $3.3-8.9 \ \mu g \ m^{-3}$). This is intriguing in light of 388 389 extensive precipitation, and thus wet scavenging of PM, during the study period, which is offset 390 by enormous anthropogenic emissions in the region, such as by powered vehicles like the *jeepneys* 391 that are notorious for BC exhaust (Kecorius et al., 2017).

392 A possible explanation for the large contribution of BC to PM, and the persistence of PM 393 after rain events (Kim Oanh et al., 2006), is that the BC is not efficiently scavenged by precipitating 394 rain drops. Small particles enter rain drops via diffusion whereas large particles enter via 395 impaction. However, particles with a diameter in the range of $0.1-1 \mu m$ (known as the Greenfield 396 gap) are too large to diffuse efficiently and too small to impact, and are therefore not efficiently 397 scavenged (Seinfeld and Pandis, 2016). Absorption spectroscopy of set MO13 (Figure 2b) reveals 398 that 95% of the BC mass is concentrated in the Greenfield gap, and thus the removal of BC due to 399 precipitation is inefficient. The Greenfield gap contains $62 \pm 11\%$ of the total mass (calculated for

MO3/MO13) and $65 \pm 10\%$ of the water-soluble mass (calculated for the other 12 MO sets). As

401 noted earlier, BC observations discussed in this paper were based only on a single MOUDI set and

402 the effect of inefficient scavenging in the Greenfield Gap could just be one of the many potential

403 processes affecting the BC mass size distribution. Subsequent work that will include BC

404 <u>measurements in the dry season will further investigate this hypothesis.</u>

405 3.2.2 Water-Soluble Ions

406 There were two characteristic mass size distribution profiles for the water-soluble ions 407 speciated by IC₂ depending on whether the species were secondarily produced via gas-to-particle 408 conversion or associated with primarily emitted supermicrometer particles. The average IC species 409 mass concentration profile across all MOUDI sets is shown in Figure 5. Secondarily-Secondarily-410 produced species exhibited a mass concentration mode between 0.32–0.56 µm, including common inorganic species (SO₄²⁻, NH₄⁺), MSA, amines (DMA, TMA+DEA), and a suite of organic acids, 411 412 such as oxalate, phthalate, succinate, and adipate, produced via precursor volatile organic 413 compounds (VOCs). Two organic acids with peaks in other size ranges included maleate (0.56-1 μ m) and pyruvate (0.1–0.18 μ m). Sources of the inorganics are well documented with SO₄²⁻ and 414 415 NH4⁺ produced by precursor vapors SO₂ and NH₃, respectively, with ocean-emitted dimethylsulfide (DMS) as an additional precursor to SO_4^{2-} and the primary precursor to MSA. 416

417 Precursors leading to secondarily-produced alkyl amines such as DMA, TMA, 418 and DEA likely originated from a combination of industrial activity, marine emissions, biomass 419 burning, vehicular activity, sewage treatment, waste incineration, and the food industry (e.g., 420 Facchini et al., 2008; Sorooshian et al., 2009; Ge et al., 2011; VandenBoer et al., 2011); another 421 key source of these species, animal husbandry (Mosier et al., 1973; Schade and Crutzen, 1995; 422 Sorooshian et al., 2008), was ruled out owing to a scarcity of such activity in the study region. 423 Secondarily-produced amine salts likely were formed with SO_4^{2-} as the chief anion 424 owing to its much higher concentrations relative to NO_3^- or organic acids.

425 Dimethylamine was the most abundant amine similar to other marine (Muller et al., 2009) 426 and urban regions (Youn et al., 2015); the average concentration of DMA integrated over all 427 MOUDI stages for all sample sets was 62.2 ng m⁻³ in contrast to 29.8 ng m⁻³ for TMA+DEA. For 428 reference, the other key cation (NH₄⁺) participating in salt formation with acids such as H₂SO₄ and 429 HNO₃ was expectedly much more abundant (1.64 μ g m⁻³). With regard to the competitive uptake of DMA versus NH₃ in particles, the molar ratio of DMA:NH₄⁺ exhibited a unimodal profile between 0.1–1.8 µm with a peak of 0.022 between 0.32–0.56 µm and the lowest values at the tails (0.004 between 0.1–0.18 and 1–1.8 µm); DMA was not above detection limits for either $D_p < 0.1$ µm or $D_p > 1.8$ µm. The molar ratios observed were consistent with values measured in urban air of Tucson, Arizona and coastal air in Marina, California (0–0.04; Youn et al., 2015) and near the lower end of the range measured in rural and urban air masses sampled near Toronto (0.005–0.2: VandenBoer et al., 2011).

The most abundant organic acid was oxalate ($195 \pm 144 \text{ ng m}^{-3}$), followed by succinate (21 437 \pm 41 ng m⁻³), phthalate (19 \pm 25 ng m⁻³), maleate (17 \pm 15 ng m⁻³), and adipate (5 \pm 8 ng m⁻³). The 438 439 observation of mass concentrations increasing with decreasing carbon number for dicarboxylic 440 acids (i.e., oxalate > succinate > adipate) is consistent with many past studies for other regions as larger chain acids undergo oxidative decay to eventually form oxalate (e.g., Kawamura and 441 442 Ikushima, 1993; Kawamura and Sakaguchi, 1999; Sorooshian et al., 2007). Maleate is an 443 unsaturated dicarboxylic acid emitted from gas and diesel engines (Rogge et al., 1993) and a 444 product from the photo-oxidation of benzene (Kawamura and Ikushima, 1993). The aromatic 445 dicarboxylic acid phthalate is a known photo-oxidation product of naphthalene and stems largely 446 from plastic processing and fuel combustion (Fraser et al., 2003; Kautzman et al., 2010; Fu et al., 447 2012; Kleindienst et al., 2012). The oxidation product (MSA) of ocean-derived DMS exhibited an overall average concentration of 11 ± 7 ng m⁻³, which is near the lower end of the range of levels 448 reported in other coastal and marine environments (from undetected up to ~200 ng m⁻³) (e.g., 449 450 Saltzman et al., 1983, 1986; Berresheim 1987; Watts et al., 1987; Burgermeister and Georgii, 451 1991; Sorooshian et al., 2015; Xu and Gao, 2015).

452 Water-soluble species exhibiting a peak in the supermicrometer range, usually between 1.8–5.6 μ m, include those with known affiliations with sea salt (Na⁺, Cl⁻, K+, Mg²⁺) and crustal 453 materials such as dust (Ca²⁺). Nitrate peaked between 1.8-3.2 µm, and was best correlated with 454 Na⁺ and Mg²⁺, suggestive of HNO₃ partitioning to sea salt as has been observed in other coastal 455 456 regions (e.g., Prabhakar et al., 2014a). There was very little NO_3^{-1} in the submicrometer range (0.05 \pm 0.04 µg m⁻³) in contrast to supermicrometer sizes (0.78 \pm 0.47 µg m⁻³). More submicrometer 457 NO₃⁻ in the form of NH₄NO₃ would be expected if there was an excess of NH₃ after neutralizing 458 SO_4^{2-} . The mean ammonium-to-sulfate molar ratio for submicrometer sizes was 2.32 ± 0.52 (range: 459 1.11 - 2.78), with full neutralization of SO₄²⁻ in 10 of 12 MOUDI sets. Thus, there was a non-460

461 negligible excess in NH₃ that presumably participated in salt formation with HNO₃ and organic 462 species. The significant levels of NO₃⁻ in the same mode as Na⁺ and Cl⁻ contributed to the 463 significant Cl⁻ depletion observed, as the mean Cl⁻:Na⁺ mass ratio between 1-10 μ m (i.e., range of 464 peak sea salt influence) was 0.81 ± 0.28, which is much lower than the ratio for pure sea salt (1.81) 465 (Martens et al., 1973). The subject of Cl⁻ depletion in this region will be investigated more 466 thoroughly in subsequent work.

467 Figure 6 shows SEM images of representative single particles in each supermicrometer 468 stage. As would be expected for sea salt and crustal material, most of the particles shown are not 469 spherical. Interestingly, only the particle shown between 1–1.8 µm was close to being spherical. 470 Its composition based on EDX analysis was accounted for mostly by carbon (93.7%) with lower 471 amounts of oxygen (5.8%) and Fe (0.5%). Sea salt particles were found in the next two stages 472 owing to the highest combined weight percentages of Na⁺ and Cl⁻ based on EDX analysis: 1.8–3.2 473 $\mu m = 36.9\%$; 3.2–5.6 $\mu m = 46.9\%$. The salt particles are not necessarily cubical but more rounded 474 with signs of agglomeration. These two particles were the only ones among the 11 MOUDI stages 475 exhibiting an EDX signal for S, with contributions amounting to $\sim 2\%$ in each particle. This may be linked to natural SO_4^{2-} existing in sea salt particles. Also, the particle between 3.2–5.6 µm 476 477 contained a trace amount of Sc (1%). The largest three particles ($\geq 5.6 \mu m$) were expectedly 478 irregularly shaped with both sharp and rounded edges, comprised mostly of oxygen, Al, Fe, and 479 Ca based on EDX analysis.

480 **3.2.3 Water-Soluble Elements**

481 Averaged data across all MOUDI sets reveal that ICP-QQQ elements exhibited a variety 482 of mass concentration profiles ranging from a distinct mode in either the sub- or supermicrometer 483 range to having multiple modes below and above 1 µm (averages across all MOUDI sets shown 484 in Figure 7). There were several elements with only one distinct peak, being in one of the two 485 stages between 0.18-1.0 µm, including As, Cd, Co, Cr, Cs, Cu, Hf, Mn, Mo, Ni, Rb, Se, Sn, Tl, V, 486 Pb, and Zn. In contrast, the following elements exhibited only one distinct peak in the 487 supermicrometer range: Al, Ba, P, Sr, Ti, Y, and Zr. The rest of the elements exhibited more 488 complex behavior with two distinct peaks in the sub- and supermicrometer range (Ag, Fe, Nb). 489 The following section discusses relationships between all of the ions and elements with a view 490 towards identifying characteristic sources.

491 **3.3 Characteristic Sources and Species Relationships**

492 A combination of PMF and correlation analysis helped identify clusters of closely-closely-493 related species stemming from distinct sources. The final PMF solution with five factors, based on 494 five groups of species (Figure 8), was chosen because it passed the criteria of physical meaningfulness associated with being physically valid and it had a the close proximity of the 495 496 calculated ratio of Q_{true}:Q_{expected} (1.2) that was very close to the theoretical value of to 1.0. There 497 was a high coefficient of variation determination between measured and predicted mass concentration when summing up all species for each MOUDI stage ($r^2 = 0.79$; sample size, n =498 499 132), which added confidence in relying on the PMF model for source apportionment of PM. The 500 five distinct clusters were named for their most plausible sources based on the species included in 501 the groupings, with their overall contributions to the total mass based on PMF analysis shown in 502 parenthesis (Table 3): Aged/Transported (48.0%), Sea Salt (22.5%), Combustion (18.7%), 503 Vehicular/Resuspended Dust (5.6%), and Waste Processing (5.1%). For reference, a previous 504 study near the northwestern edge of the Philippines identified six source factors for $PM_{2.5}$ that are 505 fairly similar to those here (Bagtasa et al., 2018): sea salt, resuspended fine dust, local solid waste 506 burning, and long range transport of (i) industrial emissions, (ii) solid waste burning, and (iii) 507 secondary sulfate. Each of our five groupings will be discussed in detail below in decreasing order 508 of contribution to total measured mass concentrations.

509 3.3.1 Aged/Transported Aerosol

510 Although not due to one individual source, there was a distinct PMF factor that included species commonly produced via gas-to-particle conversion processes (NH4⁺, SO4²⁻, MSA, 511 512 oxalate). Correlation analysis (Table 4) also pointed to a large cluster of species significantly 513 related to each other, including the aforementioned ions and a suite of other organic acids 514 (phthalate, succinate, adipate), MSA, and DMA. The latter three inorganic and organic acid ions 515 exhibited significant correlations with each other ($r \ge 0.68$), but also with several elements ($r \ge$ 516 0.36: K, V, Rb, Cs, Sn), which were likely co-emitted with the precursor vapors of the secondarily 517 produced ions. Although BC concentrations were quantified from set MO13 only, their 518 interrelationships with water soluble ions from simultaneously collected set MO14 are 519 representative for other sets. The results showed that BC was significantly correlated (r: 0.61-520 0.92) with 15 species, including those mentioned above (owing to co-emission) and also a few

elements that were found via PMF to be stronger contributors to the Combustion source discussedin Section 3.3.3 (Ni, Cu, As, Se, Cd, Tl, Pb).

523 This PMF source factor is referred to as Aged Aerosol/Transported owing to its 524 characteristic species being linked to secondary particle formation from emissions of local and 525 regional sourcessources distant from the sample site. Examples include MSA and DMA being secondarily produced from ocean-derived gaseous emissions (e.g., Sorooshian et al., 2009), and K 526 527 stemming from biomass burning emissions from upwind regions such as Sumatra and Borneo 528 (Xian et al., 2013). Previous studies (Reid et al., 2012; Wang et al., 2013) have shown that 529 phenomena such as SWM and El-Nino events not only influence biomass burning activities in the 530 Malay Peninsula but also impact the transport and distribution of emissions in the study region. 531 For instance, Reid et al. (2016b) showed that enhancement in monsoonal flow facilitates the 532 advection of biomass burning and anthropogenic emissions to the Philippines from Sumatra and 533 Borneo. Subsequent work will investigate more deeply the impact of biomass burning from those 534 upwind regions on the sample site during the SWM.

While NH_4^+ and SO_4^{2-} require time for production owing to being secondarily-secondarily-535 536 produced from precursor vapors (i.e., SO₂, NH₃), oxalate is the smallest dicarboxylic acid and requires lengthier chemistry pathways for its production and thus is more likely produced in 537 538 instances of aerosol transport and aging (e.g., Wonaschuetz et al., 2012; Ervens et al., 2018). The 539 various elements associated with this cluster are co-emitted with the precursors to the 540 aforementioned ions and are linked to a variety of sources: metallurgical processes (Anderson et 541 al., 1988; Csavina et al., 2011; Youn et al., 2016), fuel combustion (Nriagu, 1989; Allen et al., 542 2001; Shafer et al., 2012; Rocha and Correa, 2018), residual oil combustion (Watson et al., 2004), 543 biomass burning (Maudlin et al., 2015), marine and terrestrial biogenic emissions (Sorooshian et 544 al., 2015), and plastics processing (Fraser et al., 2003). In addition, there is extensive ship traffic 545 in the general study region, which is a major source of species in this cluster of species, particularly V and SO_4^{2-} (e.g., Murphy et al., 2009; Coggon et al., 2012). 546

547 PMF analysis suggested that the Aged <u>Aerosol/Transported</u> factor contributed 48.0% to 548 the total water-soluble mass budget during the study period. Most of the contribution resided in 549 the submicrometer range (68.9%) unlike the supermicrometer range (18.6%), which is consistent 550 with the overall mass size distribution of total PM peaking in the submicrometer range (Figure 2). 551 The reconstructed mass size distribution for this PMF source factor shows the dominance of the mass in the submicrometer range with a peak between $0.32-0.56 \mu m$ (Figure 9). The correlation matrices for the sub- and supermicrometer size ranges also show that the correlations between the species most prominent in the Aged<u>Aerosol/Transported</u> category are stronger for the former size range (Tables S2-S3). The contribution of this PMF factor to the supermicrometer range is likely associated with species secondarily produced on coarse aerosol such as dust and sea salt. This is evident in the individual species mass size distributions where there is a dominant submicrometer mode but also non-negligible mass above 1 μm .

559 Even though the PM in a heavily populated urban region, such as Metro Manila, is typically 560 thought to be dominated by local sources of aerosols, the current PMF results show that the largest 561 contributions to water-soluble aerosol mass are from Aged/Transported pollution.contribution 562 from long range transport is still discernible. This finding is contrary to the expectation that (a) the 563 signal of transported aerosols would be lost in the noise of locally-produced aerosols, and (b) the 564 removal of aerosols over the ocean surrounding the Philippines by processes such as wet 565 scavenging would significantly reduce the contribution of transported aerosols. Even though other 566 cities may have different pollution signatures, varying in pollutant type and amount, this 567 phenomenon of Aged/Transported pollution forming a significant portion of the water-soluble mass may be applicable to other cities, especially those in Southeast Asia. 568

569 3.3.2 Sea Salt

570 As the MO sampling site is approximately 13 km from the nearest shoreline (Figure 1a) 571 and downwind of Manila Bay in the SWM season, there was a great potential for marine emissions 572 to impact the samples. There were several species with similar mass size distributions (mode: 1.8– 573 5.6 µm) and highly correlated total mass concentrations (r \ge 0.51) that are linked to sea salt: Cl⁻, Na^+ , Ca^{2+} , Mg^{2+} , Ba, and Sr. The correlations between these species were stronger when examining 574 575 just the supermicrometer range as compared to the submicrometer range (Tables S2-S3). The 576 majority of these species was used in PMF analysis and formed a distinct cluster amounting to 577 22.0% of the total study period's mass budget. This source contributed only 0.6% to the 578 submicrometer mass concentration but 53.5% for the supermicrometer size range. The 579 reconstructed mass size distribution for this source factor is shifted farthest to the larger diameters 580 as compared to the other four sources with a peak between $1.8-3.2 \mu m$ (Figure 9).

581 It is noteworthy that this factor has the highest share of NO_3^- among all identified sources. 582 This result is consistent with mass size distributions shown in Figure 5 in which NO_3^- peaks in the 583 supermicrometer range similar to sea salt constituents (e.g., Na⁺ and Cl⁻). Although sea salt particles naturally contain NO₃⁻ (Seinfeld and Pandis, 2016) (mass ratio of NO₃⁻:Na⁺ = 9.8×10^{-8} 584 -6.5×10^{-5}), the extremely high ratio of NO₃: Na⁺ (mass ratio ~1.8) suggests that only a negligible 585 586 portion of NO_3^{-1} in this factor originated from primary sea salt particles. Thus, the majority of NO_3^{-1} 587 is most likely due to HNO₃ partitioning to existing sea salt particles (e.g., Fitzgerald, 1991; Allen 588 et al., 1996; Dasgupta et al., 2007; Maudlin et al., 2015). In addition, the Cl⁻:Na⁺ mass ratio in this 589 profile (0.65) is smaller than that in sea salt particles (1.81), indicating high Cl^{-} depletion mainly 590 due to reactions of HNO₃ with NaCl (Ro et al., 2001; Yao et al., 2003; Braun et al., 2017). 591 Moreover, elevated loadings of trace elements (e.g., Ba, Cu, Zn, and Co) could be linked to mixing 592 of marine emissions with urban sources (e.g., vehicle and industrial emissions) during their 593 transport inland to the sampling site (Roth and Okada, 1998). This process of aging is consistent 594 with the observed morphology of the sea salt particles in this study, revealing non-cubical shapes 595 that are rounded owing to the likely addition of acidic species such as HNO_3 (Figure 6).

3.3.3 Combustion

597 There are numerous sources of combustion in the study region, including a variety of 598 mobile sources (e.g., cars, utility vehicles, trucks, buses, motorcycles) and stationary sources (e.g., 599 power stations, cement works, oil refineries, boiler stations, utility boilers). Consequently, the next 600 highest contributor to total mass during the study period according to PMF (18.7%) was the cluster 601 of species including Ni, As, Co, P, Mo, and Cr, which is defined as the Combustion factor. These 602 species have been reported to be rich in particles emitted from combustion of fossil fuel and 603 residual oil (Linak and Miller, 2000; Allen et al., 2001; Wasson et al., 2005; Mahowald et al., 604 2008; Mooibroek et al., 2011; Prabhakar et al., 2014b). Although not included in PMF analysis, 605 other species significantly correlated with the previous ones include maleate and Ag, which also 606 stem from fuel combustion (Kawamura and Kaplan, 1987; Lin et al., 2005; Sorooshian et al., 607 2007). Ag specifically is an element in waste incinerator fly ash (Buchholz and Landsberger, 1993; 608 Tsakalou et al., 2018) and its strong correlation with Co (r = 0.85) and Mo (r = 0.64) provides 609 support for this source factor being linked to combustion processes. Maleate is commonly found in engine exhaust (Kawamura and Kaplan, 1987), while Cr is a tracer for power plant emissions 610

611 (Singh et al., 2002; Behera et al., 2015). Of all species examined in this study, BC was best 612 correlated with As (r = 0.92), while its correlation with Ni (r = 0.85) was among the highest.

As the elements in this cluster peaked in concentration in the submicrometer mode, the weight percentage of this factor is more than double below 1 μ m (23.9%) as compared to above 1 μ m (11.3%). The reconstructed mass size distribution for this source factor peaks between 0.18– 0.32 μ m, which is smaller than the modal diameter range for the Aged/Transported source factor (0.32–0.56 μ m) likely owing to closer sources and thus less time for growth to occur via condensation and coagulation.

619 3.3.4 Vehicular/Resuspended Dust

620 The next PMF source factor contains chemical signatures of dust because of high 621 contributions to Al, Ti, Ca, and Fe. These crustal elements are strongly related to resuspension of 622 dust by traffic and construction activities (Singh et al., 2002; Harrison et al., 2011). Other elements 623 that were prominent in this factor included Zr, Y, Mn, Cr, and Ba, which are associated with tire 624 and brake wear (Adachi and Tainosho, 2004; Gietl et al., 2010; Song and Gao, 2011; Harrison et 625 al., 2012; Vossler et al., 2016), although some of them can be linked to the exhaust as well (e.g., 626 Lin et al., 2005; Song and Gao, 2011). This source is named Vehicular/Resuspended Dust and 627 contributed 5.6% to the total study period's mass concentrations.

628 The weight percentage contribution of this factor was much higher for the supermicrometer 629 range (11.3%) as compared to the submicrometer range (1.5%), which is consistent with the Sea 630 Salt source factor owing to similar mass size distributions of the individual species associated with 631 the two source categories (Figures 5 and 7). Additional species correlated significantly with the 632 crustal species included Hf and Nb, which also exhibited mass peaks between $1.8-3.2 \mu m$. The 633 reconstructed mass size distribution for this source factor is similar to that of Sea Salt in that there 634 is a peak between 1.8–3.2 µm, but there is less of a unimodal profile owing to what appears to be 635 a secondary mode between 0.56–1.0 µm (Figure 9), which could be linked to some of the non-dust 636 components of vehicular emissions.

637 **3.3.5 Waste Processing**

The final PMF source factor, contributing the least overall to total mass (5.1%), featuredZn, Cd, Pb, Mn, and Cu as its main components. These species are linked to waste processing,

including especially electronic waste (e-waste) and battery burning and recycling (Gullett et al.,
2007; Iijima et al., 2007), which was previously reported for Manila (Pabroa et al., 2011). The
latter study reported that although there are a few licensed operations for battery recycling, there
are numerous unregulated cottage melters across Manila that regularly melt metal from batteries
and discard the waste freely. Fujimori et al. (2012) additionally showed that e-waste recycling led
to emissions of the following elements (in agreement with this PMF cluster) around Metro Manila:
Ni, Cu, Pb, Zn, Cd, Ag, in, As, Co, Fe, and Mn.

647 This was the only PMF factor exhibiting comparable weight percentages both below 648 (5.1%) and above 1 μ m (5.3%). This is reflected in the mass size distributions of the species 649 included in this cluster being fairly uniformly distributed below and above 1 μ m. This is also 650 demonstrated in the reconstructed mass size distribution of this source factor as it clearly exhibits 651 a mode between the other four sources $(0.56-1.0 \ \mu\text{m})$ and is the broadest mode (Figure 9). The 652 explanation for this is likely rooted in the diversity of sources contained within this source profile 653 that lead to different sizes of particles. Examples of such sources include processing of different 654 types of waste at varying temperatures and through various processes (e.g., burning, melting, 655 grinding) (Keshtkar and Ashbaugh, 2007),

656 **4. Conclusions**

This study used various analytical techniques (gravimetry, ICion chromatography, triple quadrupole inductively coupled plasma mass spectrometryICP-QQQ, black carbon spectroscopy, and microscopy), meteorological data, and a source apportionment model (Positive_Matrix Factorization) to characterize the sources, chemical composition, and morphology of size-resolved ambient particulate matter (PM) collected using Micro-Orifice Uniform Deposit Impactors (MOUDIs) in Metro Manila, Philippines during the southwest monsoon season (SWM) season of 2018. The main results of this study include the following:

664

The total mass concentrations were measured on two occasions and were 18.6 µg m⁻³ and 53.0 µg m⁻³. Water-soluble mass concentrations were measured on 12 occasions and were on average 8.53 ± 4.48 µg m⁻³ (range = 2.7–16.6 µg m⁻³). Simultaneous measurements of total, water-soluble, and <u>black carbon (BC)</u> mass revealed a composition of 26.9% BC, 31.3% water-soluble components, and 41.8% unaccounted mass.

23

- Size-resolved BC mass concentration was measured on one occasion, with the mass sum of all MOUDI stages reaching 14.3 μ g m⁻³. Most of the BC mass (95%) was contained in the 0.1–1 μ m range (i.e., the Greenfield gap) where wet scavenging by rain is <u>relatively</u> inefficient. The measured BC peaked in the size range of 0.18 – 0.32 μ m and accounted for 51.8% of the measured PM for that stage. In the range of 0.10 – 0.18 μ m, the mass percent contribution of BC to the measured PM was 78.1%.
- Most of the total mass resided in the submicrometer mode (0.32–0.56 μ m); however, one MOUDI set revealed an additional supermicrometer mode (1.8–3.2 μ m). Water-soluble species that peaked in the submicrometer mode were associated with secondarily produced species, including inorganic acids, amines, <u>Methanesulfonate (MSA)</u>, and organic acids. Water-soluble species that peaked in the supermicrometer mode were associated with sea salt and crustal material. Most of the unaccounted mass was for D_p > 0.32 μ m.
- The most abundant water-soluble species was SO_4^{2-} (44% ± 6%), followed by NH₄⁺ (18% ± 5%), NO₃⁻ (10 ± 3%), Na⁺ (8 ± 3%), and Cl⁻ (6% ± 3%). Correlation analysis revealed that total water-soluble mass was most correlated with temperature (r = 0.64) and rainfall accumulation (r = -0.49) among meteorological factors considered, although other factors were likely influential such as wind direction and speed.
- Regardless of particle size, the majority of single particles examined with <u>energy dispersive</u>
 X-ray spectroscopy (SEM-EDX) were non-spherical with evidence of agglomeration.
- 689 PMF analysis suggested that there were five factors influencing the water-soluble fraction of 690 PM collected at the sampling site. These factors, their contribution to total water-soluble mass, 691 and the main species that permit them to be linked to a physical source are as follows: Aged Aerosol/Transported (48.0%; NH4⁺, SO4²⁻, MSA, oxalate), Sea Salt (22.5%; Cl⁻, NO3⁻, Ca²⁺, 692 Na⁺, Mg²⁺, Ba, Sr), Combustion (18.7%; Ni, As, Co, P, Mo, Cr), Vehicular/Resuspended Dust 693 694 (5.6%; Al, Ti, Fe), and Waste Processing (5.1%; Zn, Cd, Pb, Mn, Cu). The dominant 695 contribution of Aged/Transported aerosols to water-soluble mass contradicts two expectations: 696 (i) locally-produced sources in polluted cities should drown out the signal of transported 697 aerosols, and (ii) the signal of transported aerosols should be significantly reduced due to 698 scavenging processes upwind of the measurement site.
- 699

700 Although the current study focuses exclusively on the SWM season in Metro Manila, 701 results of this study are applicable to the study of aerosol impacts on Southeast Asia and other 702 regions. First, the significant presence detection of Aged/Transported aerosols not only from local 703 but also from regional sources in Metro Manila indicates confirms previous studies that PM in the 704 region has the ability to travel long distances during the SWM season, despite the typical 705 assumption that wet scavenging effectively removes most of the particles. Characterization of 706 aerosols in Metro Manila is therefore important for better understanding the impacts that local 707 emissions will have on locations downwind of Metro Manila, including other populated cities in 708 Southeast and East Asia. Transport of pollution and decreased wet scavenging during the SWM 709 season may become increasingly important as studies have shown a decrease in SWM rainfall and 710 increase in the number of no-rain days during the SWM season in the western Philippines in recent decades (e.g., Cruz et al., 2013). 711

Second, Southeast Asia has been named "one of the most hostile environments on the planet for aerosol remote sensing" (Reid et al., 2013) <u>because of high cloud occurrence</u>. Therefore, space-based remote sensing of aerosol characteristics, such as retrievals of aerosol optical depth (AOD), in this region are difficult. In situ measurements are critical for characterization of PM in this region, especially during seasons such as the SWM when clouds are especially prevalent and remote-sensing retrievals dependent on clear-sky conditions are lacking.

Third, this study provides a valuable dataset to compare to other regions impacted by monsoons where the impacts of enhanced moisture and rainfall on size-resolved composition are not well understood. As aqueous processing results in enhanced production of water-soluble species (e.g., sulfate, organic acids), it is noteworthy for this monsoonal region that the watersoluble fraction remains low relative to BC and other insoluble components. This has major implications for the hygroscopicity of the regional PM.

Finally, the results of this study will be used to inform future sampling campaigns in this region, including CAMP²Ex planned for the SWM season of 2019 based in the Philippines. As the current MOUDI sampling campaign at <u>the Manila Observatory</u> is expected to extend for a full year, future work will focus on changes in aerosol characteristics and sources on a seasonal basis.

729 *Data availability:* All data used in this work are available upon request.

730

731 Author Contribution: MTC, MOC, JBS, ABM, CS, and AS designed the experiments and all co-

- authors carried out some aspect of the data collection. MTC, RAB, CS, LM, HD, and AS conducted
- data analysis and interpretation. MTC and AS prepared the manuscript with contributions from all
- co-authors.
- 735
- 736 *Competing interests:* The authors declare that they have no conflict of interest.
- 737

Acknowledgements: This research was funded by NASA grant 80NSSC18K0148. M. T. Cruz
acknowledges support from the Philippine Department of Science and Technology's ASTHRD
Program. R. A. Braun acknowledges support from the ARCS Foundation. A. B. MacDonald
acknowledges support from the Mexican National Council for Science and Technology
(CONACYT). We acknowledge Agilent Technologies for their support and Shane Snyder's
laboratories for ICP-QQQ data.

- 744
- 745 **References**
- Adachi, K., and Tainosho, Y.: Characterization of heavy metal particles embedded in tire dust,
 Environ Int, 30, 1009-1017, 10.1016/j.envint.2004.04.004, 2004.
- 748

749 Alas, H. D., Müller, T., Birmili, W., Kecorius, S., Cambaliza, M.O., Simpas, J. B., Cayetano, M., 750 Weinhold, K., Vallar, E., Galvez, M.C., and Wiedensohler, A.: Spatial Characterization of Black 751 Carbon Mass Concentration in the Atmosphere of a Southeast Asian Megacity: An Air Quality 752 Philippines. Case Study for Metro Manila, Aerosol Air Oual Res. 753 doi.org/10.4209/aaqr.2017.08.0281, 2017. 754

- Allen, H. C., Laux, J. M., Vogt, R., FinlaysonPitts, B. J., and Hemminger, J. C.: Water-induced
 reorganization of ultrathin nitrate films on NaCl: Implications for the tropospheric chemistry of
 sea salt particles, J Phys Chem-Us, 100, 6371-6375, DOI 10.1021/jp953675a, 1996.
- 758
- Allen, A. G., Nemitz, E., Shi, J. P., Harrison, R. M., and Greenwood, J. C.: Size distributions of
 trace metals in atmospheric aerosols in the United Kingdom, Atmos Environ, 35, 4581-4591, Doi
 10.1016/S1352-2310(01)00190-X, 2001.
- 762
- Almeida, G. P., Bittencourt, A. T., Evangelista, M. S., Vieira-Filho, M. S., and Fornaro, A.:
 Characterization of aerosol chemical composition from urban pollution in Brazil and its possible
 impacts on the aerosol hygroscopicity and size distribution, Atmos Environ, 202, 149-159,
 10.1016/j.atmosenv.2019.01.024, 2019.
- 767
- Anderson, J. R., Aggett, F. J., Buseck, P. R., Germani, M. S., and Shattuck, T. W.: Chemistry of
 Individual Aerosol-Particles from Chandler, Arizona, an Arid Urban-Environment, Environ Sci
 Technol, 22, 811-818, DOI 10.1021/es00172a011, 1988.

- 771
- Aswini, A. R., Hegde, P., Nair, P. R., and Aryasree, S.: Seasonal changes in carbonaceous
- aerosols over a tropical coastal location in response to meteorological processes, Science of The
- 774 Total Environment, 656, 1261-1279, https://doi.org/10.1016/j.scitotenv.2018.11.366, 2019.
- 775
- Bagtasa, G., Cayetano, M. G., and Yuan, C. S.: Seasonal variation and chemical characterization
 of PM_{2.5} in northwestern Philippines, Atmos Chem Phys, 18, 4965-4980, 10.5194/acp-18-49652018, 2018.
- 778 2 779
- Bautista, A. T., Pabroa, P. C. B., Santos, F. L., Racho, J. M. D., and Quirit, L. L.: Carbonaceous
 particulate matter characterization in an urban and a rural site in the Philippines, Atmos Pollut Res,
 5, 245-252, 10.5094/Apr.2014.030, 2014.
- 783

Begam, G. R., Vachaspati, C. V., Ahammed, Y. N., Kumar, K. R., Reddy, R. R., Sharma, S. K.,

- 785 Saxena, M., and Mandal, T. K.: Seasonal characteristics of water-soluble inorganic ions and
- carbonaceous aerosols in total suspended particulate matter at a rural semi-arid site, Kadapa
- 787 (India), Environmental Science and Pollution Research, 24, 1719-1734, 10.1007/s11356-016-
- 788 7917-1, 2017.
- 789
- Behera, S. N., Betha, R., Huang, X., and Balasubramanian, R.: Characterization and estimation of
 human airway deposition of size-resolved particulate-bound trace elements during a recent haze
 episode in Southeast Asia, Environ Sci Pollut R, 22, 4265-4280, 10.1007/s11356-014-3645-6,
 2015.
- 794
- Berresheim, H.: Biogenic Sulfur Emissions from the Sub-Antarctic and Antarctic Oceans, J
 Geophys Res-Atmos, 92, 13245-13262, 10.1029/JD092iD11p13245, 1987.
- 797
- Braun, R. A., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M., Maudlin, L. C., Crosbie, E.,
 Aghdam, M. A., Mardi, A. H., and Sorooshian, A.: Impact of Wildfire Emissions on Chloride
 and Bromide Depletion in Marine Aerosol Particles, Environ Sci Technol, 51, 9013-9021,
 10.1021/acs.est.7b02039, 2017.
- 802
- Buchholz, B. A., and Landsberger, S.: Trace-Metal Analysis of Size-Fractioned Municipal SolidWaste Incinerator Fly-Ash and Its Leachates, J Environ Sci Heal A, 28, 423-441, Doi
 10.1080/10934529309375887, 1993.
- 806
- Burgermeister, S., and Georgii, H. W.: Distribution of Methanesulfonate, Nss Sulfate and
 Dimethylsulfide over the Atlantic and the North-Sea, Atmos Environ a-Gen, 25, 587-595, Doi
 10.1016/0960-1686(91)90056-D, 1991.
- 810
- Chen, Y. Z., Shah, N., Huggins, F. E., and Huffman, G. P.: Microanalysis of ambient particles
 from Lexington, KY, by electron microscopy, Atmos Environ, 40, 651-663,
 10.1016/j.atmosenv.2005.09.036, 2006.
- 814

- Chithra, V. S., and Nagendra, S. M. S.: Chemical and morphological characteristics of indoor and
 outdoor particulate matter in an urban environment, Atmos Environ, 77, 579-587,
 10.1016/j.atmosenv.2013.05.044, 2013.
- 818
- 819 Chuang, M.-T., Chang, S.-C., Lin, N.-H., Wang, J.-L., Sheu, G.-R., Chang, Y.-J., and Lee, C.-T.:
- Aerosol chemical properties and related pollutants measured in Dongsha Island in the northern South China Sea during 7-SEAS/Dongsha Experiment, Atmospheric Environment, 78, 82-92,
- 822 https://doi.org/10.1016/j.atmosenv.2012.05.014, 2013.
- 823
- 824 Clarke, A. D., Shinozuka, Y., Kapustin, V. N., Howell, S., Huebert, B., Doherty, S., Anderson,
- 825 T., Covert, D., Anderson, J., Hua, X., Moore, K. G., McNaughton, C., Carmichael, G., and
- 826 Weber, R.: Size distributions and mixtures of dust and black carbon aerosol in Asian outflow:
- 827 Physiochemistry and optical properties, J Geophys Res-Atmos, 109, Artn D15s09
- 828 10.1029/2003jd004378, 2004.
- 829
- 830 Coggon, M. M., Sorooshian, A., Wang, Z., Metcalf, A. R., Frossard, A. A., Lin, J. J., Craven, J.
- 831 S., Nenes, A., Jonsson, H. H., Russell, L. M., Flagan, R. C., and Seinfeld, J. H.: Ship impacts on
- the marine atmosphere: insights into the contribution of shipping emissions to the properties of marine aerosol and clouds, Atmos Chem Phys, 12, 8439-8458, 10.5194/acp-12-8439-2012, 2012.
- 833 834
- Cohen, D. D., Stelcer, E., Santos, F. L., Prior, M., Thompson, C., and Pabroa, P. C. B.:
 Fingerprinting and source apportionment of fine particle pollution in Manila by IBA and PMF
 techniques: A 7-year study, X-Ray Spectrom, 38, 18-25, 10.1002/xrs.1112, 2009.
- 838
- Crosbie, E., Sorooshian, A., Monfared, N. A., Shingler, T., and Esmaili, O.: A Multi-Year Aerosol
 Characterization for the Greater Tehran Area Using Satellite, Surface, and Modeling Data,
 Atmosphere-Basel, 5, 178-197, 10.3390/atmos5020178, 2014.
- 842
- Crosbie, E., Youn, J. S., Balch, B., Wonaschutz, A., Shingler, T., Wang, Z., Conant, W. C.,
 Betterton, E. A., and Sorooshian, A.: On the competition among aerosol number, size and
 composition in predicting CCN variability: a multi-annual field study in an urbanized desert,
 Atmos Chem Phys, 15, 6943-6958, 10.5194/acp-15-6943-2015, 2015.
- 847
- Cruz, F. T., Narisma, G. T., Villafuerte, M. Q., Chua, K. U. C., and Olaguera, L. M.: A
 climatological analysis of the southwest monsoon rainfall in the Philippines, Atmos Res, 122, 609616, 10.1016/j.atmosres.2012.06.010, 2013.
- 851
- 852 Csavina, J., Landazuri, A., Wonaschutz, A., Rine, K., Rheinheimer, P., Barbaris, B., Conant, W.,
- Saez, A. E., and Betterton, E. A.: Metal and Metalloid Contaminants in Atmospheric Aerosols
 from Mining Operations, Water Air Soil Poll, 221, 145-157, 10.1007/s11270-011-0777-x, 2011.
- 855
- 856 Dasgupta, P. K., Campbell, S. W., Al-Horr, R. S., Ullah, S. M. R., Li, J. Z., Amalfitano, C., and
- 857 Poor, N. D.: Conversion of sea salt aerosol to NaNO(3) and the production of HCl: Analysis of
- temporal behavior of aerosol chloride/nitrate and gaseous HCl/HNO(3) concentrations with
- AIM, Atmos Environ, 41, 4242-4257, 10.1016/j.atmosenv.2006.09.054, 2007.
- 860

- 861 Dumanoglu, Y., Kara, M., Altiok, H., Odabasi, M., Elbir, T., and Bayram, A.: Spatial and seasonal 862 variation and source apportionment of volatile organic compounds (VOCs) in a heavily 863 industrialized region, Atmos Environ, 98, 168-178, 10.1016/j.atmosenv.2014.08.048, 2014. 864 865 Ervens, B., Sorooshian, A., Aldhaif, A. M., Shingler, T., Crosbie, E., Ziemba, L., Campuzano-866 Jost, P., Jimenez, J. L., and Wisthaler, A.: Is there an aerosol signature of chemical cloud 867 processing?, Atmos Chem Phys, 18, 16099-16119, 10.5194/acp-18-16099-2018, 2018. 868 869 Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S., 870 Moretti, F., Tagliavini, E., Ceburnis, D., and O'Dowd, C. D.: Important Source of Marine 871 Secondary Organic Aerosol from Biogenic Amines, Environ Sci Technol, 42, 9116-9121, 872 10.1021/es8018385, 2008. 873 874 Farren, N. J., Dunmore, R. E., Mead, M. I., Mohd Nadzir, M. S., Samah, A. A., Phang, S. M., 875 Bandy, B. J., Sturges, W. T., and Hamilton, J. F.: Chemical characterisation of water-soluble 876 ions in atmospheric particulate matter on the east coast of Peninsular Malaysia, Atmos. Chem. 877 Phys., 19, 1537-1553, 10.5194/acp-19-1537-2019, 2019. 878 879 Fitzgerald, J. W.: Marine Aerosols - a Review, Atmos Environ a-Gen, 25, 533-545, Doi 880 10.1016/0960-1686(91)90050-Н, 1991. 881 882 Fraser, M. P., Cass, G. R., and Simoneit, B. R. T.: Air quality model evaluation data for organics. 883 6. C-3-C-24 organic acids, Environ Sci Technol, 37, 446-453, 10.1021/es0209262, 2003. 884 885 Fu, P. Q., Kawamura, K., Chen, J., Li, J., Sun, Y. L., Liu, Y., Tachibana, E., Aggarwal, S. G., 886 Okuzawa, K., Tanimoto, H., Kanaya, Y., and Wang, Z. F.: Diurnal variations of organic 887 molecular tracers and stable carbon isotopic composition in atmospheric aerosols over Mt. Tai in 888 the North China Plain: an influence of biomass burning, Atmos Chem Phys, 12, 8359-8375, 889 10.5194/acp-12-8359-2012, 2012. 890 891 Fujimori, T., Takigami, H., Agusa, T., Eguchi, A., Bekki, K., Yoshida, A., Terazono, A., and 892 Ballesteros, F. C.: Impact of metals in surface matrices from formal and informal electronic-893 waste recycling around Metro Manila, the Philippines, and intra-Asian comparison, J Hazard 894 Mater, 221-222, 139-146, https://doi.org/10.1016/j.jhazmat.2012.04.019, 2012. 895 896 Ge, X. L., Wexler, A. S., and Clegg, S. L.: Atmospheric amines - Part I. A review, Atmos Environ, 897 45, 524-546, 10.1016/j.atmosenv.2010.10.012, 2011. 898 899 Gietl, J. K., Lawrence, R., Thorpe, A. J., and Harrison, R. M.: Identification of brake wear particles 900 and derivation of a quantitative tracer for brake dust at a major road, Atmos Environ, 44, 141-146, 901 10.1016/j.atmosenv.2009.10.016, 2010. 902
- Gullett, B. K., Linak, W. P., Touati, A., Wasson, S. J., Gatica, S., and King, C. J.: Characterization
 of air emissions and residual ash from open burning of electronic wastes during simulated
 rudimentary recycling operations, J Mater Cycles Waste, 9, 69-79, 10.1007/s10163-006-0161-x,
 2007.

- 908 Harrison, R. M., Beddows, D. C. S., and Dall'Osto, M.: PMF Analysis of Wide-Range Particle
- 909 Size Spectra Collected on a Major Highway, Environ Sci Technol, 45, 5522-5528,
- 910 10.1021/es2006622, 2011.
- 911
- 912 Harrison, R. M., Jones, A. M., Gietl, J., Yin, J. X., and Green, D. C.: Estimation of the
- 913 Contributions of Brake Dust, Tire Wear, and Resuspension to Nonexhaust Traffic Particles
- 914 Derived from Atmospheric Measurements, Environ Sci Technol, 46, 6523-6529,
- 915 10.1021/es300894r, 2012.
- 916
- 917 Hitzenberger, R., and Tohno, S.: Comparison of black carbon (BC) aerosols in two urban areas –
- 918 concentrations and size distributions, Atmospheric Environment, 35, 2153-2167,
- 919 https://doi.org/10.1016/S1352-2310(00)00480-5, 2001.
- 920
- 921 Hopke, P. K., Cohen, D. D., Begum, B. A., Biswas, S. K., Ni, B., Pandit, G. G., Santoso, M.,
- 922 Chung, Y. S., Davy, P., Markwitz, A., Waheed, S., Siddique, N., Santos, F. L., Pabroa, P. C. B.,
- Seneviratne, M. C. S., Wimolwattanapun, W., Bunprapob, S., Vuong, T. B., Duy Hien, P. and
 Markowicz, A.: Urban air quality in the Asian region, Sci. Total Environ., 404(1), 103–112,
 doi:10.1016/j.scitotenv.2008.05.039, 2008.
- 926
- Huang, S. L., Rahn, K. A., and Arimoto, R.: Testing and optimizing two factor-analysis techniques
 on aerosol at Narragansett, Rhode Island, Atmos Environ, 33, 2169-2185, Doi 10.1016/S13522310(98)00324-0, 1999.
- 930
- Iijima, A., Sato, K., Yano, K., Tago, H., Kato, M., Kimura, H., and Furuta, N.: Particle size and
 composition distribution analysis of automotive brake abrasion dusts for the evaluation of
 antimony sources of airborne particulate matter, Atmos Environ, 41, 4908-4919,
 10.1016/j.atmosenv.2007.02.005, 2007.
- 935
- Kahnert, M., Nousiainen, T., and Veihelmann, B.: Spherical and spheroidal model particles as an
 error source in aerosol climate forcing and radiance computations: A case study for feldspar
 aerosols, J Geophys Res-Atmos, 110, Artn D18s13, 10.1029/2004jd005558, 2005.
- 939
- Kautzman, K. E., Surratt, J. D., Chan, M. N., Chan, A. W. H., Hersey, S. P., Chhabra, P. S.,
 Dalleska, N. F., Wennberg, P. O., Flagan, R. C., and Seinfeld, J. H.: Chemical Composition of
 Gas- and Aerosol-Phase Products from the Photooxidation of Naphthalene, J Phys Chem A, 114,
 913-934, 10.1021/jp908530s, 2010.
- 944
- Kawamura, K., and Ikushima, K.: Seasonal-Changes in the Distribution of Dicarboxylic-Acids in
 the Urban Atmosphere, Environ Sci Technol, 27, 2227-2235, DOI 10.1021/es00047a033, 1993.
- 947
- Kawamura, K., and Kaplan, I. R.: Motor Exhaust Emissions as a Primary Source for DicarboxylicAcids in Los-Angeles Ambient Air, Environ Sci Technol, 21, 105-110, DOI
 10.1021/es00155a014, 1987.
- 951
- Kawamura, K., and Sakaguchi, F.: Molecular distributions of water soluble dicarboxylic acids in
 marine aerosols over the Pacific Ocean including tropics, J Geophys Res-Atmos, 104, 3501-3509,
- 954 Doi 10.1029/1998jd100041, 1999.

- 955
- 956 Kecorius, S., Madueño, L., Löndahl, J., Vallar, E., Galvez, M. C., Idolor, L. F., Gonzaga-
- 957 Cayetano, M., Müller, T., Birmili, W., and Wiedensohler, A.: Respiratory tract deposition of
- 958 inhaled roadside ultrafine refractory particles in a polluted megacity of South-East Asia, Science
- 959 of The Total Environment, 663, 265-274, https://doi.org/10.1016/j.scitotenv.2019.01.338, 2019.
- 960
- 961 Kecorius, S., Madueno, L., Vallar, E., Alas, H., Betito, G., Birmili, W., Cambaliza, M. O., Catipay,
- 962 G., Gonzaga-Cayetano, M., Galvez, M. C., Lorenzo, G., Muller, T., Simpas, J. B., Tamayo, E. G.,
- and Wiedensohler, A.: Aerosol particle mixing state, refractory particle number size distributions
 and emission factors in a polluted urban environment: Case study of Metro Manila, Philippines,
 Atmos Environ, 170, 169-183, 10.1016/j.atmosenv.2017.09.037, 2017.
- 966
- Keshtkar, H., and Ashbaugh, L. L.: Size distribution of polycyclic aromatic hydrocarbon
 particulate emission factors from agricultural burning, Atmos Environ, 41, 2729-2739,
 10.1016/j.atmosenv.2006.11.043, 2007.
- 970
- Kim Oanh, N. T., Upadhyay, N., Zhuang, Y. H., Hao, Z. P., Murthy, D. V. S., Lestari, P., Villarin,
 J. T., Chengchua, K., Co, H. X., Dung, N. T. and Lindgren, E. S.: Particulate air pollution in six
 Asian cities: Spatial and temporal distributions, and associated sources, Atmos. Environ., 40(18),
 3367–3380, doi:10.1016/j.atmosenv.2006.01.050, 2006.
- Kim Oanh, N. T., Pongkiatkul, P., Cruz, M. T., Trung Dung, N., Phillip, L., Zhang, G., and Lestari,
 P.: Monitoring and Source Apportionment for Particulate Matter Pollution in Six Asian Cities, in:
 Integrated Air Quality Management: Asian Case Studies, Kim Oanh, N. T. (Ed.), CRC Press,
 Taylor & Francis Group, USA, 97-124, 2013.
- 980

Kleindienst, T. E., Jaoui, M., Lewandowski, M., Offenberg, J. H., and Docherty, K. S.: The
formation of SOA and chemical tracer compounds from the photooxidation of naphthalene and its
methyl analogs in the presence and absence of nitrogen oxides, Atmos Chem Phys, 12, 8711-8726,
10.5194/acp-12-8711-2012, 2012.

- 985
- Liao, H., Chen, W. T., and Seinfeld, J. H.: Role of climate change in global predictions of future
 tropospheric ozone and aerosols, J Geophys Res-Atmos, 111, Artn D12304,
 10.1029/2005jd006852, 2006.
- 988 10. 989
- Lin, C. C., Chen, S. J., Huang, K. L., Hwang, W. I., Chang-Chien, G. P., and Lin, W. Y.:
 Characteristics of metals in nano/ultrafine/fine/coarse particles collected beside a heavily
 trafficked road, Environ Sci Technol, 39, 8113-8122, 10.1021/es048182a, 2005.
- 993
- Linak, W. P., and Miller, C. A.: Comparison of particle size distributions and elemental
 partitioning from the combustion of pulverized coal and residual fuel oil, J Air Waste Manage, 50,
 1532-1544, Doi 10.1080/10473289.2000.10464171, 2000.
- Ma, Y., Li, S., Zheng, J., Khalizov, A., Wang, X., Wang, Z., and Zhou, Y.: Size-resolved
 measurements of mixing state and cloud-nucleating ability of aerosols in Nanjing, China, Journal
 of Geophysical Research: Atmospheres, 122, 9430-9450, 10.1002/2017jd026583, 2017.
- 1001

- Ma, L., Dadashazar, D., Braun, R. A., MacDonald, A. B., Aghdam, M. A., Maudlin, L. C., and Sorooshian, A.: Size-resolved characteristics of water-soluble particulate elements in a coastal
- Sorooshian, A.: Size-resolved characteristics of water-soluble particulate elements in a coastal
 area: Source identification, influence of wildfires, and diurnal variability, Atmos. Environ., 206,
 72-84, https://doi.org/10.1016/j.atmosenv.2019.02.045, 2019.
- 1006
- 1007 Mahowald, N., Jickells, T. D., Baker, A. R., Artaxo, P., Benitez-Nelson, C. R., Bergametti, G.,
- 1008 Bond, T. C., Chen, Y., Cohen, D. D., Herut, B., Kubilay, N., Losno, R., Luo, C., Maenhaut, W.,
- 1009 McGee, K. A., Okin, G. S., Siefert, R. L., and Tsukuda, S.: Global distribution of atmospheric
- 1010 phosphorus sources, concentrations and deposition rates, and anthropogenic impacts, Global
- 1011 Biogeochem Cy, 22, 10.1029/2008gb003240, 2008.
- 1012
- Marple, V., Olson, B., Romay, F., Hudak, G., Geerts, S. M. and Lundgren, D.: Second generation
 micro-orifice uniform deposit impactor, 120 MOUDI-II: Design, Evaluation, and application to
 long-term ambient sampling, Aerosol Sci. Technol., 48(4), 427–433,
 doi:10.1080/02786826.2014.884274, 2014.
- 1017
- Martens, C. S., Wesolowski, J. J., Harriss, R. C., and Kaifer, R.: Chlorine Loss from Puerto-Rican
 and San-Francisco-Bay Area Marine Aerosols, J Geophys Res, 78, 8778-8792, DOI
 1020 10.1029/JC078i036p08778, 1973.
- Maudlin, L. C., Wang, Z., Jonsson, H. H., and Sorooshian, A.: Impact of wildfires on sizeresolved aerosol composition at a coastal California site, Atmos Environ, 119, 59-68,
 1024 10.1016/j.atmosenv.2015.08.039, 2015.
- 1025
- Metcalf, A. R., Craven, J. S., Ensberg, J. J., Brioude, J., Angevine, W., Sorooshian, A., Duong,
 H. T., Jonsson, H. H., Flagan, R. C., and Seinfeld, J. H.: Black carbon aerosol over the Los
- 1028 Angeles Basin during CalNex, J Geophys Res-Atmos, 117, 10.1029/2011jd017255, 2012.
- 1029
- Mielonen, T., Levy, R. C., Aaltonen, V., Komppula, M., de Leeuw, G., Huttunen, J., Lihavainen,
 H., Kolmonen, P., Lehtinen, K. E. J., and Arola, A.: Evaluating the assumptions of surface
 reflectance and aerosol type selection within the MODIS aerosol retrieval over land: the problem
 of dust type selection, Atmos Meas Tech, 4, 201-214, 10.5194/amt-4-201-2011, 2011.
- 1035 Miller, J., and Miller, J.C.: Statistics and chemometrics for analytical chemistry. Pearson 1036 Education, 2018.
- 1037
- 1038 Mishra, S. K., Agnihotri, R., Yadav, P. K., Singh, S., Prasad, M. V. S. N., Praveen, P. S., Tawale,
- 1039 J. S., Rashmi, Mishra, N. D., Arya, B. C., and Sharma, C.: Morphology of Atmospheric Particles 1040 over Semi-Arid Region (Jaipur, Rajasthan) of India: Implications for Optical Properties, Aerosol
- 1040 Over Semi-And Region (Japur, Rajasman) of mula. Implications for Optic 1041 Air Qual Res, 15, 974-+, 10.4209/aaqr.2014.10.0244, 2015.
 - 1042
 - 1043 Mooibroek, D., Schaap, M., Weijers, E. P., and Hoogerbrugge, R.: Source apportionment and 1044 spatial variability of PM_{2.5} using measurements at five sites in the Netherlands, Atmospheric 1045 Environment, 45, 4180-4191, 10.1016/j.atmosenv.2011.05.017, 2011.
 - 1046
 - Mosier, A. R., Andre, C. E., and Viets, F. G.: Identification of Aliphatic-Amines Volatilized from
 Cattle Feedyard, Environ Sci Technol, 7, 642-644, DOI 10.1021/es60079a009, 1973.

1049

- Muller, C., Iinuma, Y., Karstensen, J., van Pinxteren, D., Lehmann, S., Gnauk, T., and Herrmann,
 H.: Seasonal variation of aliphatic amines in marine sub-micrometer particles at the Cape Verde
 islands, Atmos Chem Phys, 9, 9587-9597, 2009.
- 1053
- Murphy, S. M., Agrawal, H., Sorooshian, A., Padro, L. T., Gates, H., Hersey, S., Welch, W. A.,
 Jung, H., Miller, J. W., Cocker, D. R., Nenes, A., Jonsson, H. H., Flagan, R. C., and Seinfeld, J.
 H.: Comprehensive Simultaneous Shipboard and Airborne Characterization of Exhaust from a
 Modern Container Ship at Sea, Environ Sci Technol, 43, 4626-4640, 10.1021/es802413j, 2009.
- 1058
- Norris, G., Duvall, R., Brown, S., and Bai, S.: EPA Positive Matrix Factorization (PMF) 5.0
 fundamentals and User Guide Prepared for the US Environmental Protection Agency Office of
 Research and Development, Washington, DC. Inc., Petaluma, 2014.
- 1063 Nriagu, J. O.: A Global Assessment of Natural Sources of Atmospheric Trace-Metals, Nature, 338,
 1064 47-49, DOI 10.1038/338047a0, 1989.
- 1065
- Pabroa, P. C. B., Santos, F. L., Morco, R. P., Racho, J. M. D., Bautista, A. T., and Bucal, C. G. D.:
 Receptor modeling studies for the characterization of air particulate lead pollution sources in
 Valenzuela sampling site (Philippines), Atmos Pollut Res, 2, 213-218, 10.5094/Apr.2011.027,
 2011.
- 1071 Philippine Statistics Authority: https://psa.gov.ph/, Accessed 28 August 2018.
- 1072
 1073 Polissar, A., Hopke, P., Paatero, P., Malm, W., and Sisler, J.: Atmospheric aerosol over Alaska 2.
 1074 Elemental composition and sources. Journal of Geophysical Research 103, 19045-19057, 1998.
 - 1075
 - Prabhakar, G., Ervens, B., Wang, Z., Maudlin, L. C., Coggon, M. M., Jonsson, H. H., Seinfeld, J.
 H., and Sorooshian, A.: Sources of nitrate in stratocumulus cloud water: Airborne measurements
 - 1078 during the 2011 E-PEACE and 2013 NiCE studies, Atmos Environ, 97, 166-173,
 - 1079 10.1016/j.atmosenv.2014.08.019, 2014a. 1080
 - Prabhakar, G., Sorooshian, A., Toffol, E., Arellano, A. F., and Betterton, E. A.: Spatiotemporal
 distribution of airborne particulate metals and metalloids in a populated arid region, Atmos
 - 1083 Environ, 92, 339-347, 10.1016/j.atmosenv.2014.04.044, 2014b. 1084
 - Qu, W. J., Wang, J., Zhang, X. Y., Wang, D., and Sheng, L. F.: Influence of relative humidity on
 aerosol composition: Impacts on light extinction and visibility impairment at two sites in coastal
 area of China, Atmos Res, 153, 500-511, 10.1016/j.atmosres.2014.10.009, 2015.
 - 1087
 - Raatikainen, T., Brus, D., Hyvärinen, A. P., Svensson, J., Asmi, E., and Lihavainen, H.: Black
 carbon concentrations and mixing state in the Finnish Arctic, Atmos. Chem. Phys., 15, 10057-
 - 1091 10070, 10.5194/acp-15-10057-2015, 2015.
 - 1092
 - 1093 Ramachandran, S., and Rajesh, T. A.: Black carbon aerosol mass concentrations over
 - 1094 Ahmedabad, an urban location in western India: Comparison with urban sites in Asia, Europe,
 - Canada, and the United States, J Geophys Res-Atmos, 112, 10.1029/2006jd007488, 2007.

1096

Ran, L., Deng, Z. Z., Wang, P. C., and Xia, X. A.: Black carbon and wavelength-dependent aerosol
absorption in the North China Plain based on two-year aethalometer measurements, Atmos
Environ, 142, 132-144, 10.1016/j.atmosenv.2016.07.014, 2016.

1100

Reddington, C. L., McMeeking, G., Mann, G. W., Coe, H., Frontoso, M. G., Liu, D., Flynn, M.,
Spracklen, D. V., and Carslaw, K. S.: The mass and number size distributions of black carbon
aerosol over Europe, Atmos. Chem. Phys., 13, 4917-4939, 10.5194/acp-13-4917-2013, 2013.

- 1105 aerosol över Europe, Atmos. Chem. Phys., 15, 4917-4959, 10.5194/acp-15-4917-2015, 2015. 1104
- Reff, A., Eberly, S.I., and Bhave, P.V.: Receptor modeling of ambient particulate matter data using
 positive matrix factorization: Review of existing methods. J Air Waste Manage 57, 146-154, 2007.
- Reid, J. S., Xian, P., Hyer, E. J., Flatau, M. K., Ramirez, E. M., Turk, F. J., Sampson, C. R., Zhang,
 C., Fukada, E. M., and Maloney, E. D.: Multi-scale meteorological conceptual analysis of observed
 active fire hotspot activity and smoke optical depth in the Maritime Continent, Atmos Chem Phys,
 12, 2117-2147, 10.5194/acp-12-2117-2012, 2012.
- 1112

Reid, J. S., Hyer, E. J., Johnson, R. S., Holben, B. N., Yokelson, R. J., Zhang, J. L., Campbell, J. 1113 R., Christopher, S. A., Di Girolamo, L., Giglio, L., Holz, R. E., Kearney, C., Miettinen, J., Reid, 1114 1115 E. A., Turk, F. J., Wang, J., Xian, P., Zhao, G. Y., Balasubramanian, R., Chew, B. N., Janjai, S., Lagrosas, N., Lestari, P., Lin, N. H., Mahmud, M., Nguyen, A. X., Norris, B., Oanh, N. T. K., Oo, 1116 1117 M., Salinas, S. V., Welton, E. J., and Liew, S. C.: Observing and understanding the Southeast 1118 Asian aerosol system by remote sensing: An initial review and analysis for the Seven Southeast 1119 Asian Studies (7SEAS) program, Atmos Res, 122, 403-468, 10.1016/j.atmosres.2012.06.005, 1120 2013.

1121

Reid, J. S., Xian, P., Holben, B. N., Hyer, E. J., Reid, E. A., Salinas, S. V., Zhang, J. L., Campbell,
J. R., Chew, B. N., Holz, R. E., Kuciauskas, A. P., Lagrosas, N., Posselt, D. J., Sampson, C. R.,
Walker, A. L., Welton, E. J., and Zhang, C. D.: Aerosol meteorology of the Maritime Continent
for the 2012 7SEAS southwest monsoon intensive study - Part 1: regional-scale phenomena,
Atmos Chem Phys, 16, 14041-14056, 10.5194/acp-16-14041-2016, 2016a.

1127

1128 Reid, J. S., Lagrosas, N. D., Jonsson, H. H., Reid, E. A., Atwood, S. A., Boyd, T. J., Ghate, V. P., 1129 Xian, P., Posselt, D. J., Simpas, J. B., Uy, S. N., Zaiger, K., Blake, D. R., Bucholtz, A., Campbell, 1130 J. R., Chew, B. N., Cliff, S. S., Holben, B. N., Holz, R. E., Hyer, E. J., Kreidenweis, S. M., 1131 Kuciauskas, A. P., Lolli, S., Oo, M., Perry, K. D., Salinas, S. V., Sessions, W. R., Smirnov, A., 1132 Walker, A. L., Wang, Q., Yu, L. Y., Zhang, J. L., and Zhao, Y. J.: Aerosol meteorology of 1133 Maritime Continent for the 2012 7SEAS southwest monsoon intensive study - Part 2: Philippine 1134 receptor observations of fine-scale aerosol behavior, Atmos Chem Phys, 16, 14057-14078, 1135 10.5194/acp-16-14057-2016, 2016b.

- 1136
- 1137 Rocha, L. D. S., and Correa, S. M.: Determination of size-segregated elements in diesel-biodiesel
- 1138 blend exhaust emissions, Environ Sci Pollut R, 25, 18121-18129, 10.1007/s11356-018-1980-8,
- 1139 2018.
- 1140

1141 Rogge, W. F., Mazurek, M. A., Hildemann, L. M., Cass, G. R., and Simoneit, B. R. T.: 1142 Quantification of Urban Organic Aerosols at a Molecular-Level - Identification, Abundance and 1143 Seasonal-Variation, Atmos Environ a-Gen, 27, 1309-1330, Doi 10.1016/0960-1686(93)90257-Y, 1144 1993. 1145 1146 Ro, C. U., Oh, K. Y., Kim, H., Kim, Y. P., Lee, C. B., Kim, K. H., Kang, C. H., Osan, J., De 1147 Hoog, J., Worobiec, A., and Van Grieken, R.: Single-particle analysis of aerosols at Cheju 1148 Island, Korea, using low-Z electron probe X-ray microanalysis: A direct proof of nitrate 1149 formation from sea salts, Environ Sci Technol, 35, 4487-4494, 10.1021/es0155231, 2001. 1150 1151 Rolph, G.D.: Real-time Environmental Applications and Display sYstem (READY) website 1152 (http://ready.Arl.NOAA.Gov), NOAA Air Resour. Lab., Silver Spring, Md., 2016. 1153 1154 Roth, B., and Okada, K.: On the modification of sea-salt particles in the coastal atmosphere, 1155 Atmos Environ, 32, 1555-1569, Doi 10.1016/S1352-2310(97)00378-6, 1998. 1156 1157 Saltzman, E. S., Savoie, D. L., Zika, R. G., and Prospero, J. M.: Methane Sulfonic-Acid in the 1158 Marine Atmosphere, J Geophys Res-Oceans, 88, 897-902, DOI 10.1029/JC088iC15p10897, 1983. 1159 1160 Saltzman, E. S., Savoie, D. L., Prospero, J. M., and Zika, R. G.: Methanesulfonic-Acid and Non-1161 Sea-Salt Sulfate in Pacific Air - Regional and Seasonal-Variations, J Atmos Chem, 4, 227-240, 1162 Doi 10.1007/Bf00052002, 1986. 1163 1164 Schade, G. W., and Crutzen, P. J.: Emission of Aliphatic-Amines from Animal Husbandry and 1165 Their Reactions - Potential Source of N₂O and HCN, J Atmos Chem, 22, 319-346, Doi 1166 10.1007/Bf00696641, 1995. 1167 1168 Schlosser, J. S., Braun, R. A., Bradley, T., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M., Aghdam, M. A., Mardi, A. H., Xian, P., and Sorooshian, A.: Analysis of Aerosol Composition 1169 1170 Data for Western United States Wildfires Between 2005-2015: Dust Emissions, Chloride 1171 Depletion, and Most Enhanced Aerosol Constituents, J. Geophys. Res., 122, 1172 doi:10.1002/2017JD026547, 2017. 1173 1174 Schwarz, J. P., Gao, R. S., Spackman, J. R., Watts, L. A., Thomson, D. S., Fahey, D. W., Ryerson, 1175 T. B., Peischl, J., Holloway, J. S., Trainer, M., Frost, G. J., Baynard, T., Lack, D. A., de Gouw, J.

- A., Warneke, C., and Del Negro, L. A.: Measurement of the mixing state, mass, and optical size
 of individual black carbon particles in urban and biomass burning emissions, Geophysical
 Research Letters, 35, 10.1029/2008gl033968, 2008.
- 1179
- Seinfeld, J. H., and Pandis, S. N.: Atmospheric chemistry and physics (3rd ed.). New York: Wiley-Interscience, 2016.
- 1182
- 1183 Shafer, M. M., Toner, B. M., Oyerdier, J. T., Schauer, J. J., Fakra, S. C., Hu, S. H., Herner, J. D.,
- and Ayala, A.: Chemical Speciation of Vanadium in Particulate Matter Emitted from Diesel
 Vehicles and Urban Atmospheric Aerosols, Environ Sci Technol, 46, 189-195,
- 1186 10.1021/es200463c, 2012.

- 1187
- 1188 Shingler, T., Sorooshian, A., Ortega, A., Crosbie, E., Wonaschutz, A., Perring, A. E.,
- 1189 Beyersdorf, A., Ziemba, L., Jimenez, J. L., Campuzano-Jost, P., Mikoviny, T., Wisthaler, A., and
- 1190 Russell, L. M.: Ambient observations of hygroscopic growth factor and f(RH) below 1: Case
- studies from surface and airborne measurements, J Geophys Res-Atmos, 121, 13661-13677,
- 1192 10.1002/2016jd025471, 2016.
- 1193
- 1194 Shiraiwa, M., Kondo, Y., Moteki, N., Takegawa, N., Sahu, L. K., Takami, A., Hatakeyama, S.,
- 1195 Yonemura, S., and Blake, D. R.: Radiative impact of mixing state of black carbon aerosol in Asian 1196 outflow, Journal of Geophysical Research: Atmospheres, 113, 10.1029/2008jd010546, 2008.
- 1197
- 1198 Simpas, J., Lorenzo, G., and Cruz, M. T.: Monitoring Particulate Matter Levels and Composition 1199 for Source Apportionment Study in Metro Manila, Philippines, in: Improving Air Quality in Asian
- 1200 Developing Countries: Compilation of Research Findings, Kim Oanh, N. T. (Ed.), NARENCA,
- 1201 Vietnam Publishing House of Natural Resources, Environment and Cartography, Vietnam, 239-
- 1202 261, 2014. 1203
- Singh, M., Jaques, P. A., and Sioutas, C.: Size distribution and diurnal characteristics of particlebound metals in source and receptor sites of the Los Angeles Basin, Atmos Environ, 36, 16751689, Pii S1352-2310(02)00166-8, Doi 10.1016/S1352-2310(02)00166-8, 2002.
- 1207
- Song, F., and Gao, Y.: Size distributions of trace elements associated with ambient particular
 matter in the affinity of a major highway in the New Jersey-New York metropolitan area, Atmos
 Environ, 45, 6714-6723, 10.1016/j.atmosenv.2011.08.031, 2011.
- 1211
- 1212 Sorooshian, A., Ng, N. L., Chan, A. W. H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.:
- 1213 Particulate organic acids and overall water-soluble aerosol composition measurements from the
- 1214 2006 Gulf of Mexico Atmospheric Composition and Climate Study (GoMACCS), J Geophys
- 1215 Res-Atmos, 112, 10.1029/2007jd008537, 2007.
- 1216
- Sorooshian, A., Murphy, S. N., Hersey, S., Gates, H., Padro, L. T., Nenes, A., Brechtel, F. J.,
 Jonsson, H., Flagan, R. C., and Seinfeld, J. H.: Comprehensive airborne characterization of aerosol
 from a major bovine source, Atmos Chem Phys, 8, 5489-5520, DOI 10.5194/acp-8-5489-2008,
- 1220 1221

2008.

- Sorooshian, A., Padro, L. T., Nenes, A., Feingold, G., McComiskey, A., Hersey, S. P., Gates, H.,
- 1223 Jonsson, H. H., Miller, S. D., Stephens, G. L., Flagan, R. C., and Seinfeld, J. H.: On the link
- 1224 between ocean biota emissions, aerosol, and maritime clouds: Airborne, ground, and satellite
- 1225 measurements off the coast of California, Global Biogeochem Cy, 23,
- 1226 10.1029/2009gb003464, 2009.
- 1227
- Sorooshian, A., Crosbie, E., Maudlin, L. C., Youn, J. S., Wang, Z., Shingler, T., Ortega, A. M.,
 Hersey, S., and Woods, R. K.: Surface and airborne measurements of organosulfur and
- 1230 methanesulfonate over the western United States and coastal areas, J Geophys Res-Atmos, 120,
- 1231 8535-8548, 10.1002/2015jd023822, 2015.
- 1232

1233 Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.: NOAA's 1234 Hysplit Atmospheric Transport and Dispersion Modeling System, B Am Meteorol Soc, 96, 2059-1235 2077, 10.1175/Bams-D-14-00110.1, 2015.

- 1236 1237 Tai, A. P. K., Mickley, L. J., and Jacob, D. J.: Correlations between fine particulate matter (PM2.5) 1238 and meteorological variables in the United States: Implications for the sensitivity of PM2.5 to 1239 climate change, Atmos Environ, 44, 3976-3984, 10.1016/j.atmosenv.2010.06.060, 2010.
- 1240 1241 Tsakalou, C., Papamarkou, S., Tsakiridis, P. E., Bartzas, G., and Tsakalakis, K.: Characterization 1242 and leachability evaluation of medical wastes incineration fly and bottom ashes and their 1243 vitrification outgrowths, J Environ Chem Eng, 6, 367-376, 10.1016/j.jece.2017.12.012, 2018.
- 1244
- 1245 U.S. Environmental Protection Agency: Monitoring PM2.5 in Ambient Air
- 1246 Using Designated Reference or Class I Equivalent Methods. Report No. EPA-454/B-16-001. US 1247 Environmental Protection Agency, Research Triangle Park, NC., 2016.
- 1248 1249 VandenBoer, T. C., Petroff, A., Markovic, M. Z., and Murphy, J. G.: Size distribution of alkyl 1250 amines in continental particulate matter and their online detection in the gas and particle phase, 1251 Atmos Chem Phys, 11, 4319-4332, 10.5194/acp-11-4319-2011, 2011.
- 1253 Villafuerte, M. Q., Matsumoto, J., Akasaka, I., Takahashi, H. G., Kubota, H., and Cinco, T. A.: 1254 Long-term trends and variability of rainfall extremes in the Philippines, Atmos Res, 137, 1-13, 1255 10.1016/j.atmosres.2013.09.021, 2014.
- 1256

1252

- 1257 Vossler, T., Cernikovsky, L., Novak, J., and Williams, R.: Source apportionment with 1258 uncertainty estimates of fine particulate matter in Ostrava, Czech Republic using Positive Matrix 1259 Factorization, Atmos Pollut Res, 7, 503-512, 10.1016/j.apr.2015.12.004, 2016.
- 1260 1261 Wang, J., Ge, C., Yang, Z. F., Hyer, E. J., Reid, J. S., Chew, B. N., Mahmud, M., Zhang, Y. X., 1262 and Zhang, M. G.: Mesoscale modeling of smoke transport over the Southeast Asian Maritime Continent: Interplay of sea breeze, trade wind, typhoon, and topography, Atmos Res, 122, 486-1263 1264 503, 10.1016/j.atmosres.2012.05.009, 2013.
- 1266 Wang, Y. O., Zhang, X. Y., and Draxler, R. R.: TrajStat: GIS-based software that uses various 1267 trajectory statistical analysis methods to identify potential sources from long-term air pollution 1268 measurement data, Environ Modell Softw, 24, 938-939, 10.1016/j.envsoft.2009.01.004, 2009. 1269
- 1270 Wasson, S. J., Linak, W. P., Gullett, B. K., King, C. J., Touati, A., Huggins, F. E., Chen, Y. Z.,
- 1271 Shah, N., and Huffman, G. P.: Emissions of chromium, copper, arsenic, and PCDDs/Fs from open
- 1272 burning of CCA-treated wood, Environ Sci Technol, 39, 8865-8876, 10.1021/es050891g, 2005.
- 1273

- 1274 Watson, J. G.: Protocol for Applying and Validating the CMB Model for PM_{2.5} and VOC. Report No. EPA-451/R-04-001. US Environmental Protection Agency, Research Triangle Park, NC., 1275 2004.
- 1276
- 1277

- Watts, S. F., Watson, A., and Brimblecombe, P.: Measurements of the Aerosol Concentrations of
 Methanesulfonic Acid, Dimethyl-Sulfoxide and Dimethyl Sulfone in the Marine Atmosphere of
 the British-Isles, Atmos Environ, 21, 2667-2672, Doi 10.1016/0004-6981(87)90198-3, 1987.
- Wonaschuetz, A., Sorooshian, A., Ervens, B., Chuang, P. Y., Feingold, G., Murphy, S. M., de
 Gouw, J., Warneke, C., and Jonsson, H. H.: Aerosol and gas re-distribution by shallow cumulus
 clouds: An investigation using airborne measurements, J Geophys Res-Atmos, 117,
 10 1029/2012id018089 2012
- 1285 10.1029/2012jd018089, 2012. 1286
- Wu, D., Zhang, F., Lou, W. H., Li, D., and Chen, J. M.: Chemical characterization and toxicity
 assessment of fine particulate matters emitted from the combustion of petrol and diesel fuels, Sci
 Total Environ, 605, 172-179, 10.1016/j.scitotenv.2017.06.058, 2017.
- Xian, P., Reid, J. S., Atwood, S. A., Johnson, R. S., Hyer, E. J., Westphal, D. L., and Sessions, W.:
 Smoke aerosol transport patterns over the Maritime Continent, Atmos Res, 122, 469-485,
 10.1016/j.atmosres.2012.05.006, 2013.
- 1294
 1295 Xu, G. J., and Gao, Y.: Characterization of marine aerosols and precipitation through shipboard
 1296 observations on the transect between 31 degrees N-32 degrees S in the West Pacific, Atmos Pollut
 1297 Res, 6, 154-161, 10.5094/Apr.2015.018, 2015.
 - Yao, X. H., Fang, M., and Chan, C. K.: The size dependence of chloride depletion in fine and coarse sea-salt particles, Atmos Environ, 37, 743-751, 10.1016/S1352-2310(02)00955-X, 2003.
 - Youn, J. S., Wang, Z., Wonaschutz, A., Arellano, A., Betterton, E. A., and Sorooshian, A.:
 Evidence of aqueous secondary organic aerosol formation from biogenic emissions in the North
 American Sonoran Desert, Geophys Res Lett, 40, 3468-3472, 10.1002/grl.50644, 2013.
 - Youn, J. S., Crosbie, E., Maudlin, L. C., Wang, Z., and Sorooshian, A.: Dimethylamine as a major
 alkyl amine species in particles and cloud water: Observations in semi-arid and coastal regions,
 Atmos Environ, 122, 250-258, 10.1016/j.atmosenv.2015.09.061, 2015.
 - 1309

- Youn, J. S., Csavina, J., Rine, K. P., Shingler, T., Taylor, M. P., Saez, A. E., Betterton, E. A., and
 Sorooshian, A.: Hygroscopic Properties and Respiratory System Deposition Behavior of
 Particulate Matter Emitted By Mining and Smelting Operations, Environ Sci Technol, 50, 1170611713, 10.1021/acs.est.6b03621, 2016.
- 1314

- 1315 **Table 1.** Summary of average operating parameters, meteorological conditions, and total resolved
- 1316 water-soluble mass concentration for each MOUDI sample set collected at Manila Observatory
- 1317 (MO) during the 2018 Southwest Monsoon period. On two occasions, simultaneous MOUDI sets
- 1318 were collected for one set to undergo gravimetric analysis (MO3 and MO13) to compare with mass
- 1319 resolved from chemical speciation of the water-soluble fraction (MO4 and MO14). One additional
- MOUDI set devoted to microscopy analysis was collected using aluminum substrates for one houron August 1 at 30 LPM.
- 1322

Sample	Dates	Durat	Flow	Wind	Wind	Т	Rain	Water-
set name		ion	rate	speed	direction	(°C)	(mm)	soluble mass
		(hrs)	(LPM)	(m/s)	(°)			$(\mu g m^{-3})$
MO1	Jul 19-20	24	30	3.3	90.1	24.9	47	4.6
MO2	Jul 23-25	54	30	1.3	95.8	26.7	7.8	6.5
MO3/4	Jul 25-30	119	28/30	1.2	111.8	26.7	49.6	5.2
MO5	Jul 30-Aug 1	42	29	2.6	98.1	27.5	52.8	9.2
MO6	Aug 6-8	48	27	0.9	127.5	26.1	30.4	5.1
MO7	Aug 14-16	48	28	3.0	107.8	27.8	2.8	13.7
MO8	Aug 22-24	48	29	3.5	108.7	28.1	1	12.8
MO9	Sep 1-3	48	27	0.7	98.6	26.6	51.6	6.2
MO10	Sep 10–12	48	29	1.0	94.7	26.2	78.4	6.4
MO11	Sep 18–20	48	27	0.5	290.2	27.8	0	2.7
MO12	Sep 26-28	48	27	1.2	96.3	27.8	6.8	13.5
MO13/14	Oct 6-8	48	28/26	0.6	108.2	27.8	0.8	16.6

1325	Table 2. Charge balance slopes (cations on y-axis; anions on x-axis) for the MOUDI sets shown
1326	including the averages of all sets (All) for three size ranges: submicrometer stages spanning 0.056
1327	-1.0μ m; supermicrometer stages (> 1.0 μ m); and all stages (> 0.056 μ m). The species used in
1328	the charge balance analysis include those speciated with the IC (listed in Section 2.3) plus K from
1329	ICP-QQQ analysis.

1220	
1330	

Sample set	0.056 – 1.0 μm	> 1 µm	> 0.056 µm
MO1	0.87	1.37	0.89
MO2	1.46	1.26	1.41
MO4	1.25	1.17	1.21
MO5	1.35	1.43	1.41
MO6	1.29	1.45	1.31
MO7	1.40	1.23	1.36
MO8	1.35	1.33	1.36
MO9	1.28	1.55	1.26
MO10	1.37	1.36	1.35
MO11	0.97	1.60	1.27
MO12	1.37	1.19	1.33
MO14	1.31	1.28	1.29
All	1.35	1.24	1.33

Table 3. Contributions (in weight percentage) of each PMF source factor to the total mass in different diameter ranges.

Diameter	Aged/	Sea Salt	Combustion	Vehicular/	Waste Processing
Range (µm)	Transported			Resuspended	
	-			Dust	
> 0.056	48.0%	22.5%	18.7%	5.6%	5.1%
0.056 - 1.0	68.9%	0.6%	23.9%	1.5%	5.1%
> 1.0	18.6%	53.5%	11.3%	11.3%	5.3%

- 1337 Table 4. Correlation matrix (r values) between water-soluble species based on total MOUDI-
- 1338 integrated mass concentrations (> $0.056 \mu m$). Blank cells represent statistically insignificant
- 1339 values. Results for the sub- and supermicrometer ranges are in Tables S2-S3. Panels A-E
- represent important species from each of the source profiles identified in Section 3.3: A =
- 1 β 41 Aged/Transported, B = Sea Salt, C = Combustion, D = Vehicular/Resuspended Dust, E = Waste
- 1342 Processing. DMA Dimethylamine, MSA Methanesulfonate, PH Phthalate, OX Oxalate,
- 1343 MA Maleate, SU Succinate, AD Adipate.
- 1344

A)															
OX	1.00														
SO ₄	0.74	1.00													
NH ₄	0.68	0.99	1.00												
Sn	0.71	0.87	0.85	1.00											
Rb	0.73	0.74	0.73	0.69	1.00										
K	0.76	0.71	0.69	0.69	0.97	1.00									
Cs	0.72	0.82	0.81	0.74	0.96	0.91	1.00		_						
V	0.36	0.64	0.63	0.48	0.53	0.51	0.57	1.00							
DMA		0.35		0.38	0.45	0.37	0.45		1.00						
MSA	0.71	0.89	0.89	0.79	0.90	0.85	0.92	0.51	0.47	1.00					
PH	0.68	0.67	0.68	0.73	0.82	0.76	0.80		0.38	0.88	1.00				
SU	0.63	0.56	0.59	0.44	0.87	0.81	0.82		0.68	0.78	0.84	1.00			
AD	0.40	0.66	0.70	0.62	0.70	0.70	0.77		0.84	0.74	0.75	0.90	1.00		
Se	0.75	0.75	0.73	0.66	0.80	0.78	0.79	0.32	0.34	0.78	0.80	0.88	0.88	1.00	
Tl	0.75	0.87	0.86	0.80	0.89	0.85	0.94	0.74	0.65	0.80	0.52	0.70		0.43	1.00
	OX	SO ₄	NH ₄	Sn	Rb	K	Cs	V	DMA	MSA	PH	SU	AD	Se	П

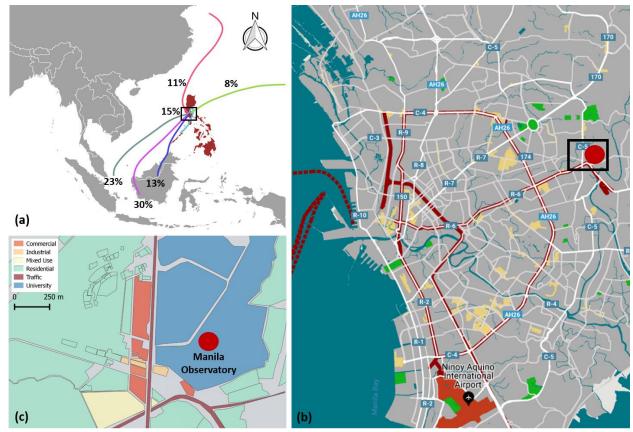
B)								
Cl	1.00							
NO ₃	0.76	1.00						
Ba	0.66	0.80	1.00					
Sr	0.78	0.87	0.91	1.00				
Ca	0.58	0.79	0.75	0.78	1.00			
Na	0.93	0.87	0.75	0.85	0.63	1.00		
Mg	0.91	0.87	0.77	0.87	0.66	0.99	1.00	
Hf					0.57			1.00
	Cl	NO ₃	Ba	Sr	Ca	Na	Mg	Hf

	As	Ni	Co	Р	Mo	Cr	Mal	Ag
Ag			0.85		0.64			1.00
MA			0.67		-0.42		1.00	
Cr	0.62	0.49		0.20		1.00		
Mo					1.00			
Р		0.33	0.34	1.00				
Со			1.00					
Ni	0.58	1.00						
As	1.00		_					
C)								

D)						
Zr	1.00					
Y	0.75	1.00				
Al	0.88	0.76	1.00			
Fe	0.33	0.61	0.25	1.00		
Ti	0.84	0.66	0.82	0.41	1.00	
Nb	0.70	0.50	0.59	0.59	0.70	1.00
	Zr	Y	Al	Fe	Ti	Nb

E)						
Cd	1.00					
Zn	0.60	1.00				
Cu	0.21	0.27	1.00			
Mn	0.28	0.61	0.22	1.00		_
Pb	0.78	0.58	0.38	0.27	1.00	
	Cd	Zn	Cu	Mn	Pb	

- 1345 1346
- 1347
- 1348



1349

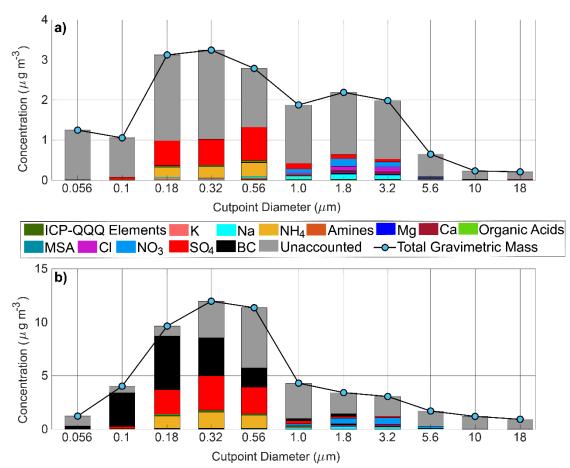
Figure 1. (a) Location of Metro Manila, Philippines relative to Southeast Asia. Also shown are 1351 5-day backward trajectory frequencies during the sampling duration based on HYSPLIT cluster 1352 analysis; note that 15% correspond to trajectories within the black square. (b) Close-up view of 1353 Metro Manila showing the location of the Manila Observatory sampling site with a black rectangle. 1354 The base map shows roads, commercial centers, and major transit lines in the city. (c) Land use 1355 classification in the vicinity of the sampling site. (Sources: GADM, Snazzy Maps, OpenStreetMap, 1356 NOAA HYSPLIT, & TrajSat)

1358

1359

1360 1361

1201



1363

1364 **Figure 2.** Mass size distributions of total PM (blue markers) and resolved chemical species

1365 (colored bars) for MOUDI sets (a) MO3/4 and (b) MO13/14. Note that set MO13 was the single

1366 MOUDI set where BC was quantified. ICP-QQQ = sum of water-soluble elements except K;

amines = sum of DMA, TMA, DEA; organic acids = sum of oxalate, succinate, adipate,

1368 pyruvate, phthalate, maleate.

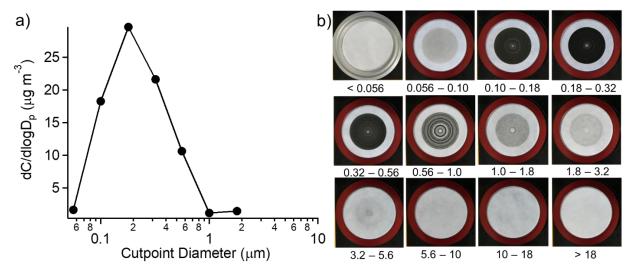
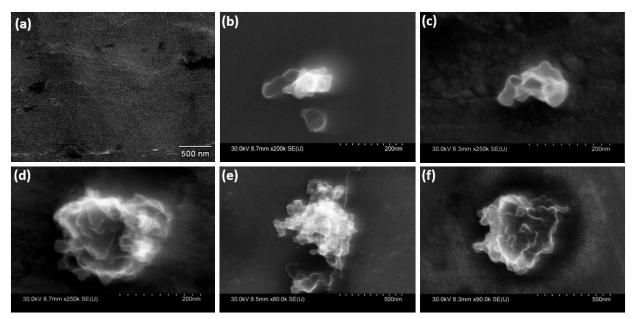


Figure 3. (a) Mass size distribution of BC retrieved from the MABI optical measurement at 870

1373 nm for set MO13. Missing values were below detection limits. (b) Photographs of each stage of 1374 set MO13 with numbers below each image representing the aerodynamic diameter ranges in units

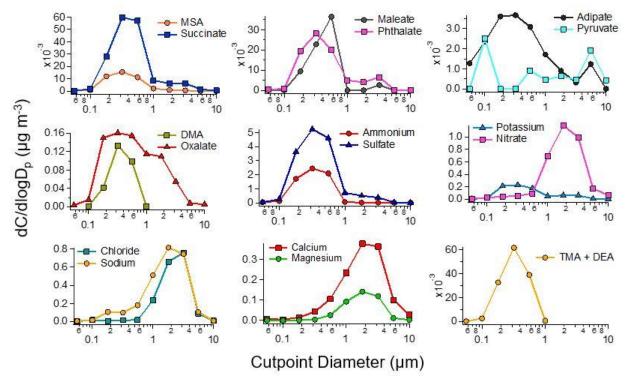
1375 of μm.



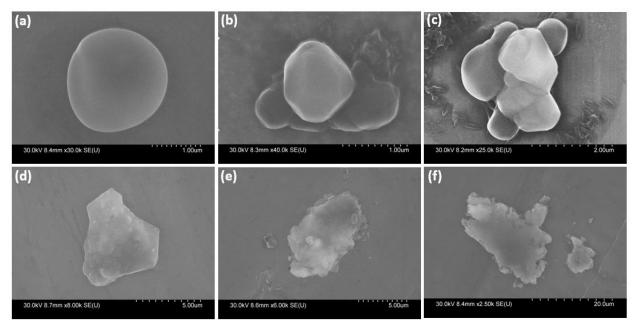


1378 Figure 4. SEM image of a (a) blank filter and (b-f) individual particles in different sub-

- micrometer aerodynamic diameter ranges sampled by the MOUDI: (b) $0.056-0.1 \mu m$, (c) 0.1-
- 0.18 μm, (d) 0.18–0.32 μm, (e) 0.32–0.56 μm, (f) 0.56–1.0 μm.



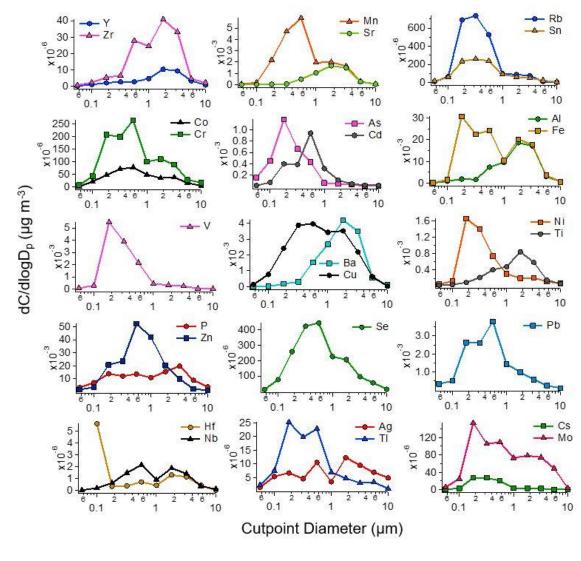
1382
1383 Figure 5. Average mass size distribution of water-soluble ions speciated via IC in addition to
1384 potassium from ICP-QQQ analysis.



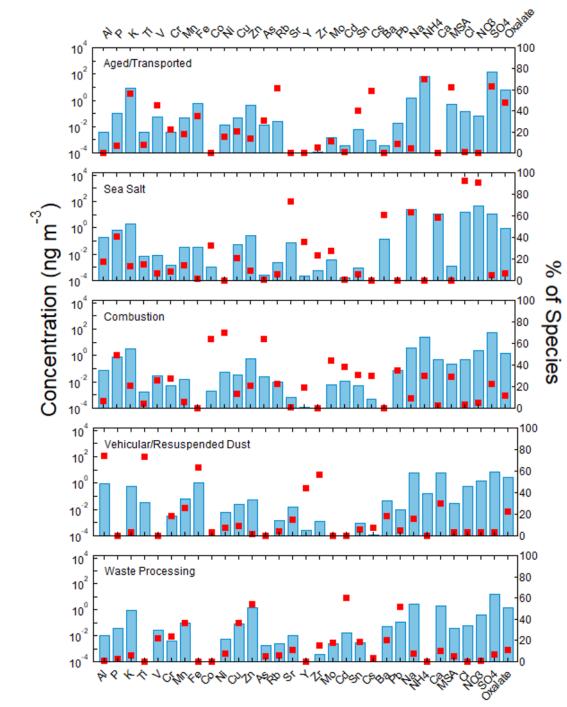


1388 Figure 6. Same as Figure 4, but for different supermicrometer aerodynamic diameter ranges sampled by the MOUDI: (a) 1.0–1.8 µm, (b) 1.8–3.2 µm; (c) 3.2–5.6 µm, (d) 5.6–10 µm, (e) 10-

- μ m, (f) > 18 μ m.



1392
1393
1394 Figure 7. Average mass size distribution of water-soluble elements speciated via ICP-QQQ.
1395



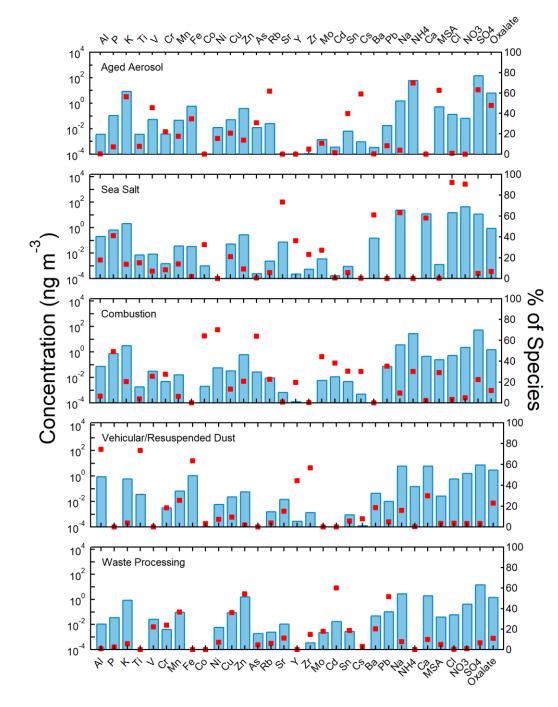
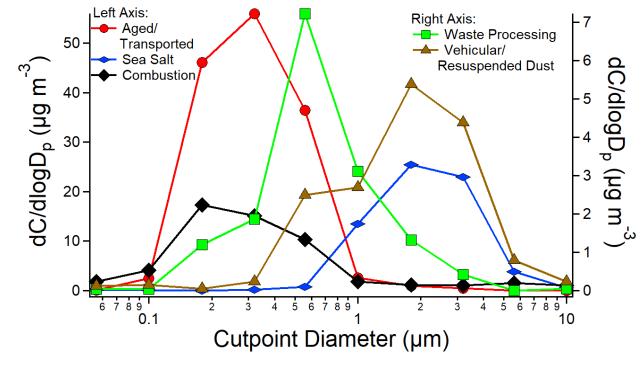


Figure 8. Overview of the PMF five factor solution with blue bars representing mass

- 1399 concentrations and red squares signifying the percentage of mass concentration contributed to1400 constituents by each source factor.
- 1401



1403 Figure 9. Reconstructed mass size distributions using PMF for the five major source profiles.