Size-resolved Composition and Morphology of Particulate

Matter During the Southwest Monsoon in Metro Manila,

Philippines

4

1

2

3

- 5 Melliza Templonuevo Cruz^{1,2}, Paola Angela Bañaga^{1,3}, Grace Betito³, Rachel A. Braun⁴, Connor
- 6 Stahl⁴, Mojtaba Azadi Aghdam⁴, Maria Obiminda Cambaliza^{1,3}, Hossein Dadashazar⁴, Miguel
- 7 Ricardo Hilario³, Genevieve Rose Lorenzo¹, Lin Ma⁴, Alexander B. MacDonald⁴, Preciosa
- 8 Corazon Pabroa⁵, John Robin Yee⁵, James Bernard Simpas^{1,3}, Armin Sorooshian^{4,6}

9

- 10 ¹Manila Observatory, Quezon City 1101, Philippines
- ²Institute of Environmental Science and Meteorology, University of the Philippines, Diliman, Quezon City 1101,
- 12 Philippines
- ³Department of Physics, School of Science and Engineering, Ateneo de Manila University, Quezon City 1108,
- 14 Philippines
- ⁴Department of Chemical and Environmental Engineering, University of Arizona, Tucson, AZ, USA
- ⁵Philippine Nuclear Research Institute, Commonwealth Avenue, Diliman, Quezon City 1101, Philippines
- 17 ⁶Department of Hydrology and Atmospheric Sciences, University of Arizona, Tucson, AZ, USA

18 19

Correspondence to: Melliza Templonuevo Cruz (liz@observatory.ph)

Abstract

21

22 This paper presents novel results from size-resolved particulate matter (PM) mass, composition, 23 and morphology measurements conducted during the 2018 Southwest Monsoon (SWM) season in 24 Metro Manila, Philippines. Micro-Orifice Uniform Deposit Impactors (MOUDIs) were used to 25 collect PM sample sets composed of size-resolved measurements at the following aerodynamic 26 cutpoint diameters (D_p): 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 µm. Each sample 27 set was analyzed for composition of the water-soluble fraction. Analysis for mass were done on 28 two sample sets while black carbon (BC) and morphology analysis were done on a single sample 29 set. The bulk of the PM mass was between 0.18-1.0 µm with a dominant mode between 0.32-30 0.56 µm. Similarly, most of the black carbon (BC) mass was found between 0.10–1.0 µm, peaking 31 between 0.18–0.32 µm. These peaks are located in the Greenfield Gap or the size range between 32 $0.10-1.0 \mu m$, where wet scavenging by rain is relatively inefficient. In the range of 0.10-0.1833 µm, BC constituted 78.1% of the measured mass. Comparable contributions of BC (26.9%) and 34 the water-soluble fraction (31.3%) to total PM were observed and most of the unresolved mass, 35 which in total amounted to 41.8%, was for diameters exceeding 0.32 µm. The water-soluble ions and elements exhibited an average combined concentration of 8.53 µg m⁻³, with SO₄²⁻, NH₄⁺, NO₃⁻ 36 37 , Na⁺, and Cl⁻ as the major contributors. Positive Matrix Factorization (PMF) was applied to 38 identify the possible aerosol sources and estimate their contribution to the water-soluble fraction 39 of collected PM. The factor with the highest contribution was attributed to "Aged" aerosol (48.0%) while "Sea Salt" (22.5%) and "Combustion" emissions (18.7%) had comparable contributions. 40 41 "Vehicular/Resuspended Dust" (5.6%) as well as "Waste Processing" emissions (5.1%) were also 42 identified. Microscopy analysis highlighted the ubiquity of non-spherical particles regardless of 43 size, which is significant when considering calculations of parameters such as single scattering 44 albedo, asymmetry parameter, and extinction efficiency. 45 The significant influence from Aged aerosol to Metro Manila during the SWM season indicates 46 that local sources in this megacity do not fully govern this coastal area's aerosol properties. That 47 the majority of the regional aerosol mass burden is accounted for by BC and other insoluble 48 components has important downstream effects on the aerosol hygroscopic properties, which 49 depend on composition. The results are relevant for understanding the impacts of monsoonal 50 features on size-resolved aerosol properties, notably aqueous processing and wet scavenging. 51 Finally, the results of this work provide contextual data for future sampling campaigns in Southeast

- Asia such as the airborne component of the Cloud, Aerosol, and Monsoon Processes Philippines
- 53 Experiment (CAMP²Ex) planned for the SWM season in 2019.

1. Introduction

Ambient atmospheric aerosol particles impact human health, visibility, climate, and the hydrological cycle. Major factors governing these behaviors, such as deposition fraction in the respiratory system and activation into cloud condensation nuclei (CCN), include size and chemical composition. Therefore, size-resolved measurements of ambient aerosol particles can lend additional insights to the behavior and implications of particulate matter (PM) in the atmosphere. One region of interest for characterization of aerosols is Southeast Asia due to increasing urbanization and the exposure of the population to a variety of aerosol sources, both natural and anthropogenic (Hopke et al., 2008). However, use of space-borne remote-sensing instrumentation presents a challenge for characterization of aerosol in this region, due to issues such as varying terrain and cloud cover (Reid et al., 2013).

The Philippines represents a country in Southeast Asia with a developing economy, rapid urbanization, old vehicular technology, and less stringent air quality regulations (e.g., Alas et al., 2017). It is also highly sensitive to the effects of climate change including prolonged dry periods and reductions in southwest monsoon (SWM) rainfall in recent decades (e.g., Cruz et al., 2013). Metro Manila is the country's capital and center of political and economic activities. Also referred to as the National Capital Region, Metro Manila is composed of 16 cities and a municipality that collectively occupy a land area of ~619 km². As of 2015, Metro Manila had a population of approximately 12.88 million (Philippine Statistics Authority, 2015). Of the cities comprising the Metro Manila area, the one that is the focus of this study, Quezon City, is the most populated (2.94 million people) with a population density of ~17,000 km² as of 2015 (Philippine Statistics Authority, 2015).

The rainfall pattern in Southeast Asia is governed by topographic effects and the prevailing surface winds brought by the monsoons. Mountain ranges in the Philippines are generally oriented north to south in the eastern and western coasts. As such, northeasterly winds during the East Asian winter monsoon that starts in November brings wetness (dryness) on the eastern (western) coasts of the country. In contrast, the rainy season starts in May when the Western North Pacific subtropical high moves northeast and the Asian summer monsoon enables the propagation of southwesterly wind through the Philippines (Villafuerte et al., 2014). Metro Manila, located on the western side of the Philippines, therefore experiences wet (May-October) and dry (November-April) seasons. The large seasonal shift in prevailing wind directions can cause changes in the

source locations of aerosol transported to the Philippines and the subsequent direction in which emissions from the Philippines are transported, such as to the northwest (e.g., Chuang et al., 2013) or southwest (e.g., Farren et al., 2019). However, one interesting feature of Metro Manila is the consistency of PM_{2.5}/PM₁₀ mass concentrations during both the dry (44/54 µg m⁻³) and wet seasons (43/55 µg m⁻³) (Kim Oanh et al., 2006), which stands in contrast to typical assumptions that increased wet scavenging during rainy seasons would lead to decreases in measured PM (e.g., Liao et al., 2006). While similar results are observed in Chennai, India, this behavior is different than other cities in Asia, including Bandung City (Indonesia), Bangkok (Thailand), Beijing (China), and Hanoi City (Vietnam), which exhibit reduced PM_{2.5} levels during the wet season as compared to the dry season (Kim Oanh et al., 2006). While the total PM levels may stay constant across the wet and dry seasons, seasonally-resolved analyses will provide additional insights into how the composition, morphology, and sources (transported vs. local emissions) change on a seasonal basis.

Metro Manila has been drawing growing interest for PM research owing to the significant levels of black carbon (BC). A large fraction of PM in Metro Manila can be attributed to BC (e.g., ~50% of PM_{2.5}; Kim Oanh et al., 2006), with previously measured average values of BC at the Manila Observatory (MO) reaching ~10 μg m⁻³ for PM_{2.5} (Simpas et al., 2014). The impacts of the high levels of BC present on human health have also received attention (Kecorius et al., 2019). Identified major sources of BC include vehicular, industrial, and cooking emissions (Bautista et al., 2014; Kecorius et al., 2017). Vehicular emissions, especially along roadways where personal cars and motorcycles, commercial trucks, and motorized public transportation, including powered tricycles and *jeepneys*, are plentiful. For instance, measurements of PM_{2.5} at the National Printing Office (NPO) located alongside the major thoroughfare Epifanio de los Santos Avenue (EDSA) were on average 72 μg m⁻³; this value is twice the average concentration at MO, an urban mixed site located approximately 5 km from NPO (Simpas et al., 2014). In addition to local emissions, long-range transport of pollution, such as biomass burning, can also impact the study region (e.g., Xian et al., 2013; Reid et al., 2016a/b). However, most past work referenced above has focused on either total PM_{2.5} or PM₁₀ composition, and therefore, detailed size-resolved composition information has been lacking in this region. Like other monsoonal regions (Crosbie et al., 2015; Qu et al., 2015), it is of interest for instance to know if products of aqueous processing (e.g., sulfate, organic acids) during the monsoonal period, promoted by the high humidity, become more

prominent in certain size ranges to ultimately enhance hygroscopicity, which is otherwise suppressed with higher BC influence.

A year-long sampling campaign (Cloud, Aerosol, and Monsoon Processes Philippines Experiment (CAMP²Ex) weatHEr and CompoSition Monitoring (CHECSM) study) was established in July 2018 to collect size-resolved aerosol measurements in Metro Manila. The aim of this study is to report size-resolved PM measurements taken over the course of the SWM (July-October) of 2018 in Quezon City, Metro Manila, Philippines as part of CHECSM. The results of this study are important for the following reasons: (i) they provide size-resolved analysis of BC in an area previously characterized as having one of the highest BC mass percentages in the whole world; (ii) they provide a basis for better understanding the unusual phenomenon of having similar PM levels during a wet and dry season; (iii) they provide contextual data for contrasting with both other coastal megacities and also other monsoonal regions; and (iv) they can lend insights into the characteristics of aerosol transported both into and out of Metro Manila and how important local sources are in Metro Manila relative to transported pollution.

Outcomes of this study include (i) the first size-resolved characterization of both aerosol composition and morphology in Metro Manila for the SWM, with implications in terms of PM effects on climate, visibility, the hydrological cycle, and public health owing to the dependence of these impacts on particle size; (ii) archival data that contributes to the timeline of aerosol research in Metro Manila, and more broadly Southeast Asia, where there is considerable concern over air pollution; and (iii) baseline data for aerosol composition to be used to inform and assist research to be conducted during future field campaigns in Southeast Asia including the same seasonal period (i.e., SWM) in 2019 as part of CAMP²Ex, which will involve both surface and airborne measurements.

2. Experimental Methods

2.1 Sample Site

Sampling was performed at MO in Quezon City, Philippines (14.64° N, 121.08° E). Two MOUDIs were placed inside an unoccupied room on the 3rd floor of the MO administration building (~87 m above sea level). The inlet, located just outside the window, consists of a 2 m long stainless steel tube and a reducer that is connected directly to the MOUDI inlet. Figure 1 visually shows the sampling location and potential surrounding aerosol sources. Past work focused

on PM_{2.5} suggested that the study location is impacted locally mostly by traffic, various forms of industrial activity, meat cooking from local eateries, and, based on the season, biomass burning (Cohen et al., 2009). This is consistent with another source apportionment study which reported that potential sources in six sites across Metro Manila include traffic, secondary particles, and biomass burning (Kim Oanh et al., 2013).

Meteorological data were collected using a Davis Vantage Pro 2 Plus weather station located on the roof (~90 m above sea level, ~15 m above ground level) above where the MOUDIs were located. Except for precipitation, which is reported here as accumulated rainfall, reported values for each meteorological parameter represent averages for the sampling duration of each aerosol measurement. The mean temperature during the periods of MOUDI sample collection ranged from 24.9 to 28.1° C, with accumulated rainfall ranging widely from no rain to up to 78.4 mm. To identify sources impacting PM via long-range transport to the Metro Manila region, Figure 1a summarizes the five-day back-trajectories for air masses arriving at MO on the days when samples were being collected, calculated using the NOAA Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Stein et al., 2015; Rolph, 2016). Trajectory calculations were started at 00, 06, 12, and 18 hours in MO at the height of the MOUDI inlet (~ 12 m above ground level) using meteorological files from the NCEP/NCAR Reanalysis dataset. Trajectory cluster analysis was conducted using TrajStat (Wang et al., 2009). The back-trajectories in Figure 1a show that indeed 66% of the wind came from the southwest during the sampling periods.

2.2 MOUDI Sample Sets

PM was collected on Teflon substrates (PTFE membrane, 2 μm pore, 46.2 mm, Whatman) in Micro-Orifice Uniform Deposit Impactors (MOUDI, MSP Corporation, Marple et al., 2014). Size-resolved measurements were taken at the following aerodynamic cutpoint diameters (D_p): 18, 10, 5.6, 3.2, 1.8, 1.0, 0.56, 0.32, 0.18, 0.10, 0.056 μm. Fourteen sample sets were collected during the SWM season (July-October 2018), with details about the operational and meteorological conditions during each sample set shown in Table 1. To determine the optimum sampling time that will collect enough sample for subsequent analyses, collection time for the first four samples ranged from 24 to 119 hours. Subsequent sampling were then fixed to 48 hours with one sample set collected every week. The sampling collection was designed to include samples from each day of the week so the collection cycled between Monday – Wednesday, Tuesday – Thursday,

Wednesday – Friday, and Saturday – Monday, starting at 1400 (local time) for the weekday samples and 0500 for the weekend samples. The Teflon substrates were pretreated by washing with deionized water and air drying in a covered box. Substrates were placed and retrieved from the cascade impactor inside the laboratory in an adjacent building and transported to and from the sampling site using an impactor holder (Csavina et al., 2011). Samples are immediately placed in the freezer upon retrieval.

On two occasions, two pairs of MOUDI sets (Sets MO3/MO4 and MO13/MO14) were collected simultaneously such that one set in each pair could undergo different types of analyses. Sets 3 and 13 underwent gravimetric analysis using a Sartorius ME5-F microbalance. Substrates were conditioned for at least 24 h at a mean temperature of 20-23 °C and a mean relative humidity of 30-40% before pre- and post-weighing (U.S. Environmental Protection Agency, 2016). MOUDI set 13 was additionally examined with a Multi-wavelength Absorption Black Carbon Instrument (MABI; Australian Nuclear Science and Technology Organisation). This optically-based instrument quantifies absorption and mass concentrations at seven wavelengths between 405 and 1050 nm; however, results are reported only for 870 nm to be consistent with other studies, as BC is the predominant absorber at that wavelength (e.g., Ramachandran and Rajesh, 2007; Ran et al., 2016). One additional sample set for microscopy analysis was collected for one hour on August 1 using aluminum substrates.

2.3 Chemical Composition Analysis

Twelve sample sets, composed of 11 samples each, were analyzed for water-soluble ions and elements (Table 2). In order to preserve samples for additional analysis, each Teflon substrate was cut in half. A half of each substrate was extracted in 8 mL of Milli-Q water (18.2 M Ω -cm) through sonication for 30 min in a sealed polypropylene vial. A blank substrate was processed in the same method to serve as a background control sample. Subsequent chemical analysis of the water-soluble components in the aqueous extracts were performed using ion chromatography (IC; Thermo Scientific Dionex ICS - 2100 system) for the following species: cations = Na⁺, NH₄⁺, Mg²⁺, Ca²⁺, dimethylamine (DMA), trimethylamine (TMA), diethylamine (DEA); anions =, methanesulfonate (MSA), pyruvate, adipate, succinate, maleate, oxalate, phthalate, Cl⁻, NO₃⁻, SO₄²⁻. Owing to co-elution of TMA and DEA in the IC system, a cumulative sum of the two is reported here, which represents an underestimate of their total mass concentration owing to overlap

in parts of their peaks. Limits of detection (LOD) were calculated for each species based on their respective calibration curve (Table S1), with LOD being three times the standard deviation of the residuals (predicted signal minus measured signal) divided by the slope of the calibration curve (Miller and Miller, 2018).

The aqueous extracts were simultaneously characterized for elemental composition using triple quadrupole inductively coupled plasma mass spectrometry (ICP-QQQ; Agilent 8800 Series) for the following species: K, Al, Fe, Mn, Ti, Ba, Zn, Cu, V, Ni, P, Cr, Co, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Ag, Cd, Sn, Cs, Hf, Tl, Pb. Limits of detection of the examined elements were calculated automatically by the ICP-QQQ instrument and were in the ppt range (Table S1). The sample concentrations represent an average of three separate measurements with a standard deviation of 3% or less.

Note that some species were detected by both IC and ICP-QQQ (i.e., Na⁺, K⁺, Mg²⁺, Ca²⁺), and that the IC concentrations are used here for all repeated species with the exception of K⁺ owing to better data quality from ICP-QQQ. All IC and ICP-QQQ species concentrations for samples have been corrected by subtracting concentrations from background control samples. For more examples of the application of these methods used for substrate collection and IC/ICP analysis, the reader is referred to other recent work (Braun et al., 2017; Ma et al., 2019; Schlosser et al., 2017).

2.4 Microscopy Analysis

As already noted, one MOUDI set on August 1 was devoted to microscopy analysis. Morphology and additional elemental composition analysis was carried out on this set of aluminum substrates using scanning electron microscopy equipped with energy dispersive X-ray spectroscopy (SEM-EDX) in the Kuiper Imaging cores at the University of Arizona. Secondary electron (SE) imaging and EDX elemental analysis were performed using a Hitachi S-4800 high resolution SEM coupled to a Noran system Six X-ray Microanalysis System by Thermo Fisher Scientific. EDX analysis on individual particles was performed with 30 kV accelerating voltage to obtain weight percentages of individual elements. SEM-EDX results showed that the background control aluminum substrate was dominated by Al (88.27%), with minor contributions from Ag (5.34%), C (4.87%), O (0.79%), Fe (0.67%), and Co (0.05%). Such contributions were manually subtracted from spectra of individual particles on sample substrates, with the remaining elements

scaled up to hundred percent. Image processing was conducted with Image J software to measure particle dimensions and adjust the contrast and brightness of images to provide better visualization.

2.5 Computational Analysis

This study reports basic descriptive statistics for chemical concentrations and correlations between different variables. Statistical significance hereafter corresponds to 95% significance based on a two-tailed Student's t-test. To complement correlative analysis for identifying sources of species, positive matrix factorization (PMF) modeling was carried out using the United States Environmental Protection Agency's (US EPA) PMF version 5. A total of 132 samples from the 12 sets analyzed for water-soluble ions and elements were used in the PMF analysis. Species concentrations were examined before being inputted to PMF. Species considered as "strong" based on high signal-to-noise ratios (S/N > 1) and those with at least 50% of the concentrations above the LOD were used in the PMF modeling (Norris et al., 2014). This resulted in a 132 (samples) × 30 (species) data matrix that was inputted to PMF. Data points with concentrations exceeding the LOD had uncertainty quantified as

$$\sigma_{ij} = 0.05 \cdot X_{ij} + LOD_{ij}, \tag{Equation 1}$$

where σ_{ij} , X_{ij} , and LOD_{ij} are the uncertainty, concentration, and LOD, respectively, of the j^{th} species in the i^{th} sample (Reff et al., 2007). When concentration data were not available for a particular stage of a MOUDI set for a species, the geometric mean of the concentrations for that MOUDI stage and species was applied with uncertainty counted as four times the geometric mean value (Polissar et al., 1998; Huang et al., 1999). A 25% extra modeling uncertainty was applied to account for other sources of errors, such as changes in the source profiles and chemical transformations (Dumanoglu et al., 2014; Norris et al., 2014). The model was run 20 times with a randomly chosen starting point for each run.

3. Results

3.1 Total Mass Concentrations and Charge Balance

The average total mass concentration (\pm standard deviation) of water-soluble species across all MOUDI stages (Table 1) during the study period was $8.53 \pm 4.48 \,\mu g \, m^{-3}$ (range = $2.7-16.6 \,\mu g$

m⁻³). The species contributing the most to the total water-soluble mass concentration during the SWM included $SO_4^{2^-}$ (44% ± 6%), NH_4^+ (18% ± 5%), NO_3^- (10 ± 3%), Na^+ (8 ± 3%), and Cl^- (6% ± 3%). The meteorological parameters from Table 1 best correlated to total water-soluble mass concentrations were temperature (r = 0.64) and rainfall (r = -0.49). The highest total mass concentration (set MO13/14 = 16.6 µg m⁻³) occurred during the period with one of the highest average temperatures (27.8 °C) and second least total rainfall (0.8 mm). Other sampling periods with high mass concentrations (sets MO7, MO8, and MO12) coincided with the highest temperature and lowest rainfall observations. High temperatures, and thus more incident solar radiation, presumably enhanced production of secondary aerosol species via photochemical reactions as has also been observed in other regions for their respective monsoon season (Youn et al., 2013).

Low rainfall is thought to have been coincident with reduced wet scavenging of aerosol at the study site as has been demonstrated for other regions such as North America (Tai et al., 2010) and megacities such as Tehran (Crosbie et al., 2014). However, set MO11 exhibited a very low concentration even with high temperature and lack of rainfall, which may be due to changes in the source and transport of aerosol since this sample set coincided with a significant change in average wind direction (290.2° for MO11 vs. 90.1° – 127.5° for all other MOUDI sets). While the reported rainfall measurements were taken at MO, inhomogeneous rainfall patterns in the regions surrounding the Philippines could also contribute to the wet scavenging of PM, thereby lowering the quantity of transported particles reaching the sample site. Future work will address the influence of spatiotemporal patterns of precipitation on PM loadings in the Philippines as a point measurement at an aerosol observing site may be misleading.

On two occasions, two simultaneous MOUDI sets (Sets MO3/MO4 and MO13/MO14) were collected for the potential to compare different properties that require separate substrates. The total mass concentrations based on gravimetric analysis of sets MO3 and MO13 were 18.6 μ g m⁻³ and 53.0 μ g m⁻³, respectively (Figure 2). Both sets exhibited a dominant concentration mode between 0.32–0.56 μ m and the MO3 set was different in that it exhibited bimodal behavior with a second peak between 1.8–3.2 μ m. The sum of speciated water-soluble species accounted for only 27.8% and 31.3% of the total gravimetric mass of sets MO3 and MO13, respectively, indicative of significant amounts of water-insoluble species undetected by IC and ICP-QQQ. When adding the total mass of BC (14.3 μ g m⁻³) to the other resolved species from set MO13 (the one time BC

was measured), there was still 22.1 μg m⁻³ of unresolved mass (41.8% of total PM). Most of the unaccounted mass was for $D_p > 0.32~\mu m$.

The observation of BC accounting for 26.9% of total PM (14.3 μg m⁻³) is consistent with past work highlighting the significant fraction of BC in the ambient aerosol of Manila (Kim Oanh et al., 2006; Bautista et al., 2014; Simpas et al., 2014; Kecorius et al., 2017). However, this fraction of BC is very high compared to measurements during the monsoon season in other parts of the world. The mass fraction of BC in total suspended PM (TSPM) was 1.6%/2.2% for the monsoon season in 2013/2014 in Kadapa in southern India, even though the TSPM measured was comparable to that in Manila (64.9 and 49.9 μg m⁻³, for 2013 and 2014 in Kadapa, respectively) (Begam et al., 2017). Multiple studies during the monsoon season in a coastal region in southwest India showed BC mass contributions of 1.9 – 5% (Aswini et al., 2019 and references therein). Airborne measurements around North America and in Asian outflow revealed that BC accounted for only ~1-2% of PM_{1.0} (Shingler et al., 2016) and ~5-15% of accumulation mode aerosol mass (Clarke et al., 2004), respectively.

To investigate further about the missing species, a charge balance was carried out for all MOUDI sets (Table 2) to compare the sum of charges for cations versus anions based on IC analysis including K from ICP-QQQ analysis (species listed in Section 2.3). The slope of the charge balances (cations on y-axis) for the cumulative dataset was 1.33 and ranged from 0.89 to 1.41 for the 12 individual MOUDI sets that had IC and ICP-QQQ analysis conducted on them. Eleven of the 12 sets exhibited slopes above unity indicating that there was a deficit in the amount of anions detected, which presumably included species such as carbonate and various organics. To further determine if there were especially large anion or cation deficits in specific size ranges, slopes are also reported for 0.056–1 μm and > 1 μm . There were no obvious differences other than two MOUDI sets exhibited slopes below 1.0 for the smaller diameter range (0.056–1 μm) while all slopes exceeded unity for > 1 μm .

3.2 Mass Size Distributions and Morphology

3.2.1 Black Carbon

The size-resolved nature of BC has not been characterized in Manila and MOUDI set MO13 offered a view into its mass size distribution (Figure 3a). There was a pronounced peak between 0.18–0.32 µm (5.0 µg m⁻³), which is evident visually in the substrate's color when

compared to all other stages of that MOUDI set (Figure 3b). This observed peak in the mass size distribution of BC is similar to previous studies of the outflow of East Asian countries (Shiraiwa et al., 2008), biomass burning and urban emissions in Texas (Schwarz et al., 2008), measurements in the Finnish Arctic (Raatikainen et al., 2015), and airborne measurements over Europe (Reddington et al., 2013). In contrast, measurements in Uji, Japan showed a bimodal size distribution for the mass concentration of BC in the submicrometer range (Hitzenberger and Tohno, 2001).

In the present study, there were significant amounts of BC extending to as low as the 0.056-0.1 μm MOUDI stage (0.28 μg m⁻³) and extending up in the supermicrometer range with up to 0.25 μg m⁻³ measured between 1.8–3.2 μm . Remarkably, BC accounted for approximately 78.1% (51.8%) by mass of the total PM in the range of 0.10 – 0.18 μm (0.18 – 0.32 μm). For comparison, the mass percent contribution of BC measured in the megacity of Nanjing, China was 3.3% (1.6%) at 0.12 (0.08) μm (Ma et al., 2017). Based on visual inspection of color on all MOUDI sets, MO13 appears to be representative of the other sets based on the relative intensity of the color black on substrates with different cutpoint diameters (Figure 3b); the 0.18–0.32 μm substrate always was the most black, with varying degrees of blackness extending consistently into the supermicrometer stages.

Microscopy analysis revealed evidence of non-spherical particles in each MOUDI stage below 1 μ m (Figure 4), which is significant as the common assumption theoretically is that submicrometer particles are typically spherical (e.g., Mielonen et al., 2011). Errors in this assumption impact numerical modeling results and interpretation of remote sensing data for aerosols (e.g., Kahnert et al., 2005), owing to incorrect calculations of parameters such as single scattering albedo, asymmetry parameter, and extinction efficiency (e.g., Mishra et al., 2015). Some studies have noted that submicrometer particles could be composed of an agglomeration of small spherical particles originally formed through gas-to-particle conversion processes (Almeida et al., 2019), which could potentially explain the appearance for some of the observed particles in Figure 4. Since only single particles were examined that may not be fully representative of all particles on a particular MOUDI substrate, it is noteworthy that all five particles shown between 0.056 – 1 μ m were irregularly shaped with signs of both multi-layering and constituents adhered to one another.

The images show that a potentially important source of BC in the area could be soot aggregates, which are formed by a vaporization-condensation process during combustion often associated with vehicular exhaust (e.g., Chen et al., 2006; Chithra and Nagendra, 2013; Wu et al., 2017). Kecorius et al. (2017) projected that 94% of total roadside refractory PM with number concentration modes at 20 and 80 nm was linked to *jeepneys*, the most popular and inexpensive mode of public transport in Metro Manila. They associated the larger mode with soot agglomerates, which is consistent with the smallest MOUDI size range examined here (0.056-0.1 µm; Figure 4b) exhibiting signs of agglomeration.

The total BC mass concentration integrated across all stages of MOUDI set MO13 (14.3 μg m⁻³) was remarkably high in contrast to BC levels measured via either filters, aethalometers, or single particle soot photometers in most other urban regions of the world (Metcalf et al., 2012 and references therein): Los Angeles Basin (airborne: 0.002–0.53 μg m⁻³), Atlanta, Georgia (ground: 0.5–3.0 μg m⁻³), Mexico City (airborne: 0.276–1.1 μg m⁻³), Sapporo, Japan (ground: 2.3–8.0 μg m⁻³), Beijing, China (ground: 6.3–11.1 μg m⁻³), Bangalor, India (ground: 0.4–10.2 μg m⁻³), Paris, France (ground: 7.9 μg m⁻³), Dushanbe, Russia (ground: 4–20 μg m⁻³), Po Valley, Italy (ground: 0.5–1.5 μg m⁻³), Thessaloniki, Greece (ground: 3.3–8.9 μg m⁻³). This is intriguing in light of extensive precipitation, and thus wet scavenging of PM, during the study period, which is offset by enormous anthropogenic emissions in the region such as by powered vehicles like the *jeepneys* that are notorious for BC exhaust (Kecorius et al., 2017).

A possible explanation for the large contribution of BC to PM, and the persistence of PM after rain events (Kim Oanh et al., 2006), is that the BC is not efficiently scavenged by precipitating rain drops. Small particles enter rain drops via diffusion whereas large particles enter via impaction. However, particles with a diameter in the range of $0.1-1~\mu m$ (known as the Greenfield gap) are too large to diffuse efficiently and too small to impact, and are therefore not efficiently scavenged (Seinfeld and Pandis, 2016). Absorption spectroscopy of set MO13 (Figure 2b) reveals that 95% of the BC mass is concentrated in the Greenfield gap, and thus the removal of BC due to precipitation is inefficient. The Greenfield gap contains $62 \pm 11\%$ of the total mass (calculated for MO3/MO13) and $65 \pm 10\%$ of the water-soluble mass (calculated for the other 12 MO sets). As noted earlier, BC observations discussed in this paper were based only on a single MOUDI set and the effect of inefficient scavenging in the Greenfield Gap could just be one of the many potential

processes affecting the BC mass size distribution. Subsequent work that will include BC measurements in the dry season will further investigate this hypothesis.

3.2.2 Water-Soluble Ions

There were two characteristic mass size distribution profiles for the water-soluble ions speciated by IC, depending on whether the species were secondarily produced via gas-to-particle conversion or associated with primarily emitted supermicrometer particles. The average IC species mass concentration profile across all MOUDI sets is shown in Figure 5. Secondarilyproduced species exhibited a mass concentration mode between $0.32-0.56~\mu m$, including common inorganic species (SO_4^{2-} , NH_4^+), MSA, amines (DMA, TMA+DEA), and a suite of organic acids, such as oxalate, phthalate, succinate, and adipate, produced via precursor volatile organic compounds (VOCs). Two organic acids with peaks in other size ranges included maleate ($0.56-1~\mu m$) and pyruvate ($0.1-0.18~\mu m$). Sources of the inorganics are well documented with SO_4^{2-} and NH_4^+ produced by precursor vapors SO_2 and NH_3 , respectively, with ocean-emitted dimethylsulfide (DMS) as an additional precursor to SO_4^{2-} and the primary precursor to MSA.

Precursors leading to secondarily-produced alkyl amines such as DMA, TMA, and DEA likely originated from a combination of industrial activity, marine emissions, biomass burning, vehicular activity, sewage treatment, waste incineration, and the food industry (e.g., Facchini et al., 2008; Sorooshian et al., 2009; Ge et al., 2011; VandenBoer et al., 2011); another key source of these species, animal husbandry (Mosier et al., 1973; Schade and Crutzen, 1995; Sorooshian et al., 2008), was ruled out owing to a scarcity of such activity in the study region. Secondarily-produced amine salts likely were formed with SO₄²⁻ as the chief anion owing to its much higher concentrations relative to NO₃⁻ or organic acids.

Dimethylamine was the most abundant amine similar to other marine (Muller et al., 2009) and urban regions (Youn et al., 2015); the average concentration of DMA integrated over all MOUDI stages for all sample sets was 62.2 ng m⁻³ in contrast to 29.8 ng m⁻³ for TMA+DEA. For reference, the other key cation (NH₄⁺) participating in salt formation with acids such as H₂SO₄ and HNO₃ was expectedly much more abundant (1.64 μ g m⁻³). With regard to the competitive uptake of DMA versus NH₃ in particles, the molar ratio of DMA:NH₄⁺ exhibited a unimodal profile between 0.1–1.8 μ m with a peak of 0.022 between 0.32–0.56 μ m and the lowest values at the tails (0.004 between 0.1–0.18 and 1–1.8 μ m); DMA was not above detection limits for either D_p < 0.1

 μm or $D_p > 1.8 \ \mu m$. The molar ratios observed were consistent with values measured in urban air of Tucson, Arizona and coastal air in Marina, California (0–0.04; Youn et al., 2015) and near the lower end of the range measured in rural and urban air masses sampled near Toronto (0.005–0.2: VandenBoer et al., 2011).

The most abundant organic acid was oxalate ($195 \pm 144 \, \text{ng m}^{-3}$), followed by succinate ($21 \pm 41 \, \text{ng m}^{-3}$), phthalate ($19 \pm 25 \, \text{ng m}^{-3}$), maleate ($17 \pm 15 \, \text{ng m}^{-3}$), and adipate ($5 \pm 8 \, \text{ng m}^{-3}$). The observation of mass concentrations increasing with decreasing carbon number for dicarboxylic acids (i.e., oxalate > succinate > adipate) is consistent with many past studies for other regions as larger chain acids undergo oxidative decay to eventually form oxalate (e.g., Kawamura and Ikushima, 1993; Kawamura and Sakaguchi, 1999; Sorooshian et al., 2007). Maleate is an unsaturated dicarboxylic acid emitted from gas and diesel engines (Rogge et al., 1993) and a product from the photo-oxidation of benzene (Kawamura and Ikushima, 1993). The aromatic dicarboxylic acid phthalate is a known photo-oxidation product of naphthalene and stems largely from plastic processing and fuel combustion (Fraser et al., 2003; Kautzman et al., 2010; Fu et al., 2012; Kleindienst et al., 2012). The oxidation product (MSA) of ocean-derived DMS exhibited an overall average concentration of $11 \pm 7 \, \text{ng m}^{-3}$, which is near the lower end of the range of levels reported in other coastal and marine environments (from undetected up to ~200 ng m $^{-3}$) (e.g., Saltzman et al., 1983, 1986; Berresheim 1987; Watts et al., 1987; Burgermeister and Georgii, 1991; Sorooshian et al., 2015; Xu and Gao, 2015).

Water-soluble species exhibiting a peak in the supermicrometer range, usually between $1.8\text{--}5.6~\mu\text{m}$, include those with known affiliations with sea salt (Na⁺, Cl⁻, K+, Mg²⁺) and crustal materials such as dust (Ca²⁺). Nitrate peaked between $1.8\text{--}3.2~\mu\text{m}$, and was best correlated with Na⁺ and Mg²⁺, suggestive of HNO₃ partitioning to sea salt as has been observed in other coastal regions (e.g., Prabhakar et al., 2014a). There was very little NO₃⁻ in the submicrometer range (0.05 \pm 0.04 μg m⁻³) in contrast to supermicrometer sizes (0.78 \pm 0.47 μg m⁻³). More submicrometer NO₃⁻ in the form of NH₄NO₃ would be expected if there was an excess of NH₃ after neutralizing SO₄²⁻. The mean ammonium-to-sulfate molar ratio for submicrometer sizes was 2.32 ± 0.52 (range: 1.11 - 2.78), with full neutralization of SO₄²⁻ in 10 of 12 MOUDI sets. Thus, there was a nonnegligible excess in NH₃ that presumably participated in salt formation with HNO₃ and organic species. The significant levels of NO₃⁻ in the same mode as Na⁺ and Cl⁻ contributed to the significant Cl⁻ depletion observed, as the mean Cl⁻:Na⁺ mass ratio between 1-10 μ m (i.e., range of

peak sea salt influence) was 0.81 ± 0.28 , which is much lower than the ratio for pure sea salt (1.81) (Martens et al., 1973). The subject of Cl⁻ depletion in this region will be investigated more thoroughly in subsequent work.

Figure 6 shows SEM images of representative single particles in each supermicrometer stage. As would be expected for sea salt and crustal material, most of the particles shown are not spherical. Interestingly, only the particle shown between 1–1.8 μ m was close to being spherical. Its composition based on EDX analysis was accounted for mostly by carbon (93.7%) with lower amounts of oxygen (5.8%) and Fe (0.5%). Sea salt particles were found in the next two stages owing to the highest combined weight percentages of Na⁺ and Cl⁻ based on EDX analysis: 1.8–3.2 μ m = 36.9%; 3.2–5.6 μ m = 46.9%. The salt particles are not necessarily cubical but more rounded with signs of agglomeration. These two particles were the only ones among the 11 MOUDI stages exhibiting an EDX signal for S, with contributions amounting to ~2% in each particle. This may be linked to natural SO₄²⁻ existing in sea salt particles. Also, the particle between 3.2–5.6 μ m contained a trace amount of Sc (1%). The largest three particles (\geq 5.6 μ m) were expectedly irregularly shaped with both sharp and rounded edges, comprised mostly of oxygen, Al, Fe, and Ca based on EDX analysis.

3.2.3 Water-Soluble Elements

Averaged data across all MOUDI sets reveal that ICP-QQQ elements exhibited a variety of mass concentration profiles ranging from a distinct mode in either the sub- or supermicrometer range to having multiple modes below and above 1 µm (averages across all MOUDI sets shown in Figure 7). There were several elements with only one distinct peak, being in one of the two stages between 0.18-1.0 µm, including As, Cd, Co, Cr, Cs, Cu, Hf, Mn, Mo, Ni, Rb, Se, Sn, Tl, V, Pb, and Zn. In contrast, the following elements exhibited only one distinct peak in the supermicrometer range: Al, Ba, P, Sr, Ti, Y, and Zr. The rest of the elements exhibited more complex behavior with two distinct peaks in the sub- and supermicrometer range (Ag, Fe, Nb). The following section discusses relationships between all of the ions and elements with a view towards identifying characteristic sources.

3.3 Characteristic Sources and Species Relationships

A combination of PMF and correlation analysis helped identify clusters of closely-related species stemming from distinct sources. The PMF solution with five factors (Figure 8) was chosen because it passed the criteria of physical meaningfulness and it had a calculated ratio of Q_{true}:Q_{expected} (1.2) that was very close to the theoretical value of 1.0. There was a high coefficient of determination between measured and predicted mass concentration when summing up all species for each MOUDI stage ($r^2 = 0.79$; sample size, n = 132), which added confidence in relying on the PMF model for source apportionment of PM. The five distinct clusters were named for their most plausible sources based on the species included in the groupings, with their overall contributions to the total mass based on PMF analysis shown in parenthesis (Table 3): Aged (48.0%), Sea Salt (22.5%), Combustion (18.7%), Vehicular/Resuspended Dust (5.6%), and Waste Processing (5.1%). For reference, a previous study near the northwestern edge of the Philippines identified six source factors for $PM_{2.5}$ that are fairly similar to those here (Bagtasa et al., 2018): sea salt, resuspended fine dust, local solid waste burning, and long range transport of (i) industrial emissions, (ii) solid waste burning, and (iii) secondary sulfate. Each of our five groupings will be discussed in detail below in decreasing order of contribution to total measured mass concentrations.

3.3.1 Aged Aerosol

Although not due to one individual source, there was a distinct PMF factor that included species commonly produced via gas-to-particle conversion processes (NH₄⁺, SO₄²⁻, MSA, oxalate). Correlation analysis (Table 4) also pointed to a large cluster of species significantly related to each other, including the aforementioned ions and a suite of other organic acids (phthalate, succinate, adipate), MSA, and DMA. The latter three inorganic and organic acid ions exhibited significant correlations with each other ($r \ge 0.68$), but also with several elements ($r \ge 0.36$: K, V, Rb, Cs, Sn), which were likely co-emitted with the precursor vapors of the secondarily produced ions. Although BC concentrations were quantified from set MO13 only, the results showed that BC was significantly correlated (r: 0.61-0.92) with 15 species, including those mentioned above (owing to co-emission) and also a few elements that were found via PMF to be stronger contributors to the Combustion source discussed in Section 3.3.3 (Ni, Cu, As, Se, Cd, Tl, Pb).

This PMF source factor is referred to as Aged Aerosol owing to its characteristic species being linked to secondary particle formation from emissions of local and regional sources. Examples include MSA and DMA being secondarily produced from ocean-derived gaseous emissions (e.g., Sorooshian et al., 2009), and K stemming from biomass burning emissions from upwind regions such as Sumatra and Borneo (Xian et al., 2013). Previous studies (Reid et al., 2012; Wang et al., 2013) have shown that phenomena such as SWM and El-Nino events not only influence biomass burning activities in the Malay Peninsula but also impact the transport and distribution of emissions in the study region. For instance, Reid et al. (2016b) showed that enhancement in monsoonal flow facilitates the advection of biomass burning and anthropogenic emissions to the Philippines from Sumatra and Borneo. Subsequent work will investigate more deeply the impact of biomass burning from those upwind regions on the sample site during the SWM.

While NH₄⁺ and SO₄²⁻ require time for production owing to being secondarily-produced from precursor vapors (i.e., SO₂, NH₃), oxalate is the smallest dicarboxylic acid and requires lengthier chemistry pathways for its production and thus is more likely produced in instances of aerosol transport and aging (e.g., Wonaschuetz et al., 2012; Ervens et al., 2018). The various elements associated with this cluster are co-emitted with the precursors to the aforementioned ions and are linked to a variety of sources: metallurgical processes (Anderson et al., 1988; Csavina et al., 2011; Youn et al., 2016), fuel combustion (Nriagu, 1989; Allen et al., 2001; Shafer et al., 2012; Rocha and Correa, 2018), residual oil combustion (Watson et al., 2004), biomass burning (Maudlin et al., 2015), marine and terrestrial biogenic emissions (Sorooshian et al., 2015), and plastics processing (Fraser et al., 2003). In addition, there is extensive ship traffic in the general study region, which is a major source of species in this cluster of species, particularly V and SO₄²⁻ (e.g., Murphy et al., 2009; Coggon et al., 2012).

PMF analysis suggested that the Aged Aerosol factor contributed 48.0% to the total water-soluble mass budget during the study period. Most of the contribution resided in the submicrometer range (68.9%) unlike the supermicrometer range (18.6%), which is consistent with the overall mass size distribution of total PM peaking in the submicrometer range (Figure 2). The reconstructed mass size distribution for this PMF source factor shows the dominance of the mass in the submicrometer range with a peak between $0.32-0.56\,\mu m$ (Figure 9). The correlation matrices for the sub- and supermicrometer size ranges also show that the correlations between the species

most prominent in the Aged Aerosol category are stronger for the former size range (Tables S2-S3). The contribution of this PMF factor to the supermicrometer range is likely associated with species secondarily produced on coarse aerosol such as dust and sea salt. This is evident in the individual species mass size distributions where there is a dominant submicrometer mode but also non-negligible mass above $1 \mu m$.

Even though the PM in a heavily populated urban region, such as Metro Manila, is typically thought to be dominated by local sources of aerosols, the current PMF results show that contribution from long range transport is still discernible. This finding is contrary to the expectation that the signal of transported aerosols would be lost in the noise of locally-produced aerosols.

3.3.2 Sea Salt

As the MO sampling site is approximately 13 km from the nearest shoreline (Figure 1a) and downwind of Manila Bay in the SWM season, there was a great potential for marine emissions to impact the samples. There were several species with similar mass size distributions (mode: 1.8–5.6 µm) and highly correlated total mass concentrations ($r \ge 0.51$) that are linked to sea salt: Cl^- , Na^+ , Ca^{2+} , Mg^{2+} , Ba, and Sr. The correlations between these species were stronger when examining just the supermicrometer range as compared to the submicrometer range (Tables S2-S3). The majority of these species was used in PMF analysis and formed a distinct cluster amounting to 22.0% of the total study period's mass budget. This source contributed only 0.6% to the submicrometer mass concentration but 53.5% for the supermicrometer size range. The reconstructed mass size distribution for this source factor is shifted farthest to the larger diameters as compared to the other four sources with a peak between 1.8-3.2 µm (Figure 9).

It is noteworthy that this factor has the highest share of NO_3^- among all identified sources. This result is consistent with mass size distributions shown in Figure 5 in which NO_3^- peaks in the supermicrometer range similar to sea salt constituents (e.g., Na^+ and Cl^-). Although sea salt particles naturally contain NO_3^- (Seinfeld and Pandis, 2016) (mass ratio of NO_3^- : $Na^+ = 9.8 \times 10^{-8} - 6.5 \times 10^{-5}$), the extremely high ratio of NO_3^- : Na^+ (mass ratio ~1.8) suggests that only a negligible portion of NO_3^- in this factor originated from primary sea salt particles. Thus, the majority of NO_3^- is most likely due to HNO_3 partitioning to existing sea salt particles (e.g., Fitzgerald, 1991; Allen et al., 1996; Dasgupta et al., 2007; Maudlin et al., 2015). In addition, the Cl^- : Na^+ mass ratio in this

profile (0.65) is smaller than that in sea salt particles (1.81), indicating high Cl⁻ depletion mainly due to reactions of HNO₃ with NaCl (Ro et al., 2001; Yao et al., 2003; Braun et al., 2017). Moreover, elevated loadings of trace elements (e.g., Ba, Cu, Zn, and Co) could be linked to mixing of marine emissions with urban sources (e.g., vehicle and industrial emissions) during their transport inland to the sampling site (Roth and Okada, 1998). This process of aging is consistent with the observed morphology of the sea salt particles in this study, revealing non-cubical shapes that are rounded owing to the likely addition of acidic species such as HNO₃ (Figure 6).

3.3.3 Combustion

There are numerous sources of combustion in the study region, including a variety of mobile sources (e.g., cars, utility vehicles, trucks, buses, motorcycles) and stationary sources (e.g., power stations, cement works, oil refineries, boiler stations, utility boilers). Consequently, the next highest contributor to total mass during the study period according to PMF (18.7%) was the cluster of species including Ni, As, Co, P, Mo, and Cr, which is defined as the Combustion factor. These species have been reported to be rich in particles emitted from combustion of fossil fuel and residual oil (Linak and Miller, 2000; Allen et al., 2001; Wasson et al., 2005; Mahowald et al., 2008; Mooibroek et al., 2011; Prabhakar et al., 2014b). Although not included in PMF analysis, other species significantly correlated with the previous ones include maleate and Ag, which also stem from fuel combustion (Kawamura and Kaplan, 1987; Lin et al., 2005; Sorooshian et al., 2007). Ag specifically is an element in waste incinerator fly ash (Buchholz and Landsberger, 1993; Tsakalou et al., 2018) and its strong correlation with Co (r = 0.85) and Mo (r = 0.64) provides support for this source factor being linked to combustion processes. Maleate is commonly found in engine exhaust (Kawamura and Kaplan, 1987), while Cr is a tracer for power plant emissions (Singh et al., 2002; Behera et al., 2015). Of all species examined in this study, BC was best correlated with As (r = 0.92), while its correlation with Ni (r = 0.85) was among the highest.

As the elements in this cluster peaked in concentration in the submicrometer mode, the weight percentage of this factor is more than double below 1 μ m (23.9%) as compared to above 1 μ m (11.3%). The reconstructed mass size distribution for this source factor peaks between 0.18–0.32 μ m, which is smaller than the modal diameter range for the Aged source factor (0.32–0.56 μ m) likely owing to closer sources and thus less time for growth to occur via condensation and coagulation.

3.3.4 Vehicular/Resuspended Dust

The next PMF source factor contains chemical signatures of dust because of high contributions to Al, Ti, Ca, and Fe. These crustal elements are strongly related to resuspension of dust by traffic and construction activities (Singh et al., 2002; Harrison et al., 2011). Other elements that were prominent in this factor included Zr, Y, Mn, Cr, and Ba, which are associated with tire and brake wear (Adachi and Tainosho, 2004; Gietl et al., 2010; Song and Gao, 2011; Harrison et al., 2012; Vossler et al., 2016), although some of them can be linked to exhaust as well (e.g., Lin et al., 2005; Song and Gao, 2011). This source is named Vehicular/Resuspended Dust and contributed 5.6% to the total study period's mass concentrations.

The weight percentage contribution of this factor was much higher for the supermicrometer range (11.3%) as compared to the submicrometer range (1.5%), which is consistent with the Sea Salt source factor owing to similar mass size distributions of the individual species associated with the two source categories (Figures 5 and 7). Additional species correlated significantly with the crustal species included Hf and Nb, which also exhibited mass peaks between $1.8-3.2~\mu m$. The reconstructed mass size distribution for this source factor is similar to that of Sea Salt in that there is a peak between $1.8-3.2~\mu m$, but there is less of a unimodal profile owing to what appears to be a secondary mode between $0.56-1.0~\mu m$ (Figure 9), which could be linked to some of the non-dust components of vehicular emissions.

3.3.5 Waste Processing

The final PMF source factor, contributing the least overall to total mass (5.1%), featured Zn, Cd, Pb, Mn, and Cu as its main components. These species are linked to waste processing, including especially electronic waste (e-waste) and battery burning and recycling (Gullett et al., 2007; Iijima et al., 2007), which was previously reported for Manila (Pabroa et al., 2011). The latter study reported that although there are a few licensed operations for battery recycling, there are numerous unregulated cottage melters across Manila that regularly melt metal from batteries and discard the waste freely. Fujimori et al. (2012) additionally showed that e-waste recycling led to emissions of the following elements (in agreement with this PMF cluster) around Metro Manila: Ni, Cu, Pb, Zn, Cd, Ag, in, As, Co, Fe, and Mn.

This was the only PMF factor exhibiting comparable weight percentages both below (5.1%) and above 1 μ m (5.3%). This is reflected in the mass size distributions of the species

included in this cluster being fairly uniformly distributed below and above 1 μ m. This is also demonstrated in the reconstructed mass size distribution of this source factor as it clearly exhibits a mode between the other four sources (0.56–1.0 μ m) and is the broadest mode (Figure 9). The explanation for this is likely rooted in the diversity of sources contained within this source profile that lead to different sizes of particles. Examples of such sources include processing of different types of waste at varying temperatures and through various processes (e.g., burning, melting, grinding) (Keshtkar and Ashbaugh, 2007),

4. Conclusions

This study used various analytical techniques (gravimetry, ion chromatography, triple quadrupole inductively coupled plasma mass spectrometry, black carbon spectroscopy, and microscopy), meteorological data, and a source apportionment model (Positive Matrix Factorization) to characterize the sources, chemical composition, and morphology of size-resolved ambient particulate matter (PM) collected using Micro-Orifice Uniform Deposit Impactors (MOUDIs) in Metro Manila, Philippines during the southwest monsoon (SWM) season of 2018. The main results of this study include the following:

- The total mass concentrations were measured on two occasions and were $18.6 \,\mu g \, m^{-3}$ and $53.0 \,\mu g \, m^{-3}$. Water-soluble mass concentrations were measured on 12 occasions and were on average $8.53 \pm 4.48 \,\mu g \, m^{-3}$ (range = $2.7 16.6 \,\mu g \, m^{-3}$). Simultaneous measurements of total, water-soluble, and black carbon (BC) mass revealed a composition of 26.9% BC, 31.3% water-soluble components, and 41.8% unaccounted mass.
- Size-resolved BC mass concentration was measured on one occasion, with the mass sum of all MOUDI stages reaching 14.3 μg m⁻³. Most of the BC mass (95%) was contained in the 0.1–1 μm range (i.e., the Greenfield gap) where wet scavenging by rain is relatively inefficient. The measured BC peaked in the size range of 0.18 0.32 μm and accounted for 51.8% of the measured PM for that stage. In the range of 0.10 0.18 μm, the mass percent contribution of BC to the measured PM was 78.1%.
- Most of the total mass resided in the submicrometer mode (0.32–0.56 µm); however, one MOUDI set revealed an additional supermicrometer mode (1.8–3.2 µm). Water-soluble species that peaked in the submicrometer mode were associated with secondarily produced

- species, including inorganic acids, amines, Methanesulfonate (MSA), and organic acids. Water-soluble species that peaked in the supermicrometer mode were associated with sea salt and crustal material. Most of the unaccounted mass was for $D_p > 0.32 \mu m$.
- The most abundant water-soluble species was SO₄²⁻ (44% ± 6%), followed by NH₄⁺ (18% ± 5%), NO₃⁻ (10 ± 3%), Na⁺ (8 ± 3%), and Cl⁻ (6% ± 3%). Correlation analysis revealed that total water-soluble mass was most correlated with temperature (r = 0.64) and rainfall accumulation (r = -0.49) among meteorological factors considered, although other factors were likely influential such as wind direction and speed.
- Regardless of particle size, the majority of single particles examined with energy dispersive X-ray spectroscopy (SEM-EDX) were non-spherical with evidence of agglomeration.

• PMF analysis suggested that there were five factors influencing the water-soluble fraction of PM collected at the sampling site. These factors, their contribution to total water-soluble mass, and the main species that permit them to be linked to a physical source are as follows: Aged Aerosol (48.0%; NH₄+, SO₄²⁻, MSA, oxalate), Sea Salt (22.5%; Cl⁻, NO₃-, Ca²⁺, Na⁺, Mg²⁺, Ba, Sr), Combustion (18.7%; Ni, As, Co, P, Mo, Cr), Vehicular/Resuspended Dust (5.6%; Al, Ti, Fe), and Waste Processing (5.1%; Zn, Cd, Pb, Mn, Cu). The dominant contribution of Aged aerosols to water-soluble mass contradicts two expectations: (i) locally-produced sources in polluted cities should drown out the signal of transported aerosols, and (ii) the signal of transported aerosols should be significantly reduced due to scavenging processes upwind of the measurement site.

Although the current study focuses exclusively on the SWM season in Metro Manila, results of this study are applicable to the study of aerosol impacts on Southeast Asia and other regions. First, the detection of Aged aerosols not only from local but also from regional sources confirms previous studies that PM in the region has the ability to travel long distances during the SWM season. Characterization of aerosols in Metro Manila is therefore important for better understanding the impacts that local emissions will have on locations downwind of Metro Manila, including other populated cities in Southeast and East Asia. Transport of pollution and decreased wet scavenging during the SWM season may become increasingly important as studies have shown a decrease in SWM rainfall and increase in the number of no-rain days during the SWM season in the western Philippines in recent decades (e.g., Cruz et al., 2013).

Second, Southeast Asia has been named "one of the most hostile environments on the planet for aerosol remote sensing" (Reid et al., 2013). Therefore, space-based remote sensing of aerosol characteristics, such as retrievals of aerosol optical depth (AOD), in this region are difficult. In situ measurements are critical for characterization of PM in this region, especially during seasons such as the SWM when clouds are especially prevalent and remote-sensing retrievals dependent on clear-sky conditions are lacking.

Third, this study provides a valuable dataset to compare to other regions impacted by monsoons where the impacts of enhanced moisture and rainfall on size-resolved composition are not well understood. As aqueous processing results in enhanced production of water-soluble species (e.g., sulfate, organic acids), it is noteworthy for this monsoonal region that the water-soluble fraction remains low relative to BC and other insoluble components. This has major implications for the hygroscopicity of the regional PM.

Finally, the results of this study will be used to inform future sampling campaigns in this region, including CAMP²Ex planned for the SWM season of 2019 based in the Philippines. As the current MOUDI sampling campaign at the Manila Observatory is expected to extend for a full year, future work will focus on changes in aerosol characteristics and sources on a seasonal basis.

Data availability: All data used in this work are available upon request.

Author Contribution: MTC, MOC, JBS, ABM, CS, and AS designed the experiments and all co-authors carried out some aspect of the data collection. MTC, RAB, CS, LM, HD, and AS conducted data analysis and interpretation. MTC and AS prepared the manuscript with contributions from all co-authors.

Competing interests: The authors declare that they have no conflict of interest.

Acknowledgements: This research was funded by NASA grant 80NSSC18K0148. M. T. Cruz acknowledges support from the Philippine Department of Science and Technology's ASTHRD Program. R. A. Braun acknowledges support from the ARCS Foundation. A. B. MacDonald acknowledges support from the Mexican National Council for Science and Technology

- 718 (CONACYT). We acknowledge Agilent Technologies for their support and Shane Snyder's
- 719 laboratories for ICP-QQQ data.

- 721 References
- Adachi, K., and Tainosho, Y.: Characterization of heavy metal particles embedded in tire dust,
- 723 Environ Int, 30, 1009-1017, 10.1016/j.envint.2004.04.004, 2004.

724

- Alas, H. D., Müller, T., Birmili, W., Kecorius, S., Cambaliza, M.O., Simpas, J. B., Cayetano, M.,
- Weinhold, K., Vallar, E., Galvez, M.C., and Wiedensohler, A.: Spatial Characterization of Black
- 727 Carbon Mass Concentration in the Atmosphere of a Southeast Asian Megacity: An Air Quality
- 728 Case Study for Metro Manila, Philippines. Aerosol Air Qual Res.
- 729 doi.org/10.4209/aaqr.2017.08.0281, 2017.

730

- Allen, H. C., Laux, J. M., Vogt, R., FinlaysonPitts, B. J., and Hemminger, J. C.: Water-induced
- reorganization of ultrathin nitrate films on NaCl: Implications for the tropospheric chemistry of
- 733 sea salt particles, J Phys Chem-Us, 100, 6371-6375, DOI 10.1021/jp953675a, 1996.

734

- Allen, A. G., Nemitz, E., Shi, J. P., Harrison, R. M., and Greenwood, J. C.: Size distributions of
- trace metals in atmospheric aerosols in the United Kingdom, Atmos Environ, 35, 4581-4591, Doi
- 737 10.1016/S1352-2310(01)00190-X, 2001.

738

- 739 Almeida, G. P., Bittencourt, A. T., Evangelista, M. S., Vieira-Filho, M. S., and Fornaro, A.:
- 740 Characterization of aerosol chemical composition from urban pollution in Brazil and its possible
- impacts on the aerosol hygroscopicity and size distribution, Atmos Environ, 202, 149-159,
- 742 10.1016/j.atmosenv.2019.01.024, 2019.

743

- Anderson, J. R., Aggett, F. J., Buseck, P. R., Germani, M. S., and Shattuck, T. W.: Chemistry of
- 745 Individual Aerosol-Particles from Chandler, Arizona, an Arid Urban-Environment, Environ Sci
- 746 Technol, 22, 811-818, DOI 10.1021/es00172a011, 1988.

747

- Aswini, A. R., Hegde, P., Nair, P. R., and Aryasree, S.: Seasonal changes in carbonaceous
- aerosols over a tropical coastal location in response to meteorological processes, Science of The
- 750 Total Environment, 656, 1261-1279, https://doi.org/10.1016/j.scitotenv.2018.11.366, 2019.

751

- 752 Bagtasa, G., Cayetano, M. G., and Yuan, C. S.: Seasonal variation and chemical characterization
- of PM_{2.5} in northwestern Philippines, Atmos Chem Phys, 18, 4965-4980, 10.5194/acp-18-4965-
- 754 2018, 2018.

755

- Bautista, A. T., Pabroa, P. C. B., Santos, F. L., Racho, J. M. D., and Quirit, L. L.: Carbonaceous
- 757 particulate matter characterization in an urban and a rural site in the Philippines, Atmos Pollut Res,
- 758 5, 245-252, 10.5094/Apr.2014.030, 2014.

- 760 Begam, G. R., Vachaspati, C. V., Ahammed, Y. N., Kumar, K. R., Reddy, R. R., Sharma, S. K.,
- Saxena, M., and Mandal, T. K.: Seasonal characteristics of water-soluble inorganic ions and
- carbonaceous aerosols in total suspended particulate matter at a rural semi-arid site, Kadapa

763 (India), Environmental Science and Pollution Research, 24, 1719-1734, 10.1007/s11356-016-764 7917-1, 2017.

765

Behera, S. N., Betha, R., Huang, X., and Balasubramanian, R.: Characterization and estimation of human airway deposition of size-resolved particulate-bound trace elements during a recent haze episode in Southeast Asia, Environ Sci Pollut R, 22, 4265-4280, 10.1007/s11356-014-3645-6, 2015.

770

Berresheim, H.: Biogenic Sulfur Emissions from the Sub-Antarctic and Antarctic Oceans, J Geophys Res-Atmos, 92, 13245-13262, 10.1029/JD092iD11p13245, 1987.

773

- Braun, R. A., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M., Maudlin, L. C., Crosbie, E., Aghdam, M. A., Mardi, A. H., and Sorooshian, A.: Impact of Wildfire Emissions on Chloride
- Aghdam, M. A., Mardi, A. H., and Sorooshian, A.: Impact of Wildfire Emissions on Chloride and Bromide Depletion in Marine Aerosol Particles, Environ Sci Technol, 51, 9013-9021,
- 777 10.1021/acs.est.7b02039, 2017.

778

Buchholz, B. A., and Landsberger, S.: Trace-Metal Analysis of Size-Fractioned Municipal Solid-Waste Incinerator Fly-Ash and Its Leachates, J Environ Sci Heal A, 28, 423-441, Doi 10.1080/10934529309375887, 1993.

782

Burgermeister, S., and Georgii, H. W.: Distribution of Methanesulfonate, Nss Sulfate and Dimethylsulfide over the Atlantic and the North-Sea, Atmos Environ a-Gen, 25, 587-595, Doi 10.1016/0960-1686(91)90056-D, 1991.

786

787 Chen, Y. Z., Shah, N., Huggins, F. E., and Huffman, G. P.: Microanalysis of ambient particles 788 from Lexington, KY, by electron microscopy, Atmos Environ, 40, 651-663, 789 10.1016/j.atmosenv.2005.09.036, 2006.

790

791 Chithra, V. S., and Nagendra, S. M. S.: Chemical and morphological characteristics of indoor and outdoor particulate matter in an urban environment, Atmos Environ, 77, 579-587, 10.1016/j.atmosenv.2013.05.044, 2013.

794

Chuang, M.-T., Chang, S.-C., Lin, N.-H., Wang, J.-L., Sheu, G.-R., Chang, Y.-J., and Lee, C.-T.: Aerosol chemical properties and related pollutants measured in Dongsha Island in the northern South China Sea during 7-SEAS/Dongsha Experiment, Atmospheric Environment, 78, 82-92, https://doi.org/10.1016/j.atmosenv.2012.05.014, 2013.

799

- 800 Clarke, A. D., Shinozuka, Y., Kapustin, V. N., Howell, S., Huebert, B., Doherty, S., Anderson,
- T., Covert, D., Anderson, J., Hua, X., Moore, K. G., McNaughton, C., Carmichael, G., and
- Weber, R.: Size distributions and mixtures of dust and black carbon aerosol in Asian outflow:
- Physiochemistry and optical properties, J Geophys Res-Atmos, 109, Artn D15s09 10.1029/2003jd004378, 2004.

- 806 Coggon, M. M., Sorooshian, A., Wang, Z., Metcalf, A. R., Frossard, A. A., Lin, J. J., Craven, J.
- S., Nenes, A., Jonsson, H. H., Russell, L. M., Flagan, R. C., and Seinfeld, J. H.: Ship impacts on the marine atmosphere: insights into the contribution of shipping emissions to the properties of
- 809 marine aerosol and clouds, Atmos Chem Phys, 12, 8439-8458, 10.5194/acp-12-8439-2012, 2012.

- 811 Cohen, D. D., Stelcer, E., Santos, F. L., Prior, M., Thompson, C., and Pabroa, P. C. B.:
- Fingerprinting and source apportionment of fine particle pollution in Manila by IBA and PMF
- 813 techniques: A 7-year study, X-Ray Spectrom, 38, 18-25, 10.1002/xrs.1112, 2009.

814

- 815 Crosbie, E., Sorooshian, A., Monfared, N. A., Shingler, T., and Esmaili, O.: A Multi-Year Aerosol
- 816 Characterization for the Greater Tehran Area Using Satellite, Surface, and Modeling Data,
- 817 Atmosphere-Basel, 5, 178-197, 10.3390/atmos5020178, 2014.

818

- 819 Crosbie, E., Youn, J. S., Balch, B., Wonaschutz, A., Shingler, T., Wang, Z., Conant, W. C.,
- 820 Betterton, E. A., and Sorooshian, A.: On the competition among aerosol number, size and
- 821 composition in predicting CCN variability: a multi-annual field study in an urbanized desert,
- 822 Atmos Chem Phys, 15, 6943-6958, 10.5194/acp-15-6943-2015, 2015.

823

- 824 Cruz, F. T., Narisma, G. T., Villafuerte, M. Q., Chua, K. U. C., and Olaguera, L. M.: A
- climatological analysis of the southwest monsoon rainfall in the Philippines, Atmos Res, 122, 609-
- 826 616, 10.1016/j.atmosres.2012.06.010, 2013.

827

- 828 Csavina, J., Landazuri, A., Wonaschutz, A., Rine, K., Rheinheimer, P., Barbaris, B., Conant, W.,
- 829 Saez, A. E., and Betterton, E. A.: Metal and Metalloid Contaminants in Atmospheric Aerosols
- from Mining Operations, Water Air Soil Poll, 221, 145-157, 10.1007/s11270-011-0777-x, 2011.

831

- Dasgupta, P. K., Campbell, S. W., Al-Horr, R. S., Ullah, S. M. R., Li, J. Z., Amalfitano, C., and
- Poor, N. D.: Conversion of sea salt aerosol to NaNO(3) and the production of HCl: Analysis of
- temporal behavior of aerosol chloride/nitrate and gaseous HCl/HNO(3) concentrations with
- 835 AIM, Atmos Environ, 41, 4242-4257, 10.1016/j.atmosenv.2006.09.054, 2007.

836

- Dumanoglu, Y., Kara, M., Altiok, H., Odabasi, M., Elbir, T., and Bayram, A.: Spatial and seasonal
- 838 variation and source apportionment of volatile organic compounds (VOCs) in a heavily
- 839 industrialized region, Atmos Environ, 98, 168-178, 10.1016/j.atmosenv.2014.08.048, 2014.

840

- 841 Ervens, B., Sorooshian, A., Aldhaif, A. M., Shingler, T., Crosbie, E., Ziemba, L., Campuzano-
- Jost, P., Jimenez, J. L., and Wisthaler, A.: Is there an aerosol signature of chemical cloud
- processing?, Atmos Chem Phys, 18, 16099-16119, 10.5194/acp-18-16099-2018, 2018.

844

- Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S.,
- Moretti, F., Tagliavini, E., Ceburnis, D., and O'Dowd, C. D.: Important Source of Marine
- 847 Secondary Organic Aerosol from Biogenic Amines, Environ Sci Technol, 42, 9116-9121,
- 848 10.1021/es8018385, 2008.

849

- Farren, N. J., Dunmore, R. E., Mead, M. I., Mohd Nadzir, M. S., Samah, A. A., Phang, S. M.,
- 851 Bandy, B. J., Sturges, W. T., and Hamilton, J. F.: Chemical characterisation of water-soluble
- ions in atmospheric particulate matter on the east coast of Peninsular Malaysia, Atmos. Chem.
- 853 Phys., 19, 1537-1553, 10.5194/acp-19-1537-2019, 2019.

- Fitzgerald, J. W.: Marine Aerosols a Review, Atmos Environ a-Gen, 25, 533-545, Doi
- 856 10.1016/0960-1686(91)90050-H, 1991.

- Fraser, M. P., Cass, G. R., and Simoneit, B. R. T.: Air quality model evaluation data for organics.
- 6. C-3-C-24 organic acids, Environ Sci Technol, 37, 446-453, 10.1021/es0209262, 2003.

860

- Fu, P. Q., Kawamura, K., Chen, J., Li, J., Sun, Y. L., Liu, Y., Tachibana, E., Aggarwal, S. G.,
- Okuzawa, K., Tanimoto, H., Kanaya, Y., and Wang, Z. F.: Diurnal variations of organic
- molecular tracers and stable carbon isotopic composition in atmospheric aerosols over Mt. Tai in
- the North China Plain: an influence of biomass burning, Atmos Chem Phys, 12, 8359-8375,
- 865 10.5194/acp-12-8359-2012, 2012.

866

- Fujimori, T., Takigami, H., Agusa, T., Eguchi, A., Bekki, K., Yoshida, A., Terazono, A., and
- Ballesteros, F. C.: Impact of metals in surface matrices from formal and informal electronic-
- waste recycling around Metro Manila, the Philippines, and intra-Asian comparison, J Hazard
- 870 Mater, 221-222, 139-146, https://doi.org/10.1016/j.jhazmat.2012.04.019, 2012.

871

- 672 Ge, X. L., Wexler, A. S., and Clegg, S. L.: Atmospheric amines Part I. A review, Atmos Environ,
- 873 45, 524-546, 10.1016/j.atmosenv.2010.10.012, 2011.

874

- 675 Gietl, J. K., Lawrence, R., Thorpe, A. J., and Harrison, R. M.: Identification of brake wear particles
- and derivation of a quantitative tracer for brake dust at a major road, Atmos Environ, 44, 141-146,
- 877 10.1016/j.atmosenv.2009.10.016, 2010.

878

- Gullett, B. K., Linak, W. P., Touati, A., Wasson, S. J., Gatica, S., and King, C. J.: Characterization
- 880 of air emissions and residual ash from open burning of electronic wastes during simulated
- rudimentary recycling operations, J Mater Cycles Waste, 9, 69-79, 10.1007/s10163-006-0161-x,
- 882 2007.

883

- Harrison, R. M., Beddows, D. C. S., and Dall'Osto, M.: PMF Analysis of Wide-Range Particle
- Size Spectra Collected on a Major Highway, Environ Sci Technol, 45, 5522-5528,
- 886 10.1021/es2006622, 2011.

887

- Harrison, R. M., Jones, A. M., Gietl, J., Yin, J. X., and Green, D. C.: Estimation of the
- 889 Contributions of Brake Dust, Tire Wear, and Resuspension to Nonexhaust Traffic Particles
- 890 Derived from Atmospheric Measurements, Environ Sci Technol, 46, 6523-6529,
- 891 10.1021/es300894r, 2012.

892

- Hitzenberger, R., and Tohno, S.: Comparison of black carbon (BC) aerosols in two urban areas –
- 894 concentrations and size distributions, Atmospheric Environment, 35, 2153-2167,
- 895 https://doi.org/10.1016/S1352-2310(00)00480-5, 2001.

896

- Hopke, P. K., Cohen, D. D., Begum, B. A., Biswas, S. K., Ni, B., Pandit, G. G., Santoso, M.,
- 898 Chung, Y. S., Davy, P., Markwitz, A., Waheed, S., Siddique, N., Santos, F. L., Pabroa, P. C. B.,
- 899 Seneviratne, M. C. S., Wimolwattanapun, W., Bunprapob, S., Vuong, T. B., Duy Hien, P. and
- 900 Markowicz, A.: Urban air quality in the Asian region, Sci. Total Environ., 404(1), 103–112,
- 901 doi:10.1016/j.scitotenv.2008.05.039, 2008.

Huang, S. L., Rahn, K. A., and Arimoto, R.: Testing and optimizing two factor-analysis techniques on aerosol at Narragansett, Rhode Island, Atmos Environ, 33, 2169-2185, Doi 10.1016/S1352-2310(98)00324-0, 1999.

906

920

927

936

- Iijima, A., Sato, K., Yano, K., Tago, H., Kato, M., Kimura, H., and Furuta, N.: Particle size and composition distribution analysis of automotive brake abrasion dusts for the evaluation of antimony sources of airborne particulate matter, Atmos Environ, 41, 4908-4919, 10.1016/j.atmosenv.2007.02.005, 2007.
- Kahnert, M., Nousiainen, T., and Veihelmann, B.: Spherical and spheroidal model particles as an error source in aerosol climate forcing and radiance computations: A case study for feldspar aerosols, J Geophys Res-Atmos, 110, Artn D18s13, 10.1029/2004jd005558, 2005.
- Kautzman, K. E., Surratt, J. D., Chan, M. N., Chan, A. W. H., Hersey, S. P., Chhabra, P. S., Dalleska, N. F., Wennberg, P. O., Flagan, R. C., and Seinfeld, J. H.: Chemical Composition of Gas- and Aerosol-Phase Products from the Photooxidation of Naphthalene, J Phys Chem A, 114, 919 913-934, 10.1021/jp908530s, 2010.
- Kawamura, K., and Ikushima, K.: Seasonal-Changes in the Distribution of Dicarboxylic-Acids in the Urban Atmosphere, Environ Sci Technol, 27, 2227-2235, DOI 10.1021/es00047a033, 1993.
- Kawamura, K., and Kaplan, I. R.: Motor Exhaust Emissions as a Primary Source for Dicarboxylic Acids in Los-Angeles Ambient Air, Environ Sci Technol, 21, 105-110, DOI
 10.1021/es00155a014, 1987.
- Kawamura, K., and Sakaguchi, F.: Molecular distributions of water soluble dicarboxylic acids in
 marine aerosols over the Pacific Ocean including tropics, J Geophys Res-Atmos, 104, 3501-3509,
 Doi 10.1029/1998jd100041, 1999.
- Kecorius, S., Madueño, L., Löndahl, J., Vallar, E., Galvez, M. C., Idolor, L. F., Gonzaga-Cayetano, M., Müller, T., Birmili, W., and Wiedensohler, A.: Respiratory tract deposition of inhaled roadside ultrafine refractory particles in a polluted megacity of South-East Asia, Science of The Total Environment, 663, 265-274, https://doi.org/10.1016/j.scitotenv.2019.01.338, 2019.
- Kecorius, S., Madueno, L., Vallar, E., Alas, H., Betito, G., Birmili, W., Cambaliza, M. O., Catipay,
 G., Gonzaga-Cayetano, M., Galvez, M. C., Lorenzo, G., Muller, T., Simpas, J. B., Tamayo, E. G.,
 and Wiedensohler, A.: Aerosol particle mixing state, refractory particle number size distributions
 and emission factors in a polluted urban environment: Case study of Metro Manila, Philippines,
 Atmos Environ, 170, 169-183, 10.1016/j.atmosenv.2017.09.037, 2017.
- 942
 943 Keshtkar, H., and Ashbaugh, L. L.: Size distribution of polycyclic aromatic hydrocarbon
 944 particulate emission factors from agricultural burning, Atmos Environ, 41, 2729-2739,
 945 10.1016/j.atmosenv.2006.11.043, 2007.
- Kim Oanh, N. T., Upadhyay, N., Zhuang, Y. H., Hao, Z. P., Murthy, D. V. S., Lestari, P., Villarin,
 J. T., Chengchua, K., Co, H. X., Dung, N. T. and Lindgren, E. S.: Particulate air pollution in six

- 949 Asian cities: Spatial and temporal distributions, and associated sources, Atmos. Environ., 40(18),
- 950 3367–3380, doi:10.1016/j.atmosenv.2006.01.050, 2006.
- 951
- Kim Oanh, N. T., Pongkiatkul, P., Cruz, M. T., Trung Dung, N., Phillip, L., Zhang, G., and Lestari,
- 953 P.: Monitoring and Source Apportionment for Particulate Matter Pollution in Six Asian Cities, in:
- 954 Integrated Air Quality Management: Asian Case Studies, Kim Oanh, N. T. (Ed.), CRC Press,
- 955 Taylor & Francis Group, USA, 97-124, 2013.
- 956 Kleindienst, T. E., Jaoui, M., Lewandowski, M., Offenberg, J. H., and Docherty, K. S.: The
- 957 formation of SOA and chemical tracer compounds from the photooxidation of naphthalene and its
- methyl analogs in the presence and absence of nitrogen oxides, Atmos Chem Phys, 12, 8711-8726,
- 959 10.5194/acp-12-8711-2012, 2012.

- Liao, H., Chen, W. T., and Seinfeld, J. H.: Role of climate change in global predictions of future
- tropospheric ozone and aerosols, J Geophys Res-Atmos, 111, Artn D12304,
- 963 10.1029/2005jd006852, 2006.

964

- 965 Lin, C. C., Chen, S. J., Huang, K. L., Hwang, W. I., Chang-Chien, G. P., and Lin, W. Y.:
- 966 Characteristics of metals in nano/ultrafine/fine/coarse particles collected beside a heavily
- 967 trafficked road, Environ Sci Technol, 39, 8113-8122, 10.1021/es048182a, 2005.

968

- 969 Linak, W. P., and Miller, C. A.: Comparison of particle size distributions and elemental
- partitioning from the combustion of pulverized coal and residual fuel oil, J Air Waste Manage, 50,
- 971 1532-1544, Doi 10.1080/10473289.2000.10464171, 2000.

972

- 973 Ma, Y., Li, S., Zheng, J., Khalizov, A., Wang, X., Wang, Z., and Zhou, Y.: Size-resolved
- 974 measurements of mixing state and cloud-nucleating ability of aerosols in Nanjing, China, Journal
- 975 of Geophysical Research: Atmospheres, 122, 9430-9450, 10.1002/2017jd026583, 2017.

976

- 977 Ma, L., Dadashazar, D., Braun, R. A., MacDonald, A. B., Aghdam, M. A., Maudlin, L. C., and
- 978 Sorooshian, A.: Size-resolved characteristics of water-soluble particulate elements in a coastal
- area: Source identification, influence of wildfires, and diurnal variability, Atmos. Environ., 206,
- 980 72-84, https://doi.org/10.1016/j.atmosenv.2019.02.045, 2019.

981

- Mahowald, N., Jickells, T. D., Baker, A. R., Artaxo, P., Benitez-Nelson, C. R., Bergametti, G.,
- Bond, T. C., Chen, Y., Cohen, D. D., Herut, B., Kubilay, N., Losno, R., Luo, C., Maenhaut, W.,
- 984 McGee, K. A., Okin, G. S., Siefert, R. L., and Tsukuda, S.: Global distribution of atmospheric
- 985 phosphorus sources, concentrations and deposition rates, and anthropogenic impacts, Global
- 986 Biogeochem Cy, 22, 10.1029/2008gb003240, 2008.

987

- 988 Marple, V., Olson, B., Romay, F., Hudak, G., Geerts, S. M. and Lundgren, D.: Second generation
- 989 micro-orifice uniform deposit impactor, 120 MOUDI-II: Design, Evaluation, and application to
- 990 long-term ambient sampling, Aerosol Sci. Technol., 48(4), 427–433,
- 991 doi:10.1080/02786826.2014.884274, 2014.

- 993 Martens, C. S., Wesolowski, J. J., Harriss, R. C., and Kaifer, R.: Chlorine Loss from Puerto-Rican
- 994 and San-Francisco-Bay Area Marine Aerosols, J Geophys Res, 78, 8778-8792, DOI
- 995 10.1029/JC078i036p08778, 1973.

- 997 Maudlin, L. C., Wang, Z., Jonsson, H. H., and Sorooshian, A.: Impact of wildfires on size-
- 998 resolved aerosol composition at a coastal California site, Atmos Environ, 119, 59-68,
- 999 10.1016/j.atmosenv.2015.08.039, 2015.

1000

- 1001 Metcalf, A. R., Craven, J. S., Ensberg, J. J., Brioude, J., Angevine, W., Sorooshian, A., Duong,
- H. T., Jonsson, H. H., Flagan, R. C., and Seinfeld, J. H.: Black carbon aerosol over the Los
- 1003 Angeles Basin during CalNex, J Geophys Res-Atmos, 117, 10.1029/2011jd017255, 2012.

1004

- 1005 Mielonen, T., Levy, R. C., Aaltonen, V., Komppula, M., de Leeuw, G., Huttunen, J., Lihavainen,
- 1006 H., Kolmonen, P., Lehtinen, K. E. J., and Arola, A.: Evaluating the assumptions of surface
- reflectance and aerosol type selection within the MODIS aerosol retrieval over land: the problem
- of dust type selection, Atmos Meas Tech, 4, 201-214, 10.5194/amt-4-201-2011, 2011.

1009

- 1010 Miller, J., and Miller, J.C.: Statistics and chemometrics for analytical chemistry. Pearson
- 1011 Education, 2018.

1012

- 1013 Mishra, S. K., Agnihotri, R., Yadav, P. K., Singh, S., Prasad, M. V. S. N., Praveen, P. S., Tawale,
- 1014 J. S., Rashmi, Mishra, N. D., Arya, B. C., and Sharma, C.: Morphology of Atmospheric Particles
- over Semi-Arid Region (Jaipur, Rajasthan) of India: Implications for Optical Properties, Aerosol
- 1016 Air Qual Res, 15, 974-+, 10.4209/aagr.2014.10.0244, 2015.

1017

- 1018 Mooibroek, D., Schaap, M., Weijers, E. P., and Hoogerbrugge, R.: Source apportionment and
- spatial variability of PM_{2.5} using measurements at five sites in the Netherlands, Atmospheric
- 1020 Environment, 45, 4180-4191, 10.1016/j.atmosenv.2011.05.017, 2011.

1021

- Mosier, A. R., Andre, C. E., and Viets, F. G.: Identification of Aliphatic-Amines Volatilized from
- 1023 Cattle Feedyard, Environ Sci Technol, 7, 642-644, DOI 10.1021/es60079a009, 1973.

1024

- Muller, C., Iinuma, Y., Karstensen, J., van Pinxteren, D., Lehmann, S., Gnauk, T., and Herrmann,
- 1026 H.: Seasonal variation of aliphatic amines in marine sub-micrometer particles at the Cape Verde
- islands, Atmos Chem Phys, 9, 9587-9597, 2009.

1028

- Murphy, S. M., Agrawal, H., Sorooshian, A., Padro, L. T., Gates, H., Hersey, S., Welch, W. A.,
- 1030 Jung, H., Miller, J. W., Cocker, D. R., Nenes, A., Jonsson, H. H., Flagan, R. C., and Seinfeld, J.
- 1031 H.: Comprehensive Simultaneous Shipboard and Airborne Characterization of Exhaust from a
- 1032 Modern Container Ship at Sea, Environ Sci Technol, 43, 4626-4640, 10.1021/es802413j, 2009.

1033

- Norris, G., Duvall, R., Brown, S., and Bai, S.: EPA Positive Matrix Factorization (PMF) 5.0
- 1035 fundamentals and User Guide Prepared for the US Environmental Protection Agency Office of
- Research and Development, Washington, DC. Inc., Petaluma, 2014.

- Nriagu, J. O.: A Global Assessment of Natural Sources of Atmospheric Trace-Metals, Nature, 338,
- 1039 47-49, DOI 10.1038/338047a0, 1989.

- Pabroa, P. C. B., Santos, F. L., Morco, R. P., Racho, J. M. D., Bautista, A. T., and Bucal, C. G. D.:
- Receptor modeling studies for the characterization of air particulate lead pollution sources in
- Valenzuela sampling site (Philippines), Atmos Pollut Res, 2, 213-218, 10.5094/Apr.2011.027,

1044 2011.

1045

1046 Philippine Statistics Authority: https://psa.gov.ph/, Accessed 28 August 2018.

1047

- Polissar, A., Hopke, P., Paatero, P., Malm, W., and Sisler, J.: Atmospheric aerosol over Alaska 2.
- Elemental composition and sources. Journal of Geophysical Research 103, 19045-19057, 1998.

1050

- Prabhakar, G., Ervens, B., Wang, Z., Maudlin, L. C., Coggon, M. M., Jonsson, H. H., Seinfeld, J.
- H., and Sorooshian, A.: Sources of nitrate in stratocumulus cloud water: Airborne measurements
- during the 2011 E-PEACE and 2013 NiCE studies, Atmos Environ, 97, 166-173,
- 1054 10.1016/j.atmosenv.2014.08.019, 2014a.

1055

- 1056 Prabhakar, G., Sorooshian, A., Toffol, E., Arellano, A. F., and Betterton, E. A.: Spatiotemporal
- distribution of airborne particulate metals and metalloids in a populated arid region, Atmos
- 1058 Environ, 92, 339-347, 10.1016/j.atmosenv.2014.04.044, 2014b.

1059

- Qu, W. J., Wang, J., Zhang, X. Y., Wang, D., and Sheng, L. F.: Influence of relative humidity on
- aerosol composition: Impacts on light extinction and visibility impairment at two sites in coastal
- area of China, Atmos Res, 153, 500-511, 10.1016/j.atmosres.2014.10.009, 2015.

1063

- Raatikainen, T., Brus, D., Hyvärinen, A. P., Svensson, J., Asmi, E., and Lihavainen, H.: Black
- carbon concentrations and mixing state in the Finnish Arctic, Atmos. Chem. Phys., 15, 10057-
- 1066 10070, 10.5194/acp-15-10057-2015, 2015.

1067

- Ramachandran, S., and Rajesh, T. A.: Black carbon aerosol mass concentrations over
- Ahmedabad, an urban location in western India: Comparison with urban sites in Asia, Europe,
- 1070 Canada, and the United States, J Geophys Res-Atmos, 112, 10.1029/2006jd007488, 2007.

1071

- Ran, L., Deng, Z. Z., Wang, P. C., and Xia, X. A.: Black carbon and wavelength-dependent aerosol
- absorption in the North China Plain based on two-year aethalometer measurements, Atmos
- 1074 Environ, 142, 132-144, 10.1016/j.atmosenv.2016.07.014, 2016.

1075

- Reddington, C. L., McMeeking, G., Mann, G. W., Coe, H., Frontoso, M. G., Liu, D., Flynn, M.,
- 1077 Spracklen, D. V., and Carslaw, K. S.: The mass and number size distributions of black carbon
- aerosol over Europe, Atmos. Chem. Phys., 13, 4917-4939, 10.5194/acp-13-4917-2013, 2013.

1079

- 1080 Reff, A., Eberly, S.I., and Bhave, P.V.: Receptor modeling of ambient particulate matter data using
- positive matrix factorization: Review of existing methods. J Air Waste Manage 57, 146-154, 2007.

- Reid, J. S., Xian, P., Hyer, E. J., Flatau, M. K., Ramirez, E. M., Turk, F. J., Sampson, C. R., Zhang,
- 1084 C., Fukada, E. M., and Maloney, E. D.: Multi-scale meteorological conceptual analysis of observed
- active fire hotspot activity and smoke optical depth in the Maritime Continent, Atmos Chem Phys,
- 1086 12, 2117-2147, 10.5194/acp-12-2117-2012, 2012.

- Reid, J. S., Hyer, E. J., Johnson, R. S., Holben, B. N., Yokelson, R. J., Zhang, J. L., Campbell, J.
- 1089 R., Christopher, S. A., Di Girolamo, L., Giglio, L., Holz, R. E., Kearney, C., Miettinen, J., Reid,
- 1090 E. A., Turk, F. J., Wang, J., Xian, P., Zhao, G. Y., Balasubramanian, R., Chew, B. N., Janjai, S.,
- Lagrosas, N., Lestari, P., Lin, N. H., Mahmud, M., Nguyen, A. X., Norris, B., Oanh, N. T. K., Oo,
- 1092 M., Salinas, S. V., Welton, E. J., and Liew, S. C.: Observing and understanding the Southeast
- Asian aerosol system by remote sensing: An initial review and analysis for the Seven Southeast
- 1094 Asian Studies (7SEAS) program, Atmos Res, 122, 403-468, 10.1016/j.atmosres.2012.06.005,
- 1095 2013.

1096

- Reid, J. S., Xian, P., Holben, B. N., Hyer, E. J., Reid, E. A., Salinas, S. V., Zhang, J. L., Campbell,
- J. R., Chew, B. N., Holz, R. E., Kuciauskas, A. P., Lagrosas, N., Posselt, D. J., Sampson, C. R.,
- Walker, A. L., Welton, E. J., and Zhang, C. D.: Aerosol meteorology of the Maritime Continent
- 1100 for the 2012 7SEAS southwest monsoon intensive study Part 1: regional-scale phenomena,
- 1101 Atmos Chem Phys, 16, 14041-14056, 10.5194/acp-16-14041-2016, 2016a.

1102

- Reid, J. S., Lagrosas, N. D., Jonsson, H. H., Reid, E. A., Atwood, S. A., Boyd, T. J., Ghate, V. P.,
- 1104 Xian, P., Posselt, D. J., Simpas, J. B., Uy, S. N., Zaiger, K., Blake, D. R., Bucholtz, A., Campbell,
- J. R., Chew, B. N., Cliff, S. S., Holben, B. N., Holz, R. E., Hyer, E. J., Kreidenweis, S. M.,
- Kuciauskas, A. P., Lolli, S., Oo, M., Perry, K. D., Salinas, S. V., Sessions, W. R., Smirnov, A.,
- Walker, A. L., Wang, Q., Yu, L. Y., Zhang, J. L., and Zhao, Y. J.: Aerosol meteorology of
- 1108 Maritime Continent for the 2012 7SEAS southwest monsoon intensive study Part 2: Philippine
- 1109 receptor observations of fine-scale aerosol behavior, Atmos Chem Phys, 16, 14057-14078,
- 1110 10.5194/acp-16-14057-2016, 2016b.

1111

- Rocha, L. D. S., and Correa, S. M.: Determination of size-segregated elements in diesel-biodiesel
- blend exhaust emissions, Environ Sci Pollut R, 25, 18121-18129, 10.1007/s11356-018-1980-8,
- 1114 2018.

1115

- 1116 Rogge, W. F., Mazurek, M. A., Hildemann, L. M., Cass, G. R., and Simoneit, B. R. T.:
- 1117 Quantification of Urban Organic Aerosols at a Molecular-Level Identification, Abundance and
- 1118 Seasonal-Variation, Atmos Environ a-Gen, 27, 1309-1330, Doi 10.1016/0960-1686(93)90257-Y,
- 1119 1993.

1120

- 1121 Ro, C. U., Oh, K. Y., Kim, H., Kim, Y. P., Lee, C. B., Kim, K. H., Kang, C. H., Osan, J., De
- Hoog, J., Worobiec, A., and Van Grieken, R.: Single-particle analysis of aerosols at Cheju
- 1123 Island, Korea, using low-Z electron probe X-ray microanalysis: A direct proof of nitrate
- formation from sea salts, Environ Sci Technol, 35, 4487-4494, 10.1021/es0155231, 2001.

1125

- Rolph, G.D.: Real-time Environmental Applications and Display sYstem (READY) website
- 1127 (http://ready.Arl.NOAA.Gov), NOAA Air Resour. Lab., Silver Spring, Md., 2016.

1128

- Roth, B., and Okada, K.: On the modification of sea-salt particles in the coastal atmosphere,
- 1130 Atmos Environ, 32, 1555-1569, Doi 10.1016/S1352-2310(97)00378-6, 1998.

- Saltzman, E. S., Savoie, D. L., Zika, R. G., and Prospero, J. M.: Methane Sulfonic-Acid in the
- 1133 Marine Atmosphere, J Geophys Res-Oceans, 88, 897-902, DOI 10.1029/JC088iC15p10897, 1983.
- 1134
- Saltzman, E. S., Savoie, D. L., Prospero, J. M., and Zika, R. G.: Methanesulfonic-Acid and Non-
- 1136 Sea-Salt Sulfate in Pacific Air Regional and Seasonal-Variations, J Atmos Chem, 4, 227-240,
- 1137 Doi 10.1007/Bf00052002, 1986.
- 1138
- 1139 Schade, G. W., and Crutzen, P. J.: Emission of Aliphatic-Amines from Animal Husbandry and
- 1140 Their Reactions Potential Source of N₂O and HCN, J Atmos Chem, 22, 319-346, Doi
- 1141 10.1007/Bf00696641, 1995.
- 1142
- 1143 Schlosser, J. S., Braun, R. A., Bradley, T., Dadashazar, H., MacDonald, A. B., Aldhaif, A. M.,
- 1144 Aghdam, M. A., Mardi, A. H., Xian, P., and Sorooshian, A.: Analysis of Aerosol Composition
- Data for Western United States Wildfires Between 2005-2015: Dust Emissions, Chloride
- 1146 Depletion, and Most Enhanced Aerosol Constituents, J. Geophys. Res., 122,
- 1147 doi:10.1002/2017JD026547, 2017.
- 1148
- Schwarz, J. P., Gao, R. S., Spackman, J. R., Watts, L. A., Thomson, D. S., Fahey, D. W., Ryerson,
- T. B., Peischl, J., Holloway, J. S., Trainer, M., Frost, G. J., Baynard, T., Lack, D. A., de Gouw, J.
- 1151 A., Warneke, C., and Del Negro, L. A.: Measurement of the mixing state, mass, and optical size
- of individual black carbon particles in urban and biomass burning emissions, Geophysical
- 1153 Research Letters, 35, 10.1029/2008gl033968, 2008.
- Seinfeld, J. H., and Pandis, S. N.: Atmospheric chemistry and physics (3rd ed.). New York: Wiley-
- 1155 Interscience, 2016.
- 1156
- Shafer, M. M., Toner, B. M., Oyerdier, J. T., Schauer, J. J., Fakra, S. C., Hu, S. H., Herner, J. D.,
- and Ayala, A.: Chemical Speciation of Vanadium in Particulate Matter Emitted from Diesel
- 1159 Vehicles and Urban Atmospheric Aerosols, Environ Sci Technol, 46, 189-195
- 1160 10.1021/es200463c, 2012.
- 1161
- Shingler, T., Sorooshian, A., Ortega, A., Crosbie, E., Wonaschutz, A., Perring, A. E.,
- Beyersdorf, A., Ziemba, L., Jimenez, J. L., Campuzano-Jost, P., Mikoviny, T., Wisthaler, A., and
- Russell, L. M.: Ambient observations of hygroscopic growth factor and f(RH) below 1: Case
- studies from surface and airborne measurements, J Geophys Res-Atmos, 121, 13661-13677,
- 1166 10.1002/2016jd025471, 2016.
- 1167
- Shiraiwa, M., Kondo, Y., Moteki, N., Takegawa, N., Sahu, L. K., Takami, A., Hatakeyama, S.,
- Yonemura, S., and Blake, D. R.: Radiative impact of mixing state of black carbon aerosol in Asian
- outflow, Journal of Geophysical Research: Atmospheres, 113, 10.1029/2008jd010546, 2008.
- 1171
- Simpas, J., Lorenzo, G., and Cruz, M. T.: Monitoring Particulate Matter Levels and Composition
- for Source Apportionment Study in Metro Manila, Philippines, in: Improving Air Quality in Asian
- Developing Countries: Compilation of Research Findings, Kim Oanh, N. T. (Ed.), NARENCA,
- 1175 Vietnam Publishing House of Natural Resources, Environment and Cartography, Vietnam, 239-
- 1176 261, 2014.
- 1177

- Singh, M., Jaques, P. A., and Sioutas, C.: Size distribution and diurnal characteristics of particle-
- bound metals in source and receptor sites of the Los Angeles Basin, Atmos Environ, 36, 1675-
- 1180 1689, Pii \$1352-2310(02)00166-8, Doi 10.1016/\$1352-2310(02)00166-8, 2002.
- 1181
- Song, F., and Gao, Y.: Size distributions of trace elements associated with ambient particular
- matter in the affinity of a major highway in the New Jersey-New York metropolitan area, Atmos
- 1184 Environ, 45, 6714-6723, 10.1016/j.atmosenv.2011.08.031, 2011.
- 1185
- Sorooshian, A., Ng, N. L., Chan, A. W. H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.:
- Particulate organic acids and overall water-soluble aerosol composition measurements from the
- 1188 2006 Gulf of Mexico Atmospheric Composition and Climate Study (GoMACCS), J Geophys
- 1189 Res-Atmos, 112, 10.1029/2007jd008537, 2007.
- 1190
- Sorooshian, A., Murphy, S. N., Hersey, S., Gates, H., Padro, L. T., Nenes, A., Brechtel, F. J.,
- Jonsson, H., Flagan, R. C., and Seinfeld, J. H.: Comprehensive airborne characterization of aerosol
- from a major bovine source, Atmos Chem Phys, 8, 5489-5520, DOI 10.5194/acp-8-5489-2008,
- 1194 2008.
- 1195
- Sorooshian, A., Padro, L. T., Nenes, A., Feingold, G., McComiskey, A., Hersey, S. P., Gates, H.,
- Jonsson, H. H., Miller, S. D., Stephens, G. L., Flagan, R. C., and Seinfeld, J. H.: On the link
- between ocean biota emissions, aerosol, and maritime clouds: Airborne, ground, and satellite
- measurements off the coast of California, Global Biogeochem Cy, 23,
- 1200 10.1029/2009gb003464, 2009.
- 1201
- Sorooshian, A., Crosbie, E., Maudlin, L. C., Youn, J. S., Wang, Z., Shingler, T., Ortega, A. M.,
- 1203 Hersey, S., and Woods, R. K.: Surface and airborne measurements of organosulfur and
- methanesulfonate over the western United States and coastal areas, J Geophys Res-Atmos, 120,
- 1205 8535-8548, 10.1002/2015jd023822, 2015.
- 1206
- Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.: NOAA's
- Hysplit Atmospheric Transport and Dispersion Modeling System, B Am Meteorol Soc, 96, 2059-
- 1209 2077, 10.1175/Bams-D-14-00110.1, 2015.
- 1210
- Tai, A. P. K., Mickley, L. J., and Jacob, D. J.: Correlations between fine particulate matter (PM2.5)
- and meteorological variables in the United States: Implications for the sensitivity of PM2.5 to
- 1213 climate change, Atmos Environ, 44, 3976-3984, 10.1016/j.atmosenv.2010.06.060, 2010.
- 1214
- Tsakalou, C., Papamarkou, S., Tsakiridis, P. E., Bartzas, G., and Tsakalakis, K.: Characterization
- and leachability evaluation of medical wastes incineration fly and bottom ashes and their
- 1217 vitrification outgrowths, J Environ Chem Eng, 6, 367-376, 10.1016/j.jece.2017.12.012, 2018.
- 1218
- 1219 U.S. Environmental Protection Agency: Monitoring PM2.5 in Ambient Air
- Using Designated Reference or Class I Equivalent Methods. Report No. EPA-454/B-16-001. US
- Environmental Protection Agency, Research Triangle Park, NC., 2016.
- 1222

- 1223 VandenBoer, T. C., Petroff, A., Markovic, M. Z., and Murphy, J. G.: Size distribution of alkyl
- amines in continental particulate matter and their online detection in the gas and particle phase,
- 1225 Atmos Chem Phys, 11, 4319-4332, 10.5194/acp-11-4319-2011, 2011.

- 1227 Villafuerte, M. Q., Matsumoto, J., Akasaka, I., Takahashi, H. G., Kubota, H., and Cinco, T. A.:
- Long-term trends and variability of rainfall extremes in the Philippines, Atmos Res, 137, 1-13,
- 1229 10.1016/j.atmosres.2013.09.021, 2014.

1230

- Vossler, T., Cernikovsky, L., Novak, J., and Williams, R.: Source apportionment with
- uncertainty estimates of fine particulate matter in Ostrava, Czech Republic using Positive Matrix
- 1233 Factorization, Atmos Pollut Res, 7, 503-512, 10.1016/j.apr.2015.12.004, 2016.

1234

- 1235 Wang, J., Ge, C., Yang, Z. F., Hyer, E. J., Reid, J. S., Chew, B. N., Mahmud, M., Zhang, Y. X.,
- and Zhang, M. G.: Mesoscale modeling of smoke transport over the Southeast Asian Maritime
- 1237 Continent: Interplay of sea breeze, trade wind, typhoon, and topography, Atmos Res, 122, 486-
- 1238 503, 10.1016/j.atmosres.2012.05.009, 2013.

1239

- Wang, Y. Q., Zhang, X. Y., and Draxler, R. R.: TrajStat: GIS-based software that uses various
- trajectory statistical analysis methods to identify potential sources from long-term air pollution
- measurement data, Environ Modell Softw, 24, 938-939, 10.1016/j.envsoft.2009.01.004, 2009.

1243

- Wasson, S. J., Linak, W. P., Gullett, B. K., King, C. J., Touati, A., Huggins, F. E., Chen, Y. Z.,
- Shah, N., and Huffman, G. P.: Emissions of chromium, copper, arsenic, and PCDDs/Fs from open
- burning of CCA-treated wood, Environ Sci Technol, 39, 8865-8876, 10.1021/es050891g, 2005.

1247

- Watson, J. G.: Protocol for Applying and Validating the CMB Model for PM_{2.5} and VOC. Report
- No. EPA-451/R-04-001. US Environmental Protection Agency, Research Triangle Park, NC.,
- 1250 2004.

1251

- Watts, S. F., Watson, A., and Brimblecombe, P.: Measurements of the Aerosol Concentrations of
- Methanesulfonic Acid, Dimethyl-Sulfoxide and Dimethyl Sulfone in the Marine Atmosphere of
- the British-Isles, Atmos Environ, 21, 2667-2672, Doi 10.1016/0004-6981(87)90198-3, 1987.

1255

- Wonaschuetz, A., Sorooshian, A., Ervens, B., Chuang, P. Y., Feingold, G., Murphy, S. M., de
- Gouw, J., Warneke, C., and Jonsson, H. H.: Aerosol and gas re-distribution by shallow cumulus
- 1258 clouds: An investigation using airborne measurements, J Geophys Res-Atmos, 117,
- 1259 10.1029/2012jd018089, 2012.

1260

- Wu, D., Zhang, F., Lou, W. H., Li, D., and Chen, J. M.: Chemical characterization and toxicity
- assessment of fine particulate matters emitted from the combustion of petrol and diesel fuels, Sci
- Total Environ, 605, 172-179, 10.1016/j.scitotenv.2017.06.058, 2017.

1264

- 1265 Xian, P., Reid, J. S., Atwood, S. A., Johnson, R. S., Hyer, E. J., Westphal, D. L., and Sessions, W.:
- Smoke aerosol transport patterns over the Maritime Continent, Atmos Res, 122, 469-485,
- 1267 10.1016/j.atmosres.2012.05.006, 2013.

- 1269 Xu, G. J., and Gao, Y.: Characterization of marine aerosols and precipitation through shipboard
- observations on the transect between 31 degrees N-32 degrees S in the West Pacific, Atmos Pollut
- 1271 Res, 6, 154-161, 10.5094/Apr.2015.018, 2015.
- 1272
- 1273 Yao, X. H., Fang, M., and Chan, C. K.: The size dependence of chloride depletion in fine and
- 1274 coarse sea-salt particles, Atmos Environ, 37, 743-751, 10.1016/S1352-2310(02)00955-X, 2003.
- 1275
- 1276 Youn, J. S., Wang, Z., Wonaschutz, A., Arellano, A., Betterton, E. A., and Sorooshian, A.:
- 1277 Evidence of aqueous secondary organic aerosol formation from biogenic emissions in the North
- 1278 American Sonoran Desert, Geophys Res Lett, 40, 3468-3472, 10.1002/grl.50644, 2013.
- 1279
- Youn, J. S., Crosbie, E., Maudlin, L. C., Wang, Z., and Sorooshian, A.: Dimethylamine as a major
- alkyl amine species in particles and cloud water: Observations in semi-arid and coastal regions,
- 1282 Atmos Environ, 122, 250-258, 10.1016/j.atmosenv.2015.09.061, 2015.
- 1283

- Youn, J. S., Csavina, J., Rine, K. P., Shingler, T., Taylor, M. P., Saez, A. E., Betterton, E. A., and
- 1285 Sorooshian, A.: Hygroscopic Properties and Respiratory System Deposition Behavior of
- Particulate Matter Emitted By Mining and Smelting Operations, Environ Sci Technol, 50, 11706-
- 1287 11713, 10.1021/acs.est.6b03621, 2016.

Table 1. Summary of average operating parameters, meteorological conditions, and total resolved water-soluble mass concentration for each MOUDI sample set collected at Manila Observatory (MO) during the 2018 Southwest Monsoon period. On two occasions, simultaneous MOUDI sets were collected for one set to undergo gravimetric analysis (MO3 and MO13) to compare with mass resolved from chemical speciation of the water-soluble fraction (MO4 and MO14). One additional MOUDI set devoted to microscopy analysis was collected using aluminum substrates for one hour on August 1 at 30 LPM.

Sample	Dates	Durat	Flow	Wind	Wind	T	Rain	Water-
set name		ion	rate	speed	direction	(°C)	(mm)	soluble mass
		(hrs)	(LPM)	(m/s)	(°)			$(\mu g m^{-3})$
MO1	Jul 19-20	24	30	3.3	90.1	24.9	47	4.6
MO2	Jul 23-25	54	30	1.3	95.8	26.7	7.8	6.5
MO3/4	Jul 25-30	119	28/30	1.2	111.8	26.7	49.6	5.2
MO5	Jul 30-Aug 1	42	29	2.6	98.1	27.5	52.8	9.2
MO6	Aug 6-8	48	27	0.9	127.5	26.1	30.4	5.1
MO7	Aug 14-16	48	28	3.0	107.8	27.8	2.8	13.7
MO8	Aug 22-24	48	29	3.5	108.7	28.1	1	12.8
MO9	Sep 1-3	48	27	0.7	98.6	26.6	51.6	6.2
MO10	Sep 10–12	48	29	1.0	94.7	26.2	78.4	6.4
MO11	Sep 18–20	48	27	0.5	290.2	27.8	0	2.7
MO12	Sep 26-28	48	27	1.2	96.3	27.8	6.8	13.5
MO13/14	Oct 6-8	48	28/26	0.6	108.2	27.8	0.8	16.6

Table 2. Charge balance slopes (cations on y-axis; anions on x-axis) for the MOUDI sets shown including the averages of all sets (All) for three size ranges: submicrometer stages spanning 0.056 $-1.0~\mu m$; supermicrometer stages (> 1.0 μm); and all stages (> 0.056 μm). The species used in the charge balance analysis include those speciated with the IC (listed in Section 2.3) plus K from ICP-QQQ analysis.

Sample set	0.056 – 1.0 μm	> 1 µm	> 0.056 µm
MO1	0.87	1.37	0.89
MO2	1.46	1.26	1.41
MO4	1.25	1.17	1.21
MO5	1.35	1.43	1.41
MO6	1.29	1.45	1.31
MO7	1.40	1.23	1.36
MO8	1.35	1.33	1.36
MO9	1.28	1.55	1.26
MO10	1.37	1.36	1.35
MO11	0.97	1.60	1.27
MO12	1.37	1.19	1.33
MO14	1.31	1.28	1.29
All	1.35	1.24	1.33

Table 3. Contributions (in weight percentage) of each PMF source factor to the total mass in different diameter ranges.

Diameter Range (µm)	Aged/ Transported	Sea Salt	Combustion	Vehicular/ Resuspended	Waste Processing
	11 4 115p 01 00 0			Dust	
> 0.056	48.0%	22.5%	18.7%	5.6%	5.1%
0.056 - 1.0	68.9%	0.6%	23.9%	1.5%	5.1%
> 1.0	18.6%	53.5%	11.3%	11.3%	5.3%

Table 4. Correlation matrix (r values) between water-soluble species based on total MOUDI-integrated mass concentrations (> $0.056~\mu m$). Blank cells represent statistically insignificant values. Results for the sub- and supermicrometer ranges are in Tables S2-S3. Panels A-E represent important species from each of the source profiles identified in Section 3.3: A = Aged, B = Sea Salt, C = Combustion, D = Vehicular/Resuspended Dust, E = Waste Processing. DMA – Dimethylamine, MSA – Methanesulfonate, PH – Phthalate, OX – Oxalate, MA – Maleate, SU – Succinate, AD – Adipate.

A)															
OX	1.00														
SO_4	0.74	1.00		_											
NH ₄	0.68	0.99	1.00												
Sn	0.71	0.87	0.85	1.00											
Rb	0.73	0.74	0.73	0.69	1.00										
K	0.76	0.71	0.69	0.69	0.97	1.00									
Cs	0.72	0.82	0.81	0.74	0.96	0.91	1.00		_						
\mathbf{V}	0.36	0.64	0.63	0.48	0.53	0.51	0.57	1.00		-,					
DMA		0.35		0.38	0.45	0.37	0.45		1.00						
MSA	0.71	0.89	0.89	0.79	0.90	0.85	0.92	0.51	0.47	1.00		_			
PH	0.68	0.67	0.68	0.73	0.82	0.76	0.80		0.38	0.88	1.00		_		
SU	0.63	0.56	0.59	0.44	0.87	0.81	0.82		0.68	0.78	0.84	1.00			
AD	0.40	0.66	0.70	0.62	0.70	0.70	0.77		0.84	0.74	0.75	0.90	1.00		
Se	0.75	0.75	0.73	0.66	0.80	0.78	0.79	0.32	0.34	0.78	0.80	0.88	0.88	1.00	
Tl	0.75	0.87	0.86	0.80	0.89	0.85	0.94	0.74	0.65	0.80	0.52	0.70		0.43	1.00
	OX	SO_4	NH_4	Sn	Rb	K	Cs	V	DMA	MSA	PH	SU	AD	Se	Tl

	Cl	NO ₃	Ba	Sr	Ca	Na	Mg	Hf
Hf			·		0.57			1.00
Mg	0.91	0.87	0.77	0.87	0.66	0.99	1.00	
Na	0.93	0.87	0.75	0.85	0.63	1.00		
Ca	0.58	0.79	0.75	0.78	1.00			
Sr	0.78	0.87	0.91	1.00		_		
Ba	0.66	0.80	1.00					
NO ₃	0.76	1.00						
Cl	1.00							
B)		_						

	As	Ni	Co	P	Mo	Cr	Mal	Ag
Ag			0.85		0.64			1.00
MA			0.67		-0.42		1.00	
Cr	0.62	0.49		0.20		1.00		
Mo					1.00			
P		0.33	0.34	1.00				
Co			1.00					
Ni	0.58	1.00		_				
As	1.00		_					
C)								

	Zr	Y	Al	Fe	Ti	Nb
Nb	0.70	0.50	0.59	0.59	0.70	1.00
Ti	0.84	0.66	0.82	0.41	1.00	
Fe	0.33	0.61	0.25	1.00		
Al	0.88	0.76	1.00			
Y	0.75	1.00				
Zr	1.00					
D)						

E)					
Cd	1.00		_		
Zn	0.60	1.00			
Cu	0.21	0.27	1.00		
Mn	0.28	0.61	0.22	1.00	
Pb	0.78	0.58	0.38	0.27	1.00
	Cd	Zn	Cu	Mn	Pb

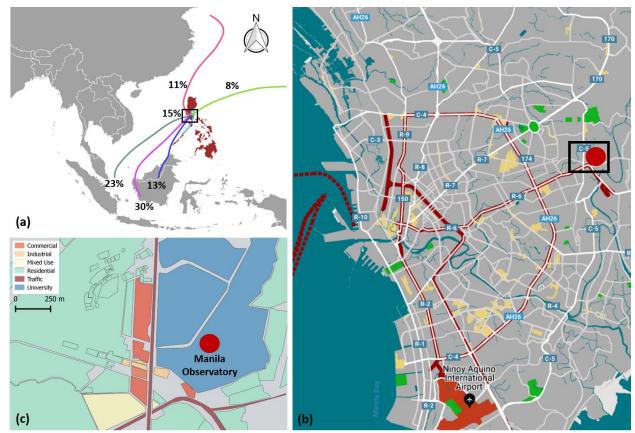


Figure 1. (a) Location of Metro Manila, Philippines relative to Southeast Asia. Also shown are 5-day backward trajectory frequencies during the sampling duration based on HYSPLIT cluster analysis; note that 15% correspond to trajectories within the black square. (b) Close-up view of Metro Manila showing the location of the Manila Observatory sampling site with a black rectangle. The base map shows roads, commercial centers, and major transit lines in the city. (c) Land use classification in the vicinity of the sampling site. (Sources: GADM, Snazzy Maps, OpenStreetMap, NOAA HYSPLIT, & TrajSat)

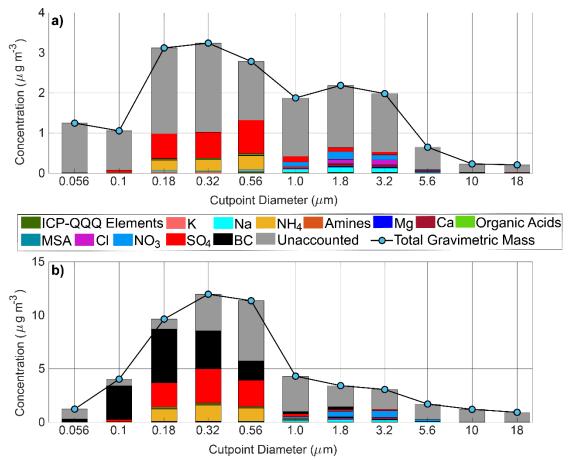


Figure 2. Mass size distributions of total PM (blue markers) and resolved chemical species (colored bars) for MOUDI sets (a) MO3/4 and (b) MO13/14. Note that set MO13 was the single MOUDI set where BC was quantified. ICP-QQQ = sum of water-soluble elements except K; amines = sum of DMA, TMA, DEA; organic acids = sum of oxalate, succinate, adipate, pyruvate, phthalate, maleate.



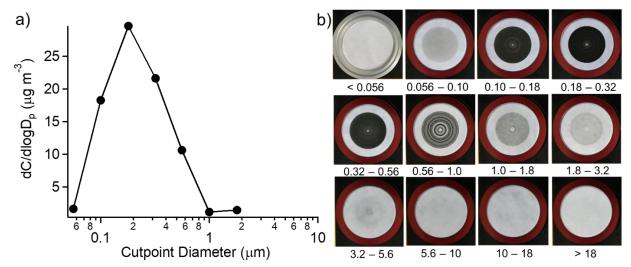


Figure 3. (a) Mass size distribution of BC retrieved from the MABI optical measurement at 870 nm for set MO13. Missing values were below detection limits. (b) Photographs of each stage of set MO13 with numbers below each image representing the aerodynamic diameter ranges in units of μm .

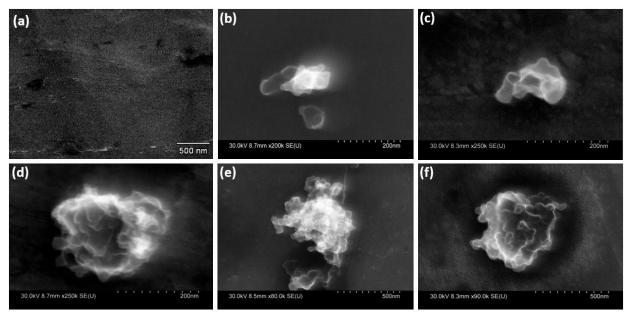


Figure 4. SEM image of a (a) blank filter and (b-f) individual particles in different submicrometer aerodynamic diameter ranges sampled by the MOUDI: (b) 0.056–0.1 μm , (c) 0.1–0.18 μm , (d) 0.18–0.32 μm , (e) 0.32–0.56 μm , (f) 0.56–1.0 μm .

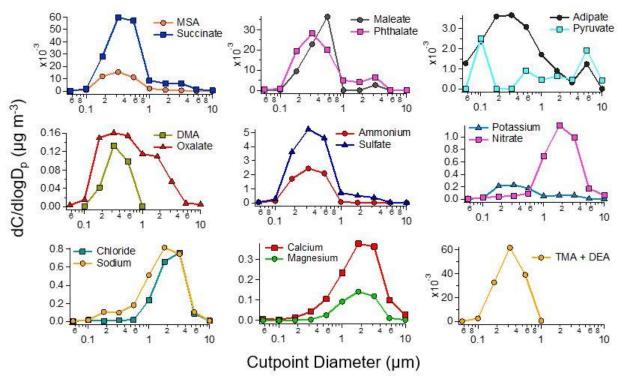


Figure 5. Average mass size distribution of water-soluble ions speciated via IC in addition to potassium from ICP-QQQ analysis.

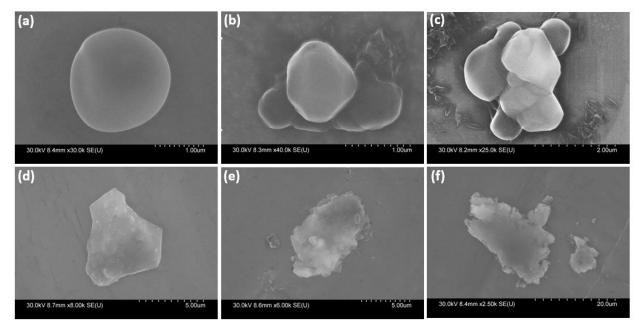


Figure 6. Same as Figure 4, but for different supermicrometer aerodynamic diameter ranges sampled by the MOUDI: (a) $1.0-1.8 \mu m$, (b) $1.8-3.2 \mu m$; (c) $3.2-5.6 \mu m$, (d) $5.6-10 \mu m$, (e) $10-18 \mu m$, (f) $> 18 \mu m$.

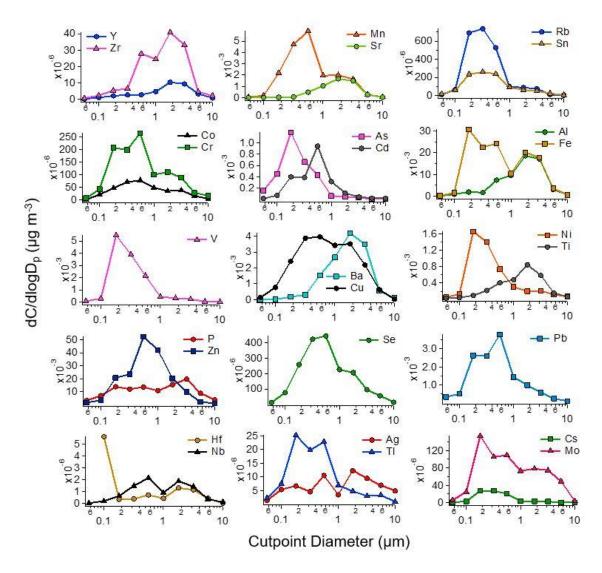


Figure 7. Average mass size distribution of water-soluble elements speciated via ICP-QQQ.

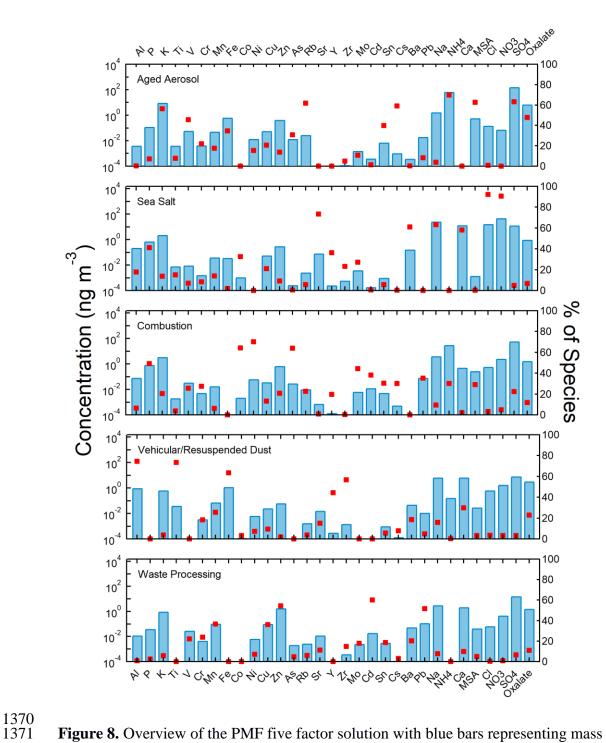


Figure 8. Overview of the PMF five factor solution with blue bars representing mass concentrations and red squares signifying the percentage of mass concentration contributed to constituents by each source factor.

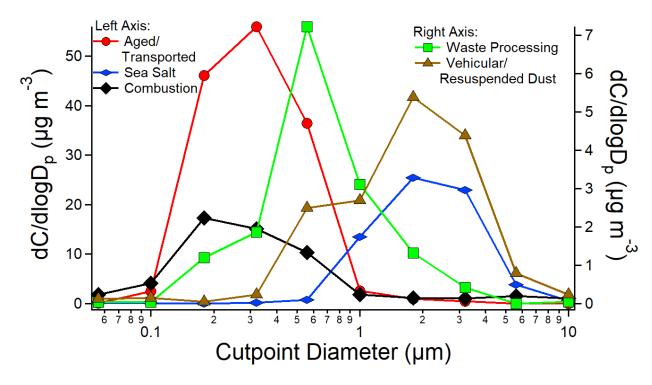


Figure 9. Reconstructed mass size distributions using PMF for the five major source profiles.